

Poly(amino-acid)-mucin interactions: the role of the charge density



Gergely Stankovits^{1*}, Ágnes Ábrahám², Éva Kiss², Zoltán Varga^{1,3}, Benedetta Attaianesi⁴, Ruth Cardinaels⁴, András Szilágyi¹, Benjámín Gyarmati¹

¹Department of Physical Chemistry and Materials Science, Faculty of Chemical Technology and Biotechnology, Budapest University of Technology and Economics, Műegyetem rkp. 3., H-1111 Budapest, Hungary

²Laboratory of Interfaces and Nanostructures, Institute of Chemistry, Eötvös Loránd University, Pázmány Péter Sétány 1/A, H-1117 Budapest, Hungary

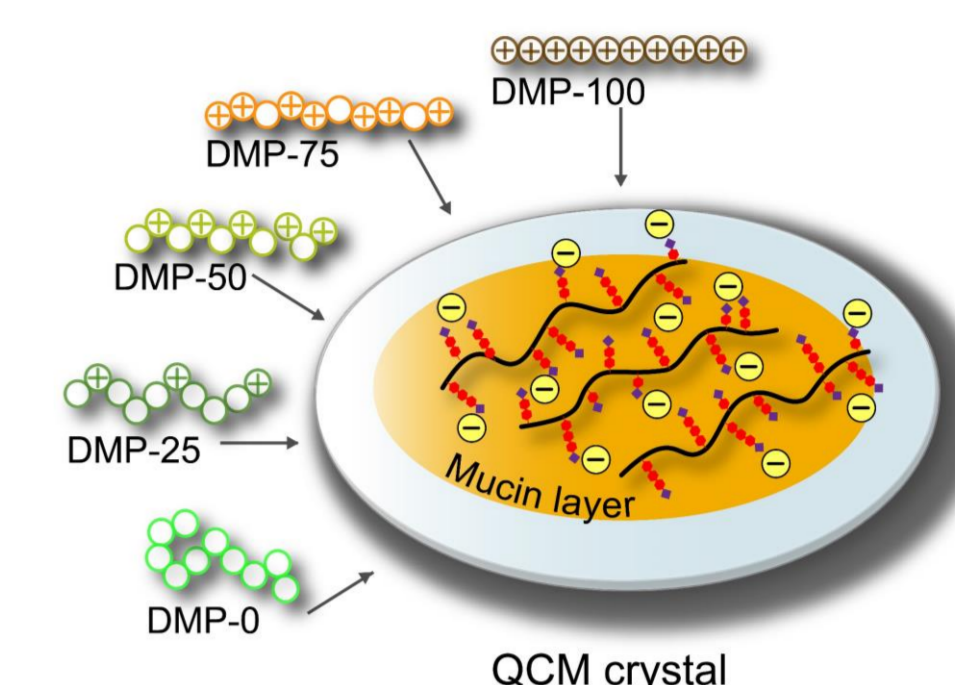
³Research Centre for Natural Sciences, Institute of Materials and Environmental Chemistry, Magyar tudósok körútja 2, 1117 Budapest, Hungary

⁴Soft Matter Rheology and Technology, Department of Chemical Engineering, KU Leuven, Celestijnenlaan 200f, Box 2424, B3001, Leuven, Belgium

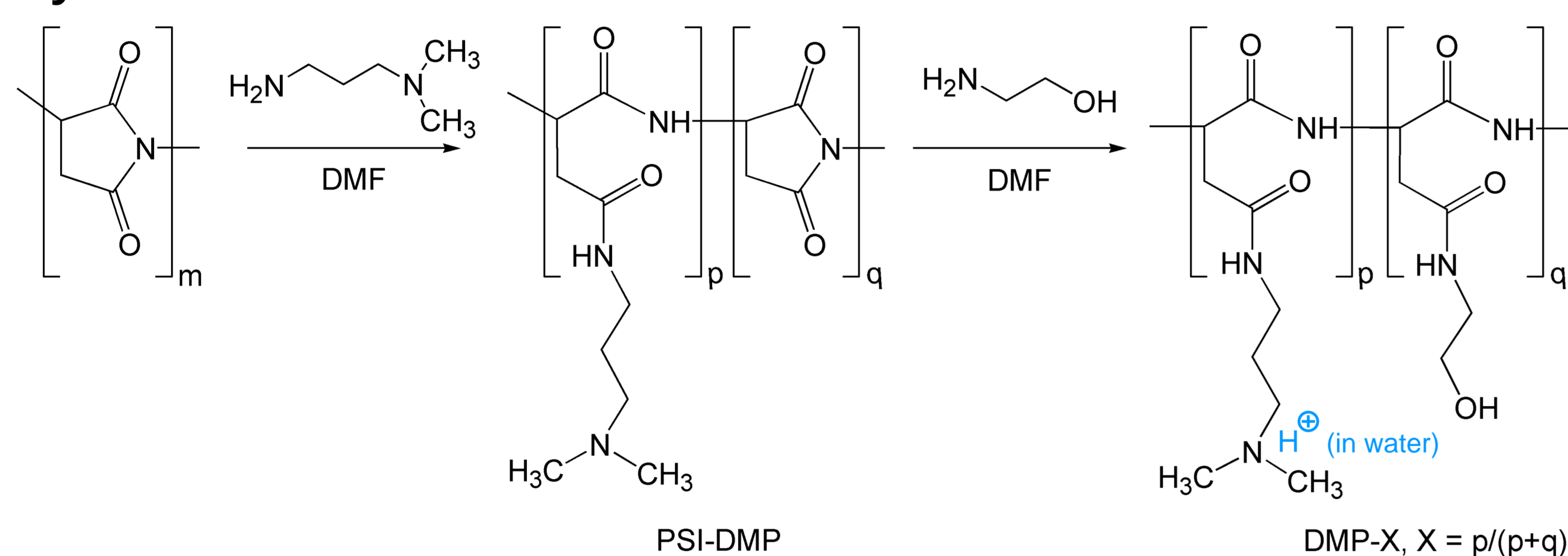
E-mail: softmatter@mail.bme.hu

Introduction

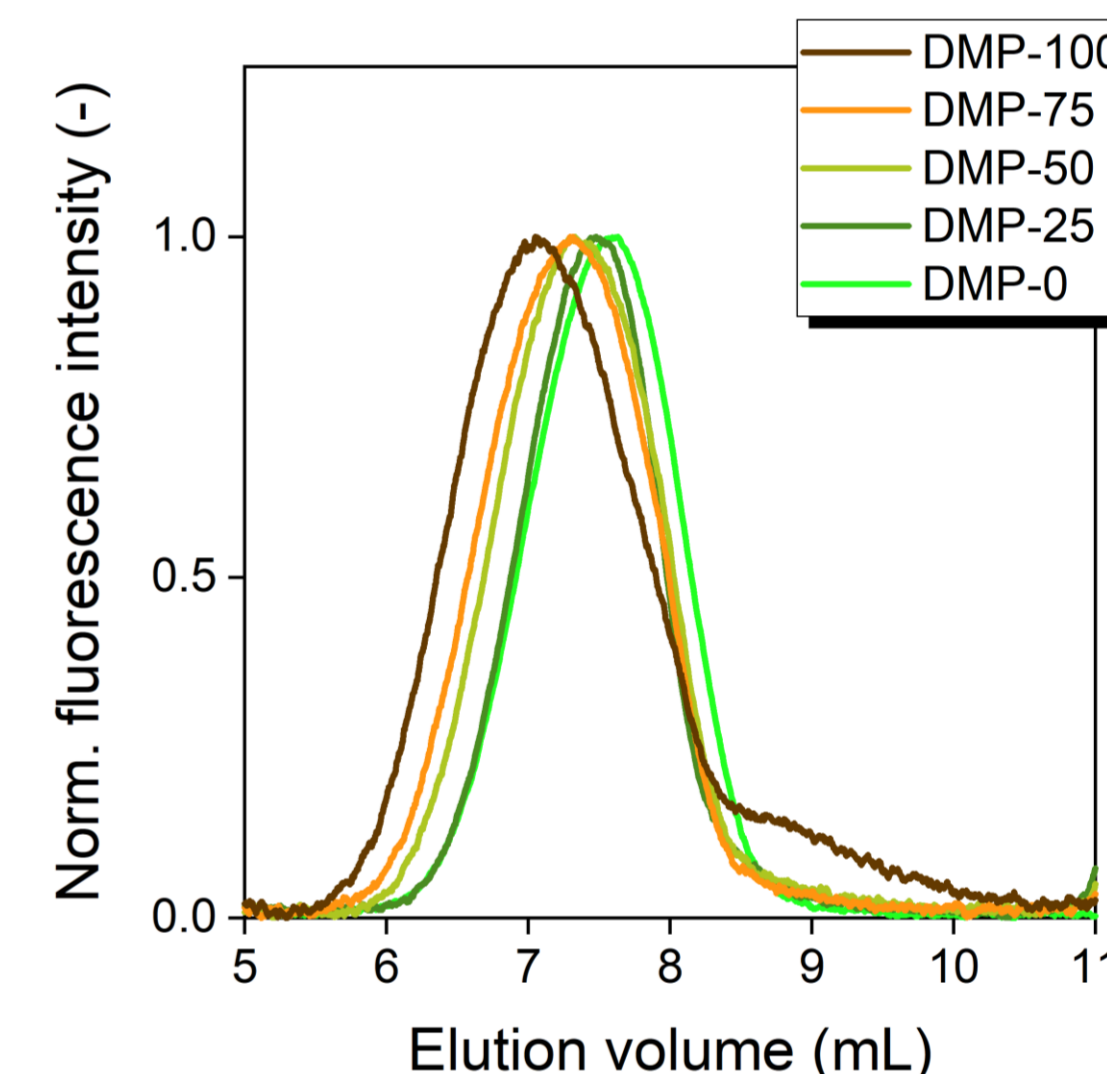
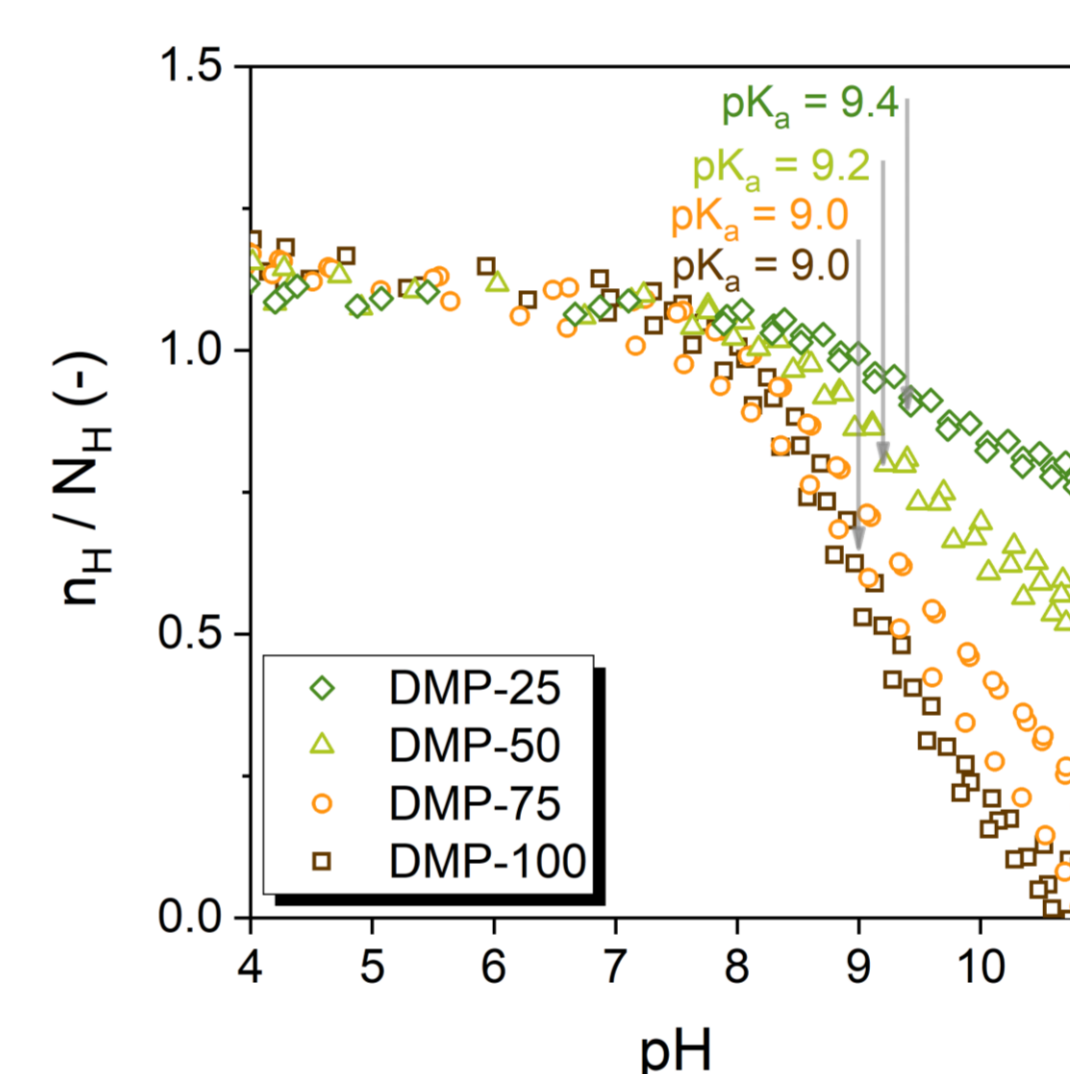
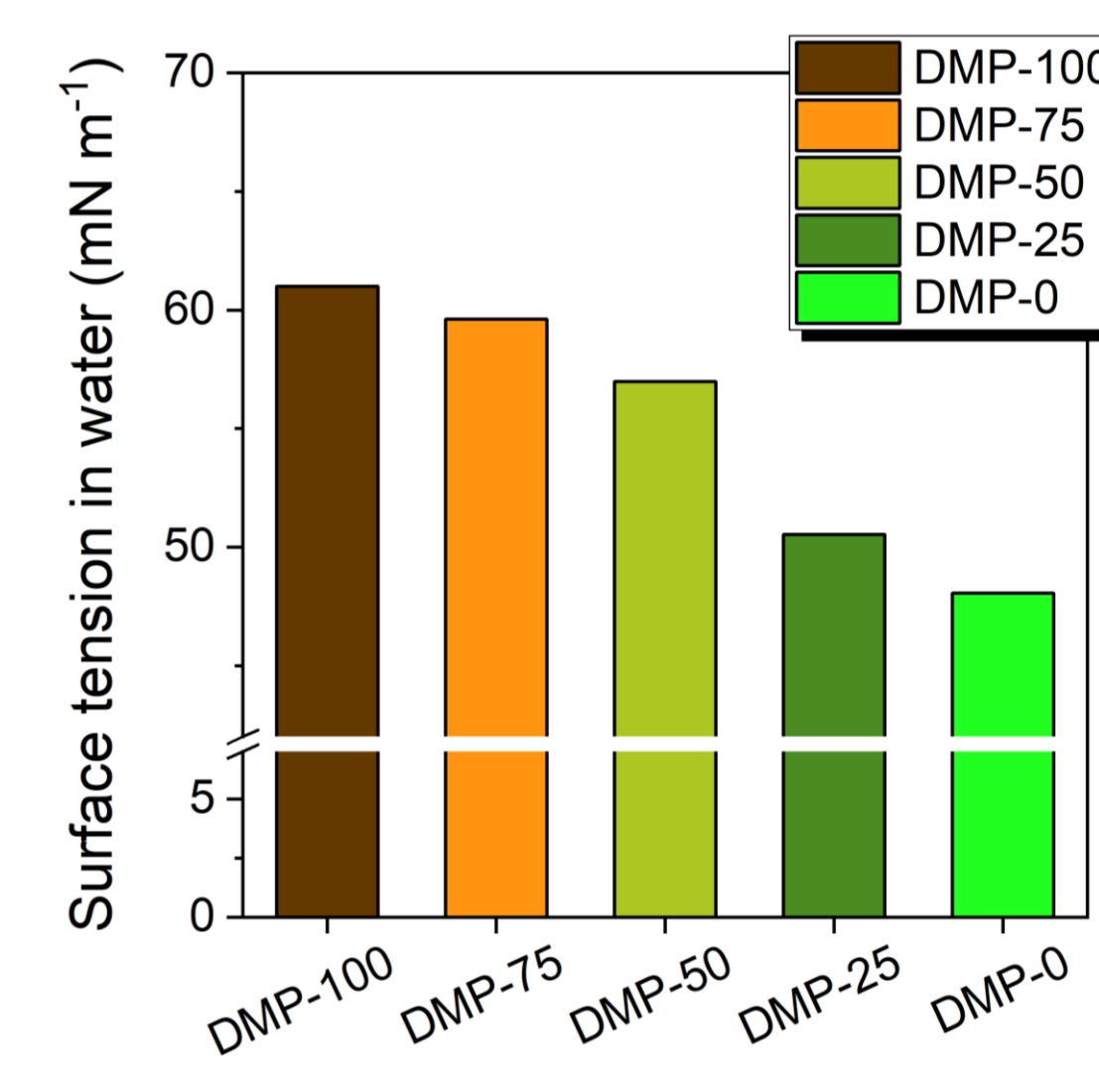
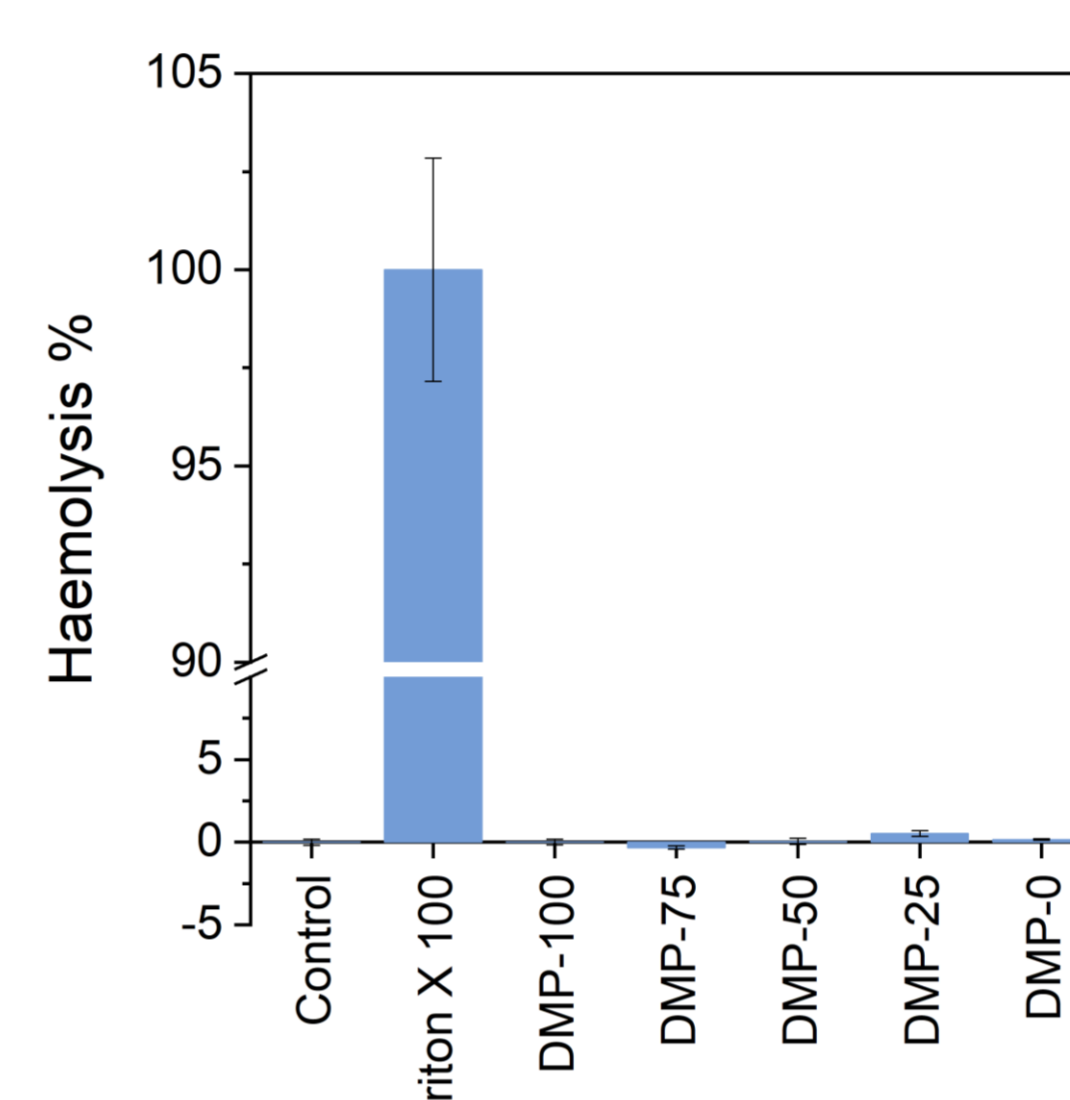
Mucosal surfaces are important targets for drug formulations as they provide high permeability and enable avoiding first-pass metabolism [1]. The two main concepts for utilising these surfaces are mucoadhesion and mucopenetration; both require the fine-tuning of colloidal interactions between the polymer excipient(s) of the formulations and the mucin proteins, the major component of the mucus secreted by the mucosa. Our goal here is to reveal the role of charge density in mucin-polymer interactions both in dispersion and in thin layers [2]. To this end, the class of polyaspartamides was chosen because of their chemical versatility, biocompatibility and biodegradability [3].



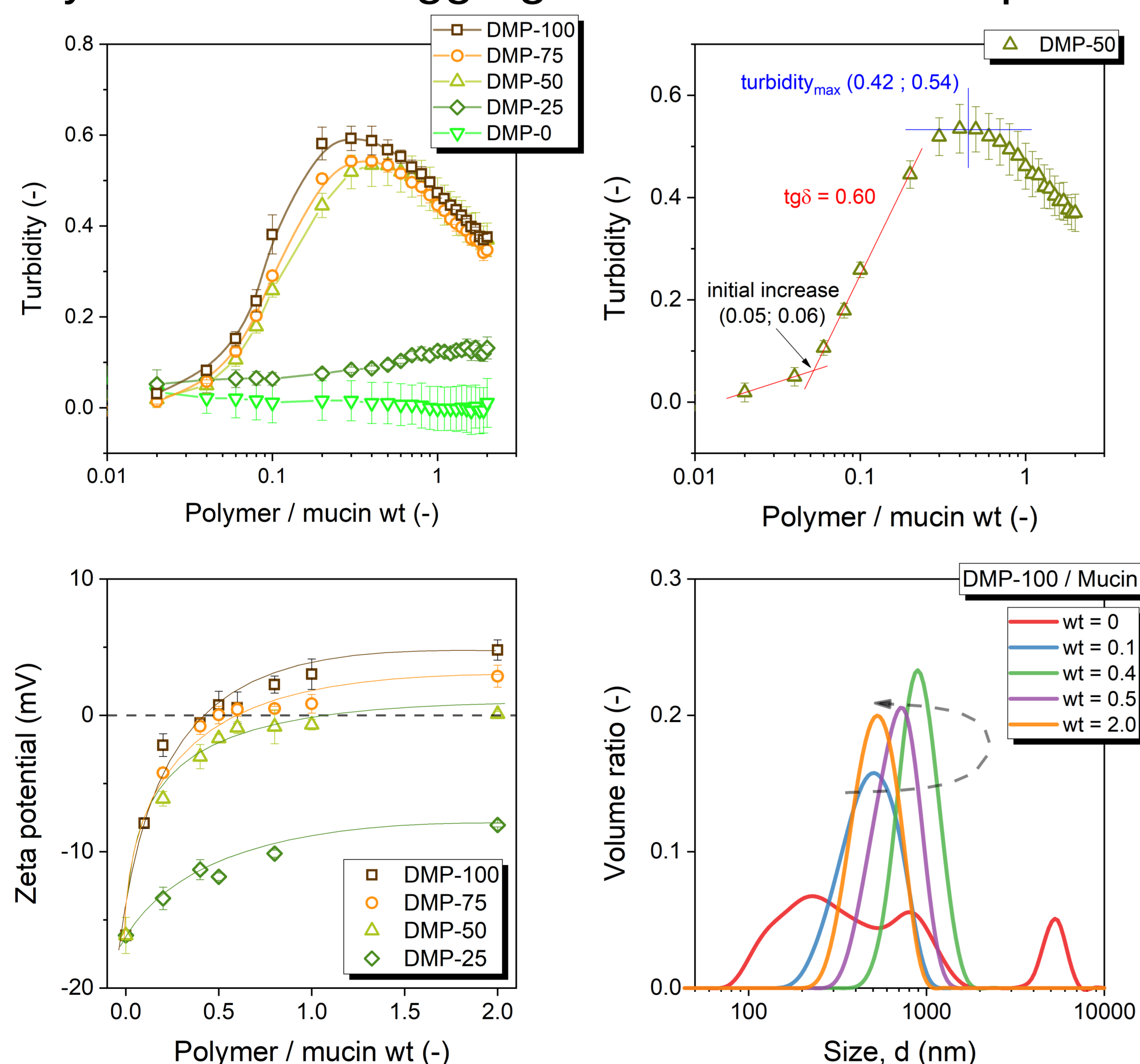
Synthesis and characterisation



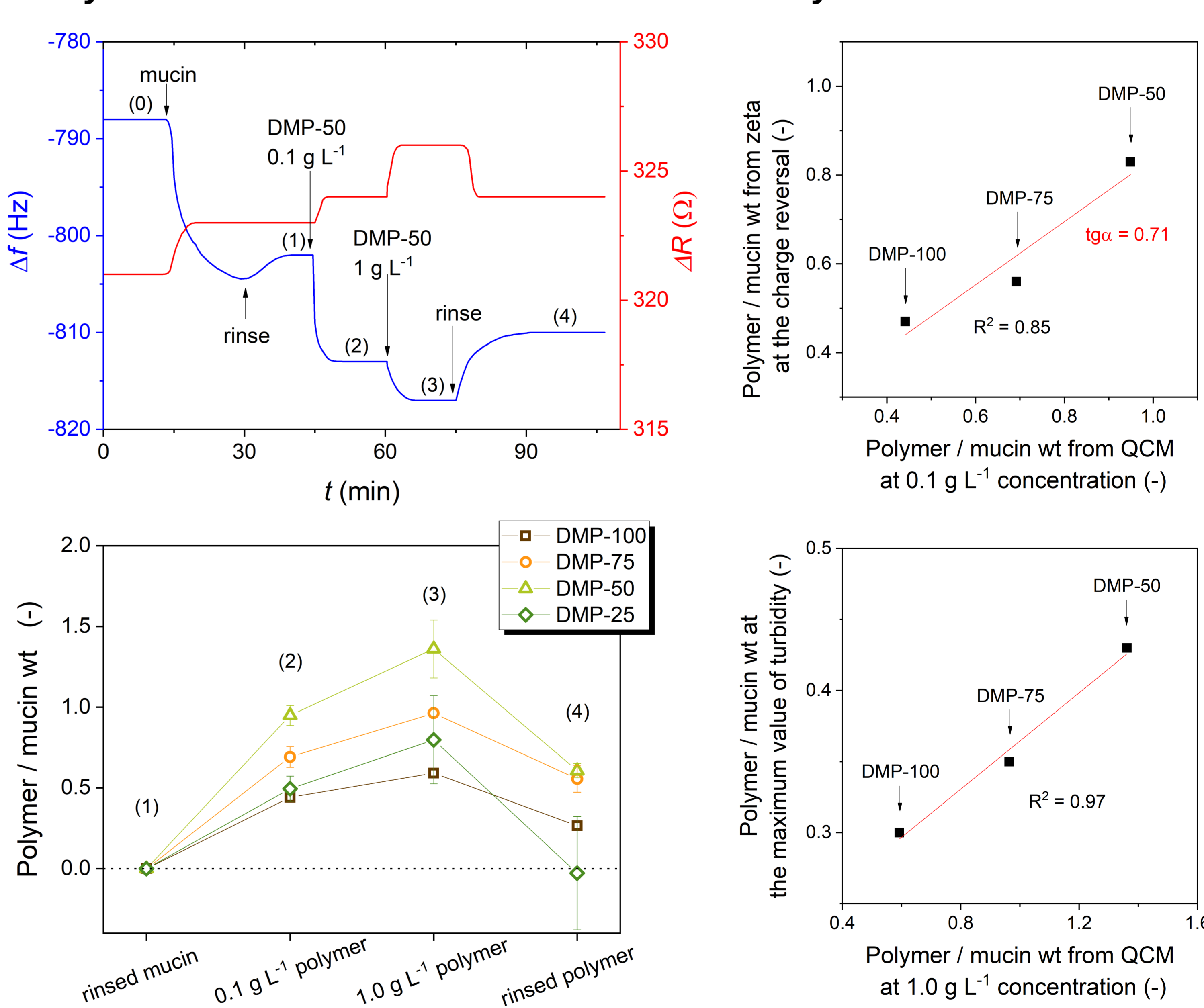
Composition	X _{DMP, NMR} (%)	X _{DMP, potentiometry} (%)	pK _a (-)	Zeta potential (mV)	d _{h, SEC} (nm)
DMP-100	98	100	9.0	11.4 ± 1.0	8.3
DMP-75	74	75	9.0	9.8 ± 2.7	7.4
DMP-50	51	50	9.2	8.3 ± 1.7	7.0
DMP-25	23	28	9.4	0.6 ± 1.0	6.5
DMP-0	0	-	-	-4.2 ± 3.9	6.2



Polymer-induced aggregation of mucin in dispersion



Polymer-mucin interaction in a thin layer



Conclusions

- Cationic polyaspartamides with controlled cationic group content were synthesised
- The strength of mucin-polymer interactions changed non-linearly with composition
- The highest adsorption was at intermediate cationic side group content shown by QCM
- A close correlation was found between the bulk (3D) and thin layer (2D) interactions

References

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Acknowledgements

Project no. TKP-6-6/PALY-2021 has been implemented with the support provided by the Ministry of Culture and Innovation of Hungary from the National Research, Development and Innovation (NRDI) Fund, financed under the TKP2021-NVA funding scheme. Further support was provided by the NRDI Office via grants FK 138029 and 2021-4.1.2-NEMZ_KI-2022-00026. B. Gyarmati acknowledges the János Bolyai Research Scholarship of the Hungarian Academy of Sciences. The work was also supported by the UNKP-22-5 New National Excellence Program (UNKP-22-5-BME-297) of the Ministry for Innovation and Technology from the source of the NRDI Fund. G. Stankovits is grateful for the scholarship of the National Talent Programme of Hungary