



# Structure elucidation and occurrence of Tc(IV) pyrogallol complexes

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- Fission product of uranium-235
- $\beta$  - emitter
- **Long half-life:  $2.13 \times 10^5$  y**
- High yield in radio-active waste (6% of all fission products !)
- **Potential migration upon storage**
- **Potential association with HS**

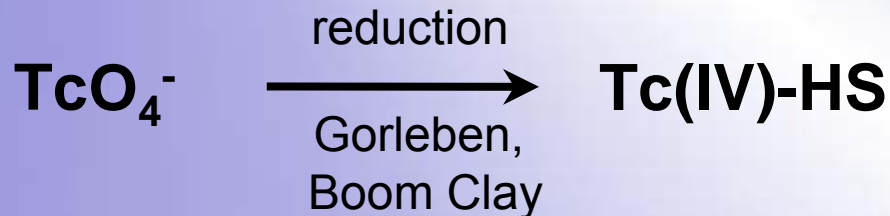


# $^{99}\text{Tc}$ behaviour in natural systems

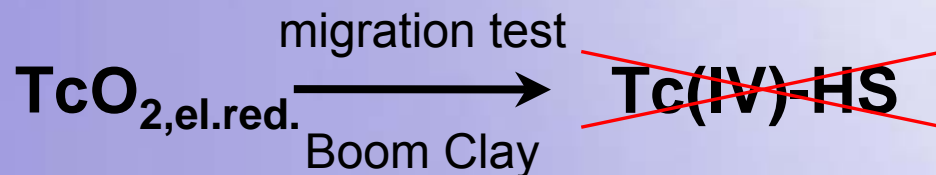


Organic matter

(Maes *et al.*, 2003;2004)

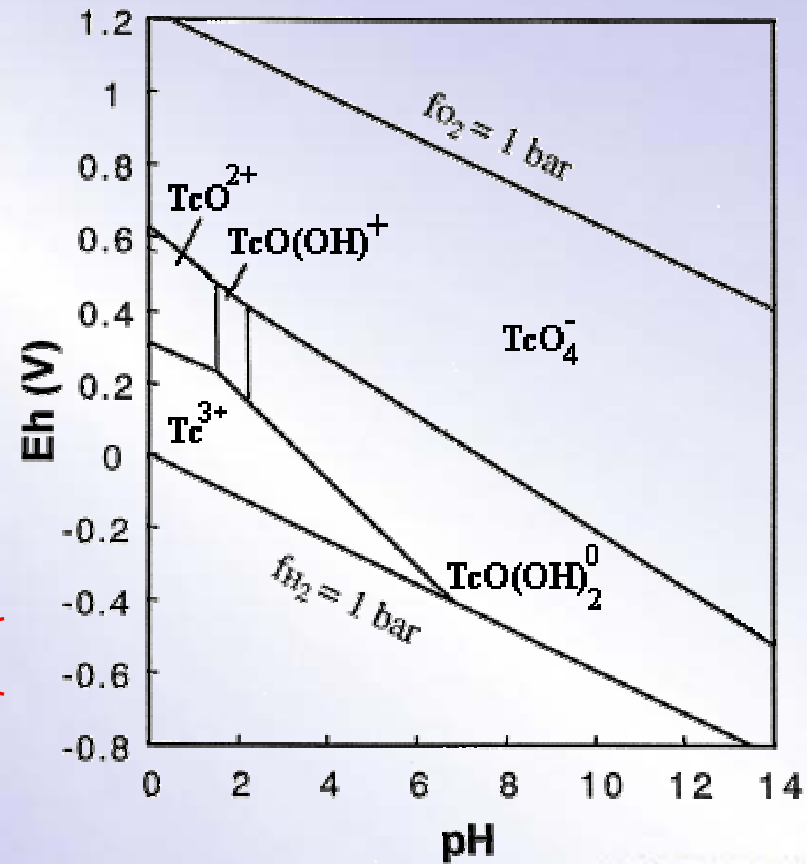


First sight contradiction



(Baston *et al.*, 2002)

~~Organic matter~~



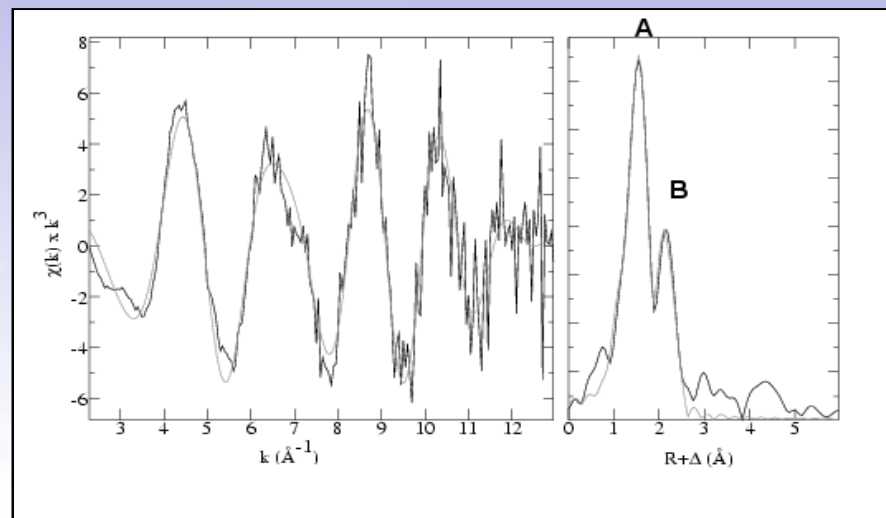
# Tc(IV) - HS

## Identity ?

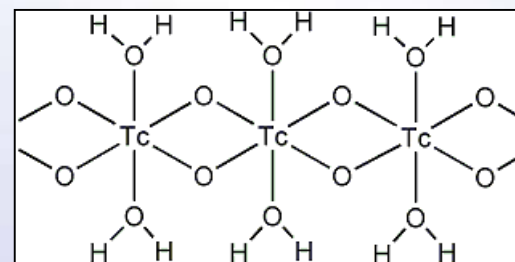
- Colloid association Tc(IV) eigencolloids and HS
- Hydrophobic interaction

## Proof ?

- XANES, EXAFS
- SEC
- Modelling BC batch SHUBERT approach



Tc-O			Tc-Tc		
C.N.	R	$\sigma^2$	C.N.	R	$\sigma^2$
6	2.03	0.006	1.5	2.56	0.005

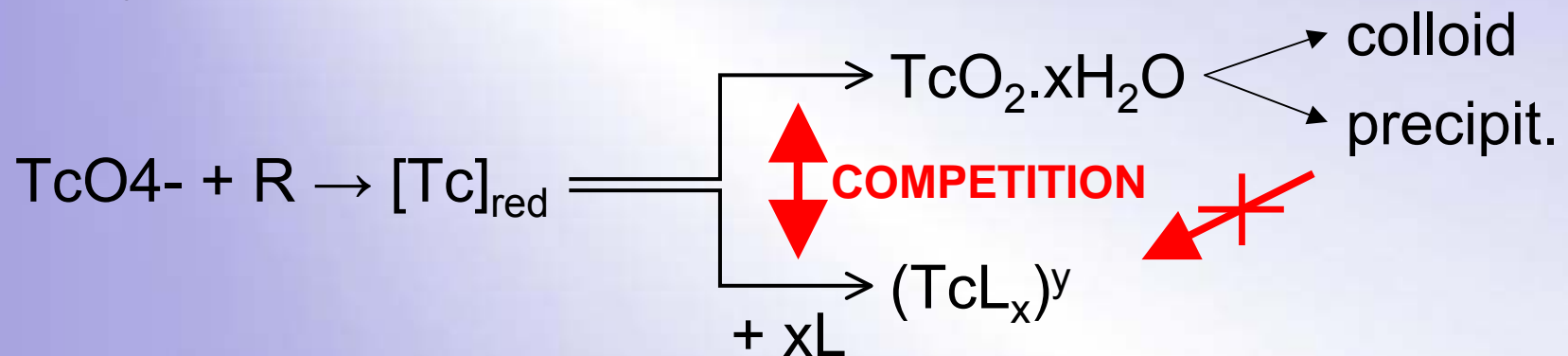




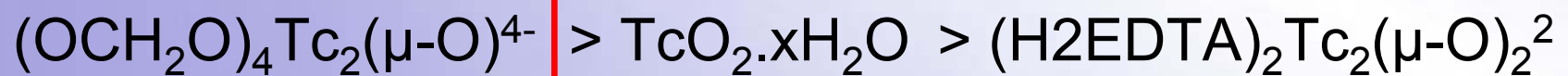
# Tc(IV)-HS Complexation ?

- Possible ?
- Nuclear medicine

range of synthetic Tc(IV) complexes e.g. EDTA, cysteine, ...



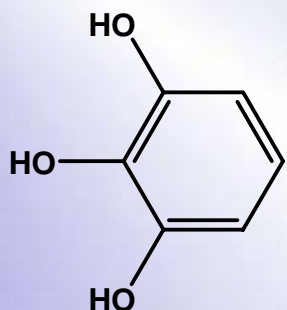
- High pH (2M NaOH) Tc(IV) complex stability series (Lukens *et al.*, 2002)



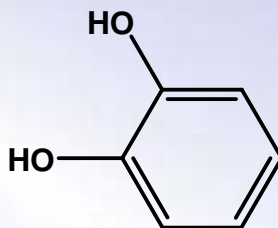
**DIOL**

# Simple HS-like ligands

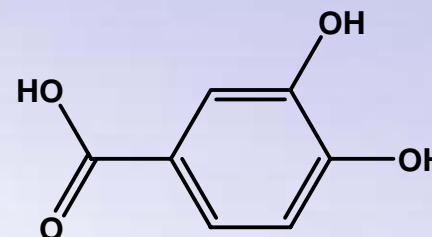
- Intermediates of wood degradation process



PYROGALLOL



CATECHOL



PROTOCATECHUIC ACID

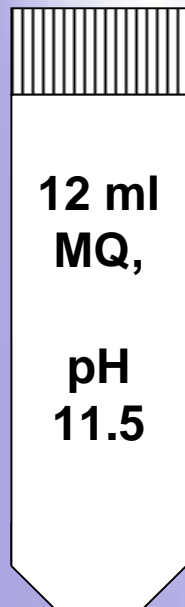
- High pH Tc(IV) species stability

diol > oxide > aminopolycarboxylic

- Choice : pyrogallol
- Phenolic, model building block of HS
- Probably highest stability versus oxide



# Tc(IV)-pyrogallol solutions



Daily, 12d

+

- 1ml pyrogallol  $1.25\text{E-}2\text{M}$
- 1ml  $\text{N}_2\text{H}_6\text{SO}_4$   $1.5\text{E-}3\text{M}$  or dithionite  $6.2\text{E-}3\text{M}$
- $50\mu\text{l}$   $6.2\text{E-}2\text{M}$   $\text{NH}_4\text{TcO}_4$

- sample preparation in glovebox ( $\text{N}_2/\text{H}_2$  95/5) ( $< 1$  ppm  $\text{O}_2$ )
- 2 weeks equilibrium
- 8 ml samples, titration to pH 11, 8, 5, 2
- another 2 weeks equilibrium



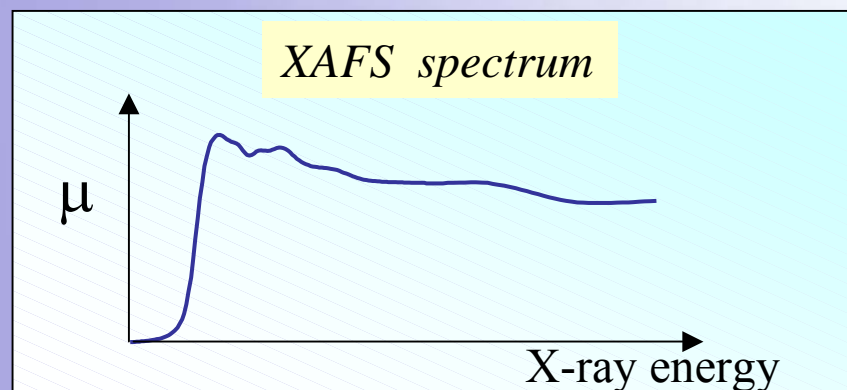
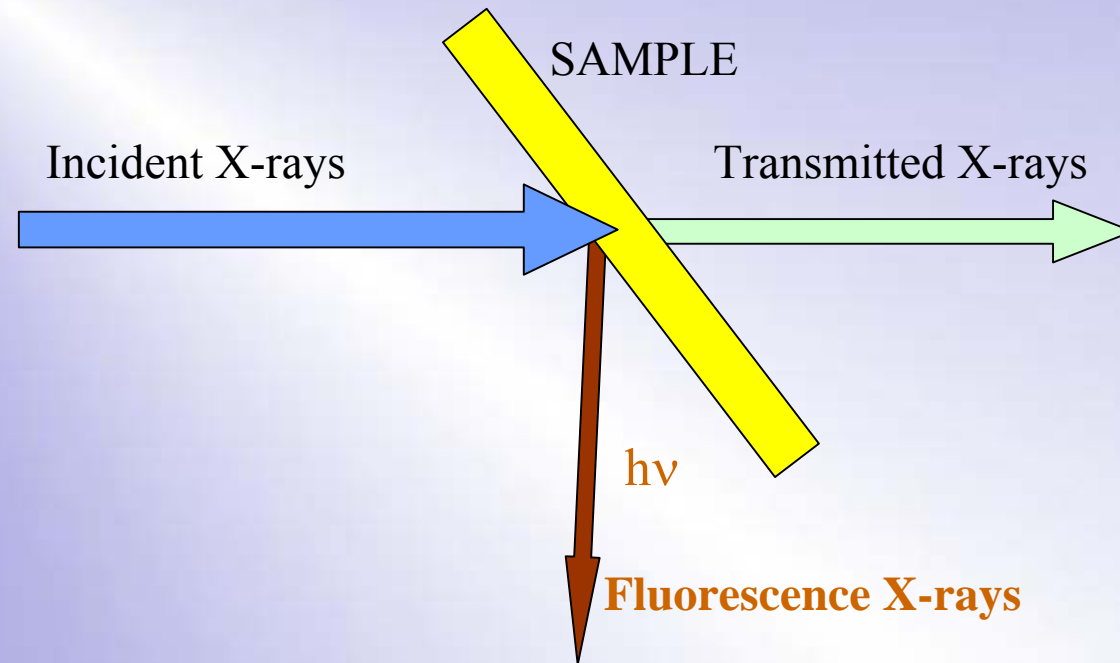
# XAS Sample Preparation & Measurement

- 8 ml sample, concentrated to 2 ml
- vac. distillation, 25°C, liquid N<sub>2</sub> trap
- sample container sealed in 2 PE bags
- transport to ESRF in steel container under N<sub>2</sub> atmosphere
- Measurement:
  - ROBL beamline (BM20)
  - Transmission mode, 21 → 22.5 keV
  - Mo foil standard, energy recalibration



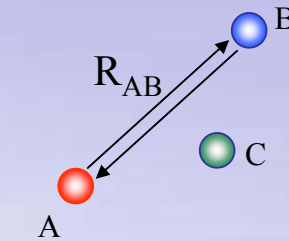
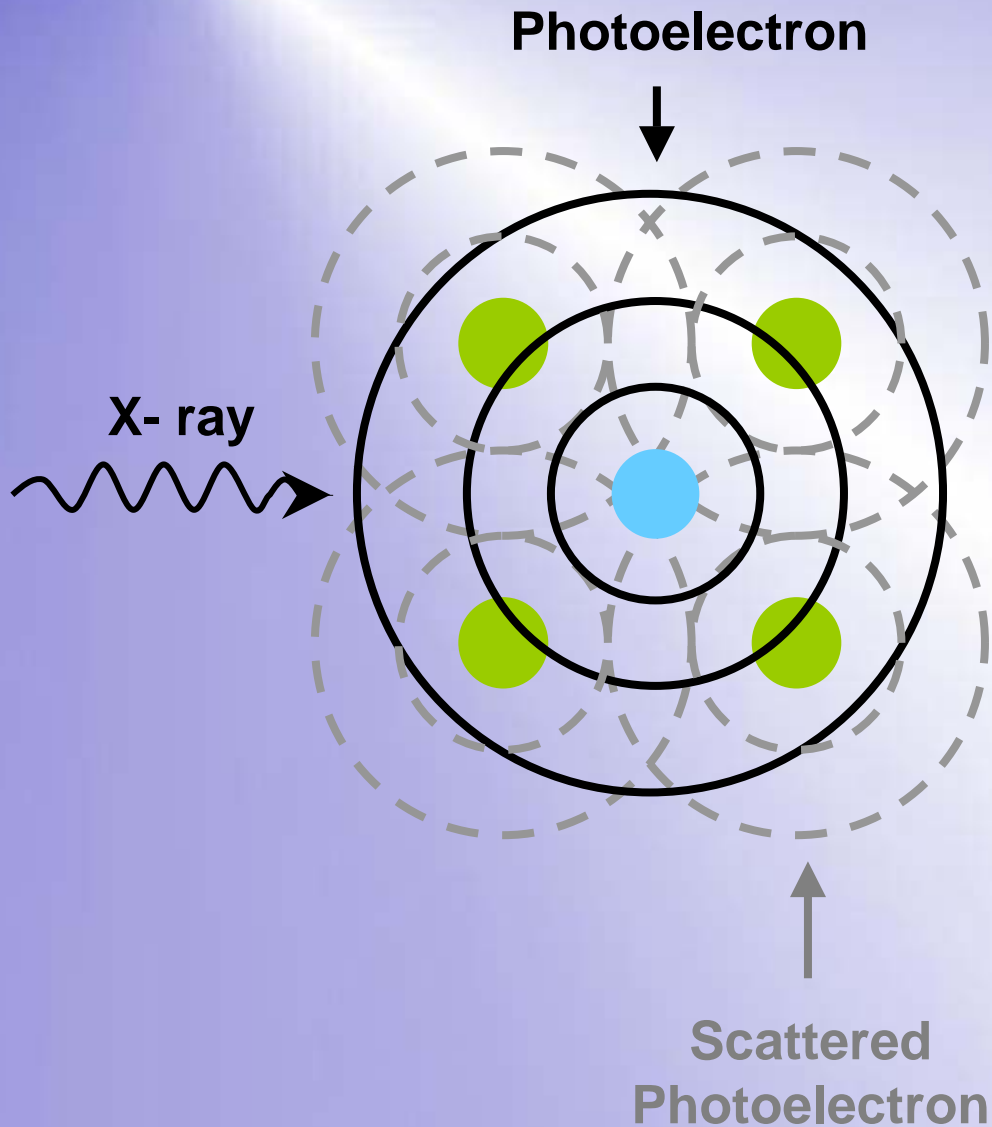


# XAFS measurements

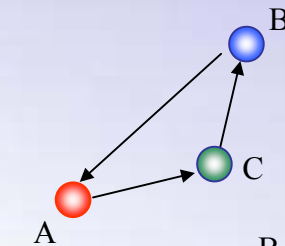




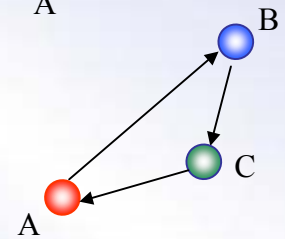
# EXAFS phenomenon



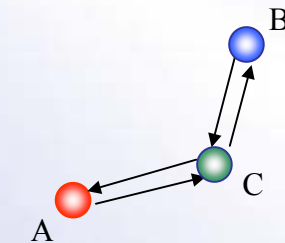
**Single scattering**  
' 2 legs '



**Double scattering**  
' 3 legs '



**Double scattering**  
' 3 legs '



**Triple scattering**  
' 4 legs '

Angular info



# XAS data analysis

**EXAFSPAK**

- Data recalibration
- Averaging raw data

**ATHENA**

- Data reduction

**EXAFSPAK**

- Fitting
  - 1) SS
  - 2) MS

**DFT**

- g03 rev. D.02
  - UB3LYP
  - LANL2DZ : Tc
  - 6-31G(d) : C, H, O, S

**FEFF**

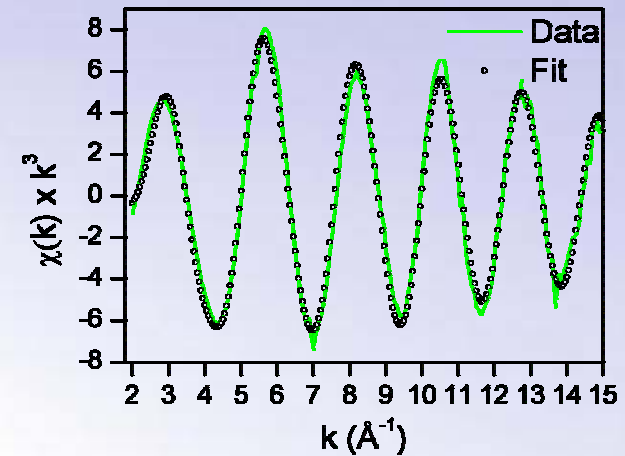
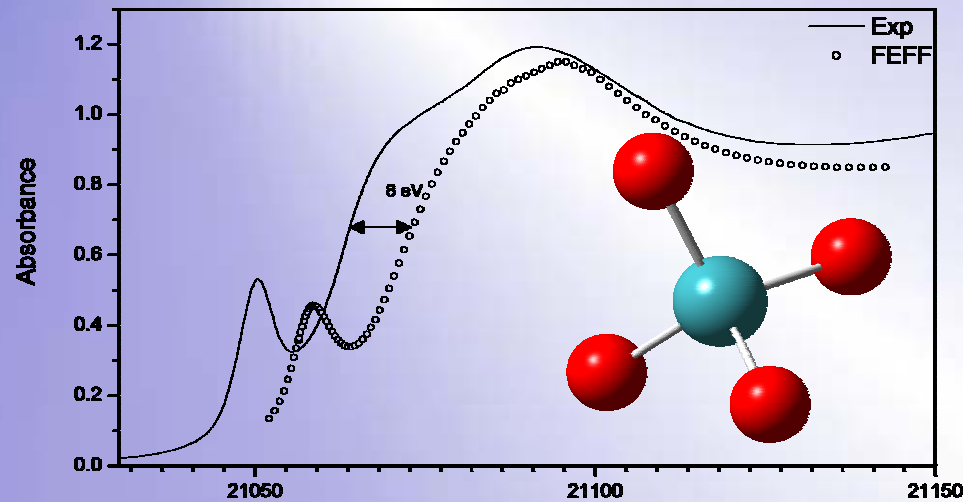
- feff 8.2
  - SCF
  - all atom

## - EXAFS data fitting

- k-space
- $S_0 = 1$
- $E_0 \cong -7$ , float, paths equal
- N = fixed
- $\sigma^2 = \text{float}$
- R = float



# Software Test $\text{TcO}_4^-$

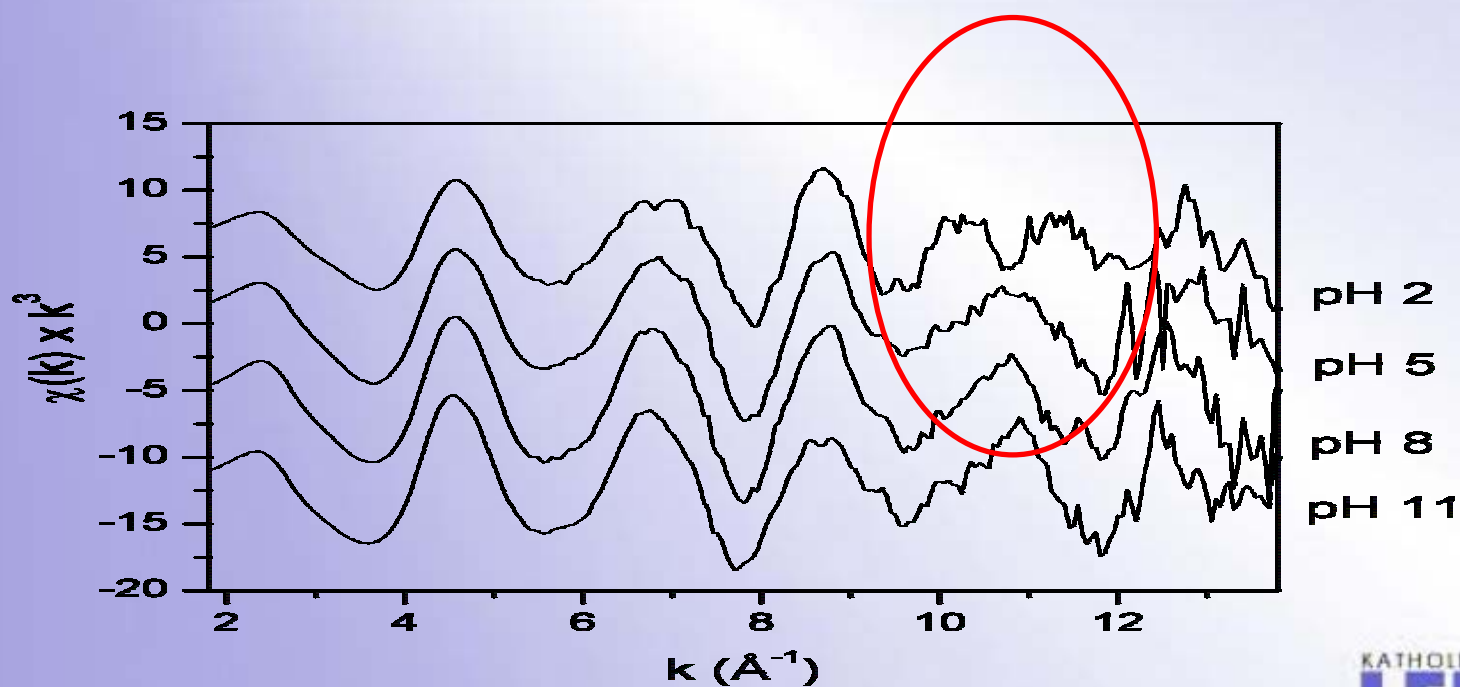


Energy (eV)	DFT	EXAFS		
	$d(\text{Tc-O})$ ( $\text{\AA}$ )	N	R ( $\text{\AA}$ )	$\sigma^2$
SS: Tc-O	1.75561	4	1.725	0.0016 4
MS: Tc-O-O		12	3.191	0.0057 7



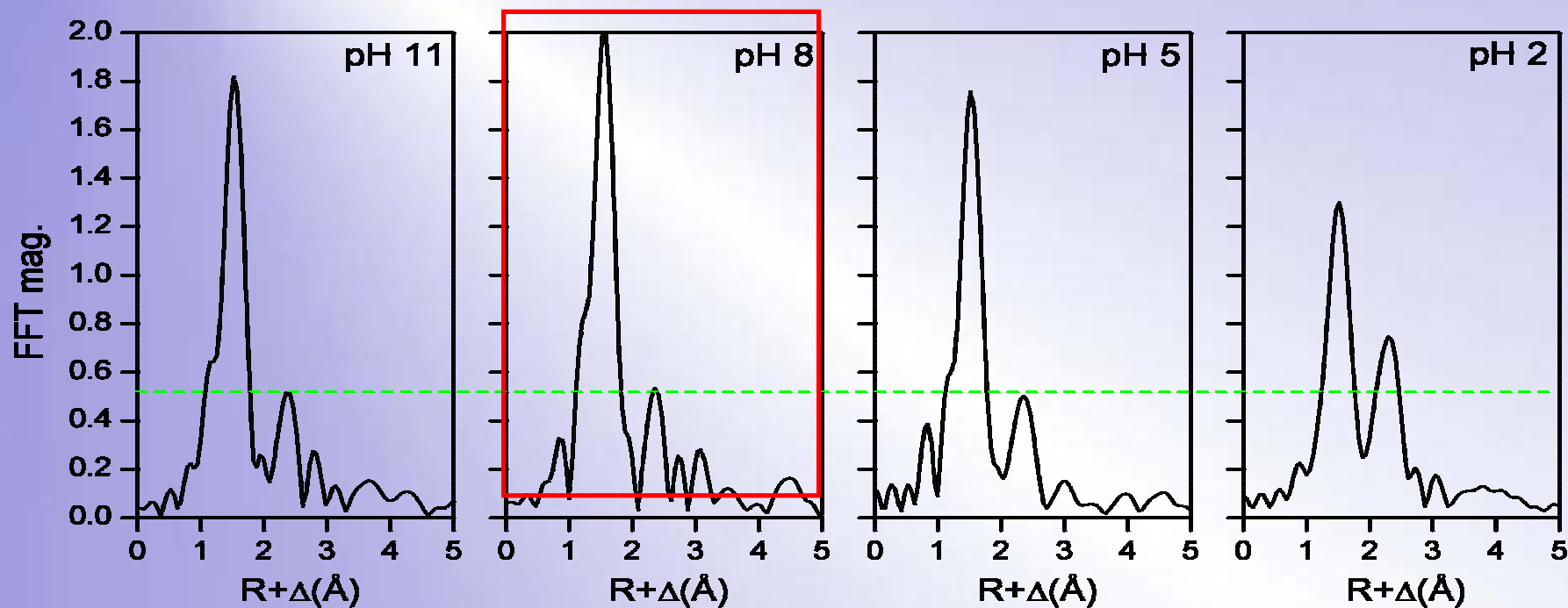
# Tc(IV)-pyrogallol ( $\text{N}_2\text{H}_4$ )

- amplitude difference
- pH 2  $\leftrightarrow$  pH 5, 8, 11



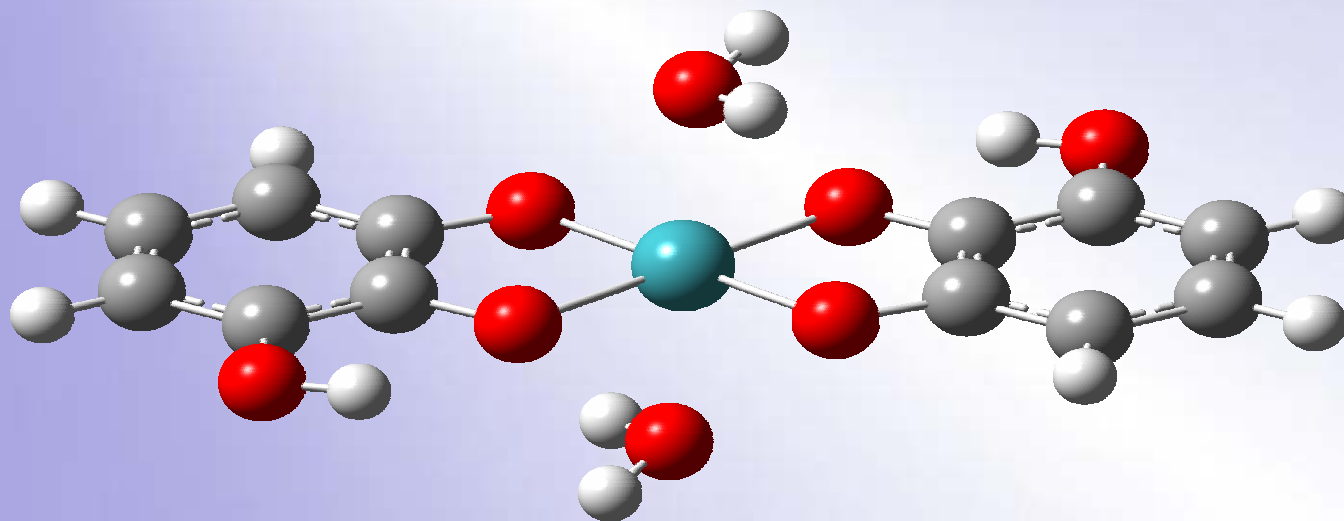


# Tc(IV)-pyrogallol ( $\text{N}_2\text{H}_4$ )





# Tc(IV)-pyro ( $\text{N}_2\text{H}_4$ ) – pH 8





# pH 8 – EXAFS FIT

bond	DFT	EXAFS	$\sigma^2$
Tc - O_eq	2.003	1.995	0.00093
Tc - O_ax	2.186	2.109	0.00130
Tc - C1	2.827	2.805	0.00157
Tc - C2	4.188	4.065	0.00208
Tc - O_pyr	4.788	4.79	0.00253

MS path	R	$\sigma^2$
Tc - O_eq - C1	3.09	0.00255
Tc - O - Tc - O	4.01; 4.10	0.00521
Tc - C1 - C2	4.19	0.00817
Tc - O_eq - O_pyr	4.81	0.00506





# Tc(IV)-pyro ( $N_2H_4$ ) – pH 8

