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<https://scholar.google.com.hk/citations?user=xl7Qd9AAA-AAJ&hl=zh-CN&oi=ao>



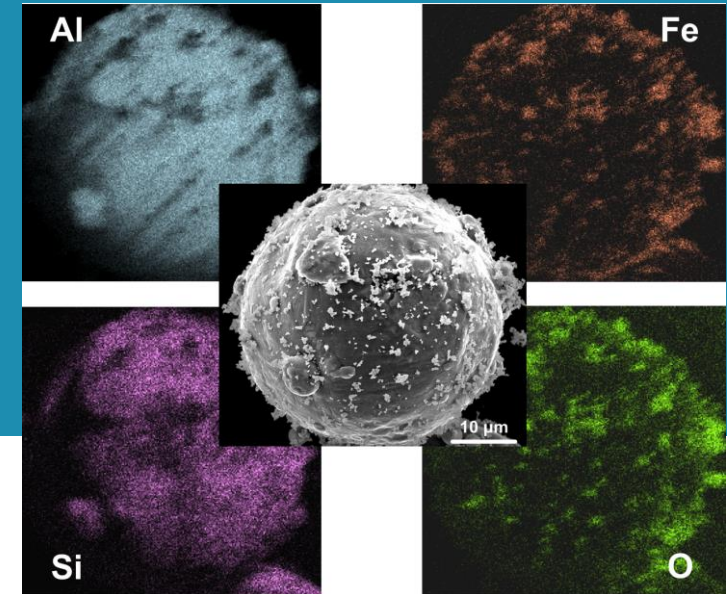
In-situ formation of particle reinforced Al matrix composites by laser melting + remelting strategy in laser powder bed fusion of $\text{Fe}_2\text{O}_3/\text{AlSi12}$ powder mixture

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Materials for Production | Barcelona, Spain, 16 Dec. 2021

Outline

- **Introduction**
- **Materials and methods**
- **Results and discussion**

Densification | Top surface & cross section morphology

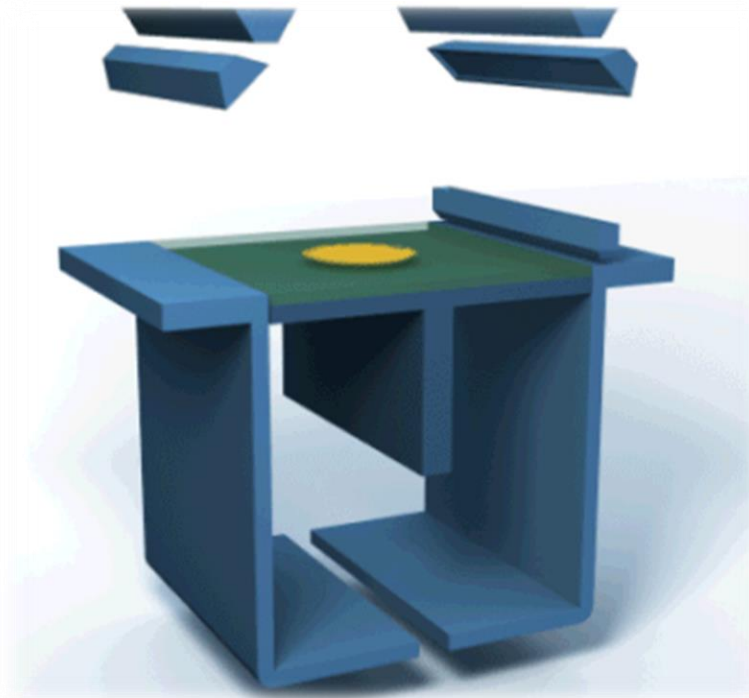
Microstructure & Phase | Microstructural evolution with phase analysis

Mechanical performance | Tensile property

- **Conclusions**

Introduction | LPBF

- **LPBF** is a metal additive manufacturing technology that uses a bed of powder with laser heat source to create metal parts in a layer-by-layer manner

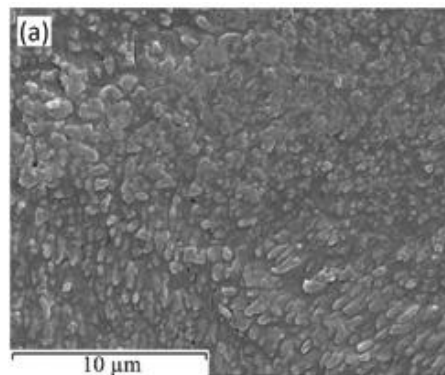


Introduction | Advantages

- **In-situ formation** of particle reinforced Aluminium matrix composites (PRAMCs) by LPBF of **$\text{Fe}_2\text{O}_3/\text{AlSi12}$ powder mixture** is a promising strategy to strengthen Al-based alloys

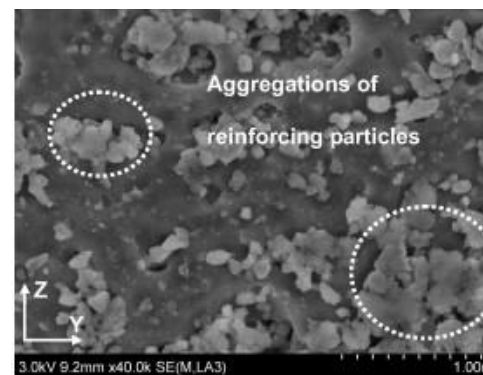
(1) 'In-situ' mechanism creates stronger interfacial bonding & better distribution of fine reinforcements, compared to the 'ex-situ' mechanism

In-situ



Uniform distribution of fine Al_2O_3 and FeAl_2O_4 particles in LPBF-fabricated parts from $\text{Fe}_2\text{O}_3/\text{Al}$ powder mixture
[Adv. Eng. Mater. 21(2019) 1801244]

Ex-situ

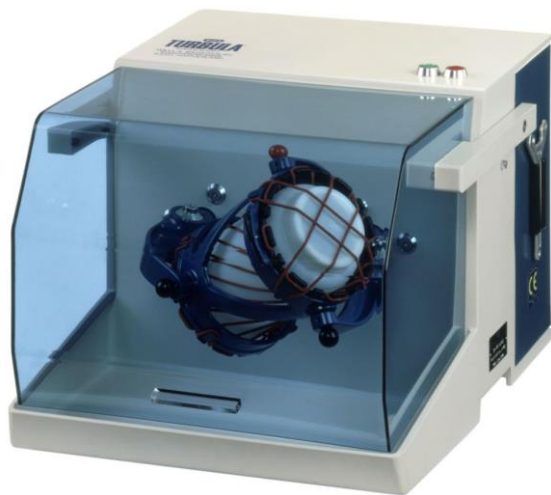


TiC aggregations in LPBF-fabricated TiC/Al composites
[Mater. Des. 82(2015) 46-55]

Introduction | Advantages

(2) $\text{Fe}_2\text{O}_3/\text{AlSi12}$ powder mixture is easy to prepare, compared to precursor powders

Direct mechanical mixing to prepare $\text{Fe}_2\text{O}_3/\text{AlSi12}$ powder mixture



3D shaker-mixer

Multiple procedures to prepare precursor powders

e.g., Preparation of graphene oxide coated AlSi10Mg powder to prepare:

Step 1, mechanical stirring and ultrasonication of AlSi10Mg in water in an ice-water bath for 1h;

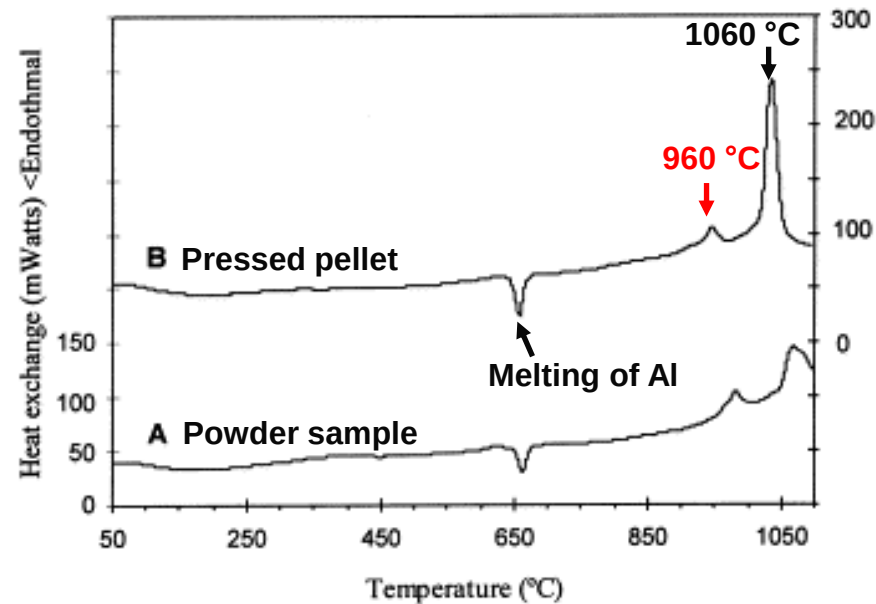
Step 2, adding graphene oxide drop-by-drop into the AlSi10Mg suspension under mechanical stirring;

Step 3, filtration and vacuum drying at 298 K for 24h.

[Mater. Charact. 170(2020) 110678]

Introduction | Advantages

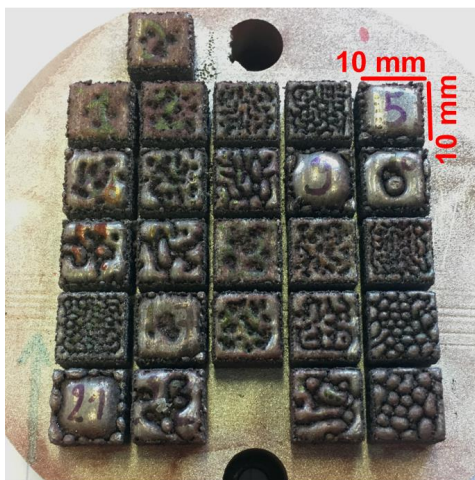
(3) In-situ $\text{Fe}_2\text{O}_3/\text{Al}$ reaction is easy to initiate at 960 °C



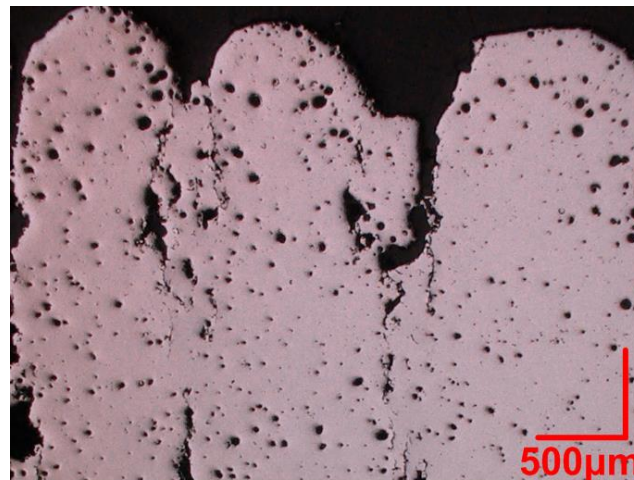
Differential thermal analysis of mixed $\text{Fe}_2\text{O}_3/\text{Al}$ samples
[Scripta Mater. 41(1999) 541-548]

Introduction | Challenge

- By-product O_2 reacts with Al, forming Al_2O_3 film that restricts the metal liquid's spreading and impact the densification
- **Objective:** densify the LPBF parts from the $Fe_2O_3/AlSi12$ powder mixture



Without laser remelting: Top surface of LPBF parts from the $Fe_2O_3/AlSi12$ powder mixture



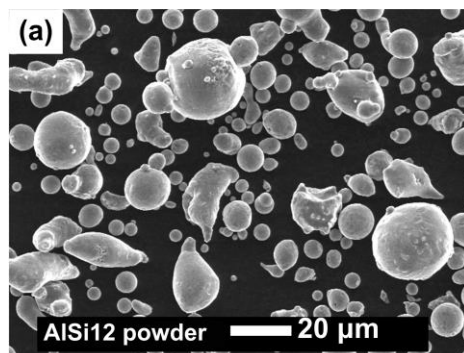
Without laser remelting: Cross section of LPBF parts from the $Fe_2O_3/AlSi12$ powder mixture

Materials and methods | Powder preparation

- Homogeneous 5wt%Fe₂O₃/AlSi12 powder mixture was obtained by mechanical mixing for 10 hours

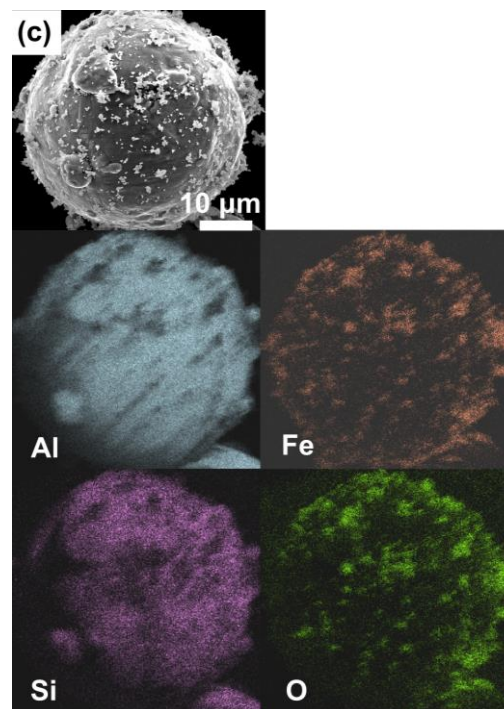
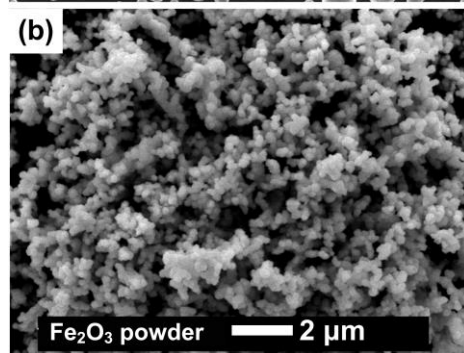
AlSi12

- Gas-atomized, spherical
- Particle size < 45 μm
- Average size = 25 μm



Fe₂O₃

- Spherical
- Average size = 200 nm
- Purity = 99.8%

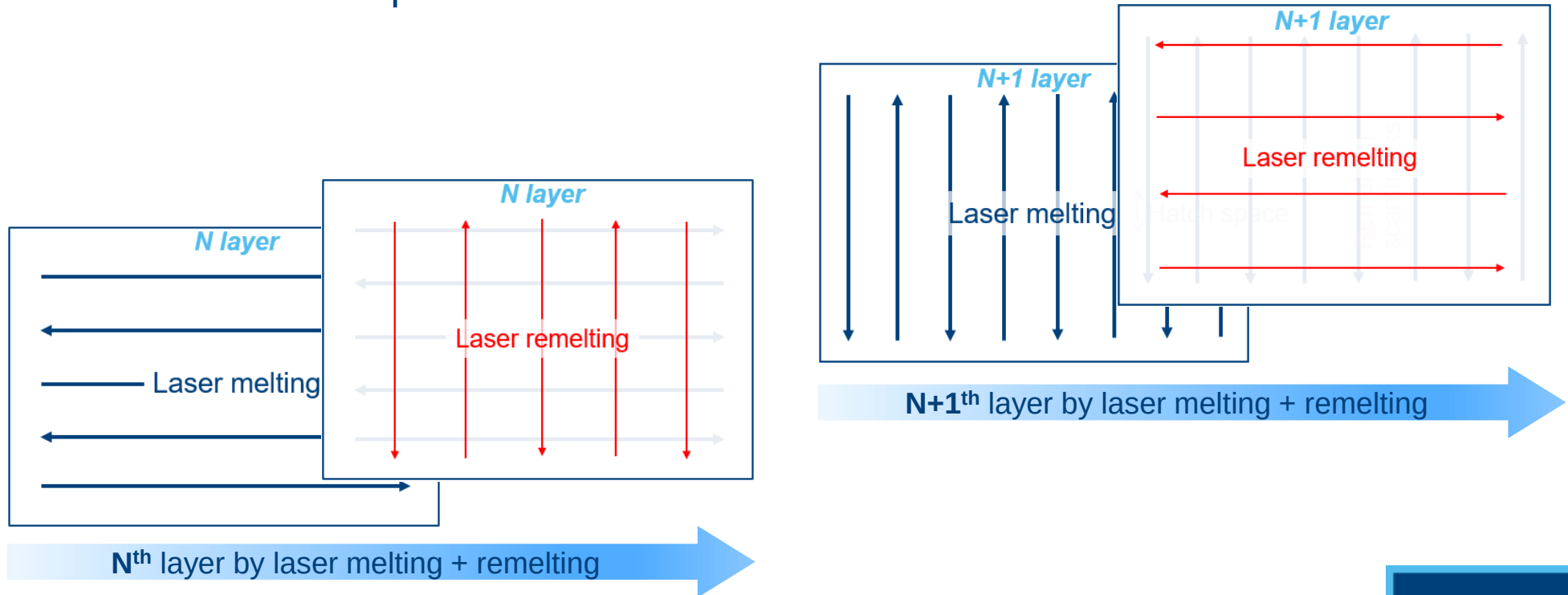


Powder mixture

- Fine Fe₂O₃ is well assembled on the surface of AlSi12

Materials and methods | Laser scan process

- A consecutive **high-energy melting scan** and **low-energy remelting scan** was used in the LPBF process



Materials and methods | Laser parameters

- Melting process optimization

Variable E_{melting} : 20, 25, 30, 35 J/mm²

Fixed $E_{\text{remelting}}$: 4.17 J/mm²



	P (W)	v (mm/s)	h (μm)	t (μm)
Melting	100-175	100	50	30
Remelting	250	600	50	/

- Remelting process optimization

Fixed E_{melting} : 35 J/mm²

Variable $E_{\text{remelting}}$: 2.92, 4.17, 4.29, 5.00 J/mm²



	P (W)	v (mm/s)	h (μm)	t (μm)
Melting	175	100	50	30
Remelting	210-350	600-1200	70-100	/

- Comparison group

Same laser melting parameters but no remelting

Laser energy surface density (J/mm²)

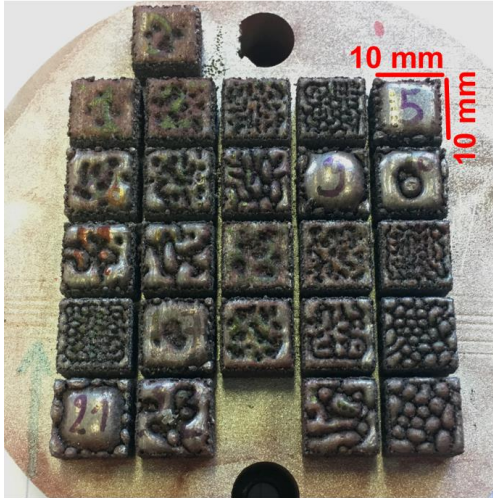
$$E_{\text{melting}} = P/v \cdot h$$

$$E_{\text{remelting}} = P/v \cdot h$$

P , laser power; v , laser scan speed;
 h , hatch space; t , powder laser thickness

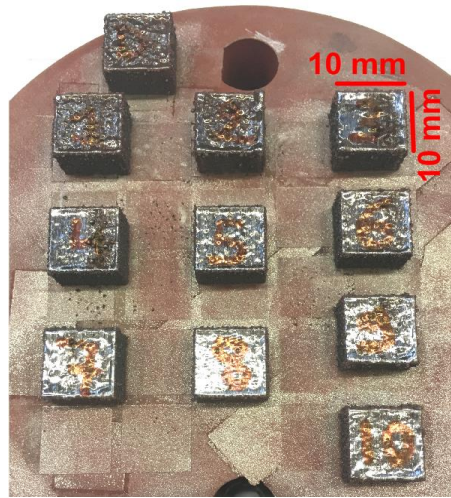
Materials and methods | Outcomes

- Big improvement of sample quality was achieved by laser melting + remelting

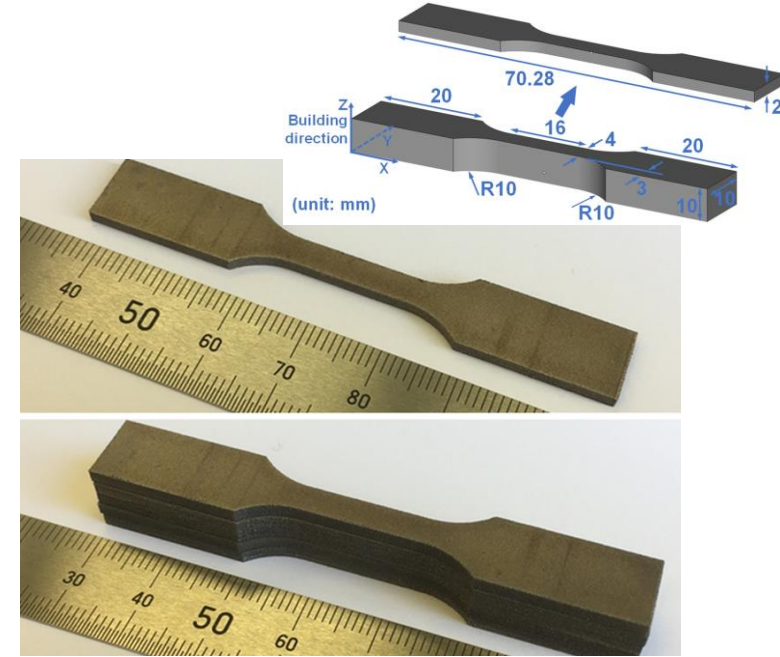


Only laser melting process

vs.



Laser melting + remelting process

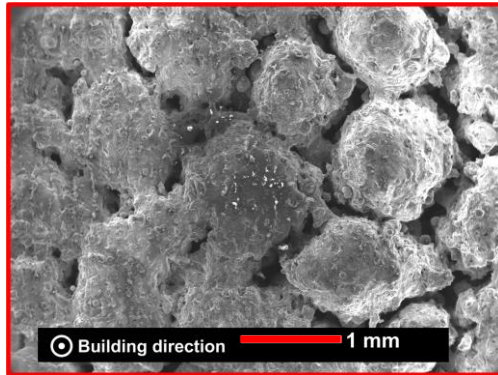


Tensile pieces by laser melting + remelting process

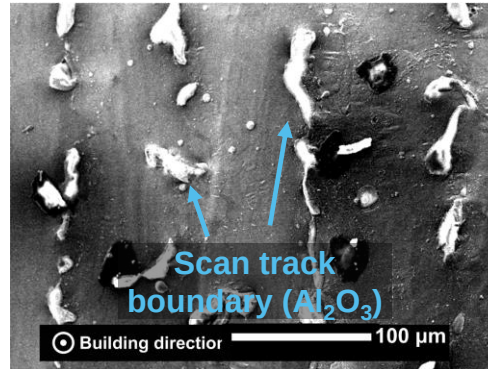
Densification | Top surface morphology

- Laser melting + remelting smoothed the top surface

E_{melting} of 35 J/mm²
No remelting!

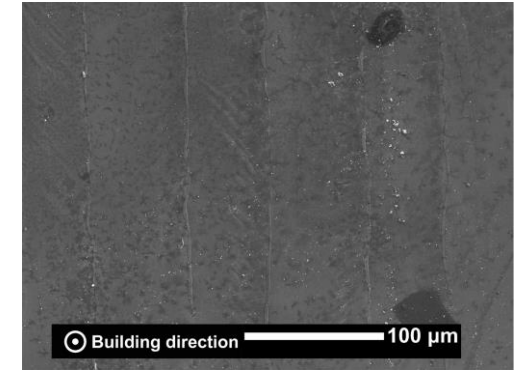


E_{melting} of 35 J/mm²
 $E_{\text{remelting}}$ of 4.17 J/mm²



Melting process optimization

E_{melting} of 35 J/mm²
 $E_{\text{remelting}}$ of 5.00 J/mm²

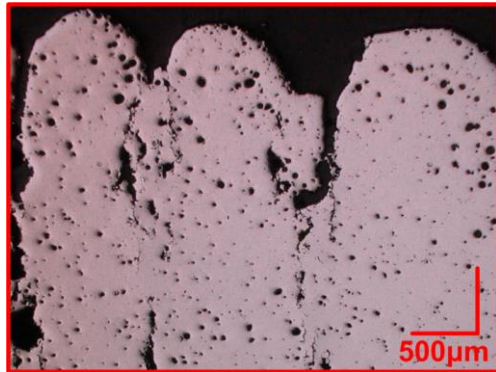


Remelting process optimization

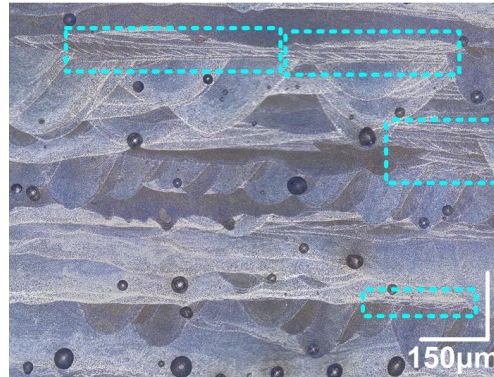
Densification | Side surface morphology

- Laser melting + remelting densified the sample by forming horizontally layered solidified structure

E_{melting} of 35 J/mm²
No remelting!



E_{melting} of 35 J/mm²
 $E_{\text{remelting}}$ of 4.17 J/mm²



Melting process optimization

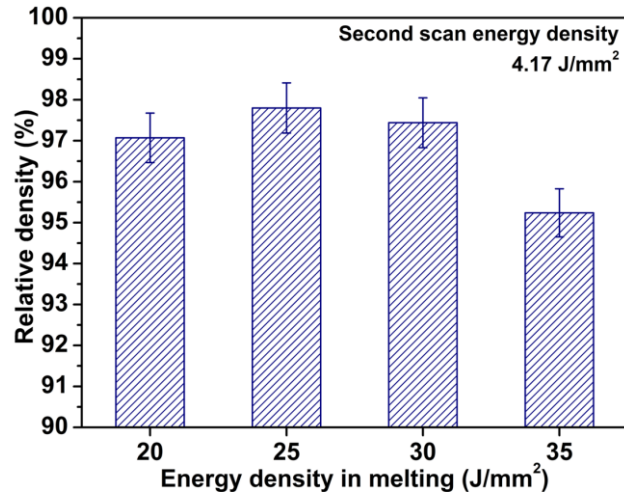
E_{melting} of 35 J/mm²
 $E_{\text{remelting}}$ of 5.00 J/mm²



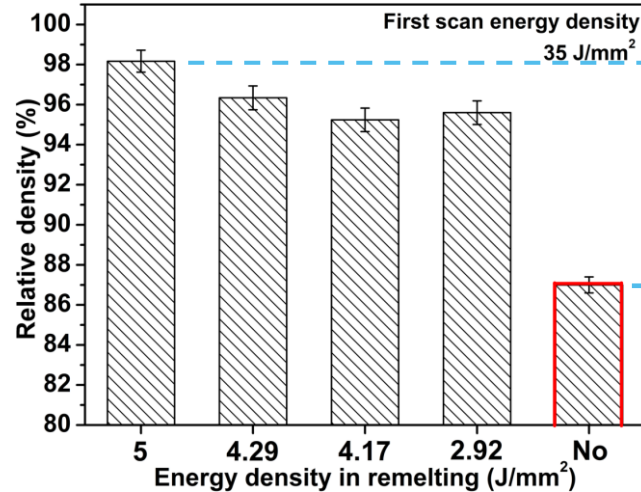
Remelting process optimization

Densification | Relative density

- Laser melting + remelting improved the relative density to $98.2 \pm 0.55 \%$



Melting process optimization

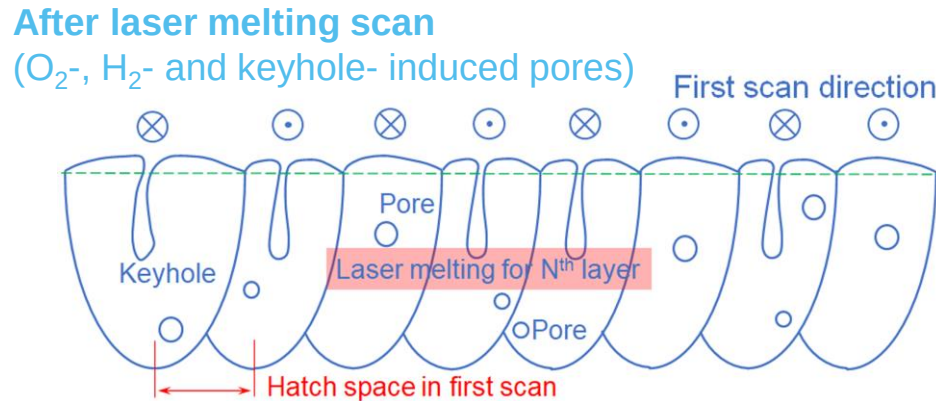


Remelting process optimization

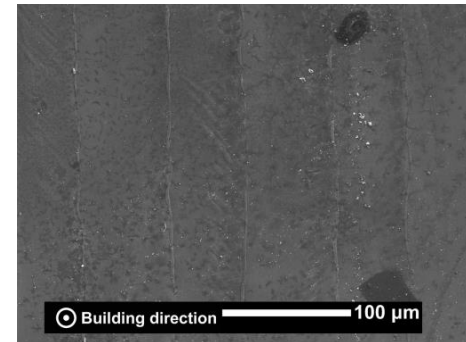
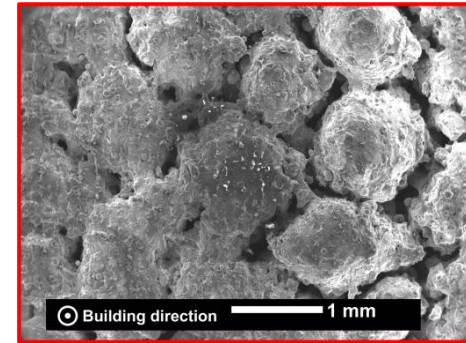
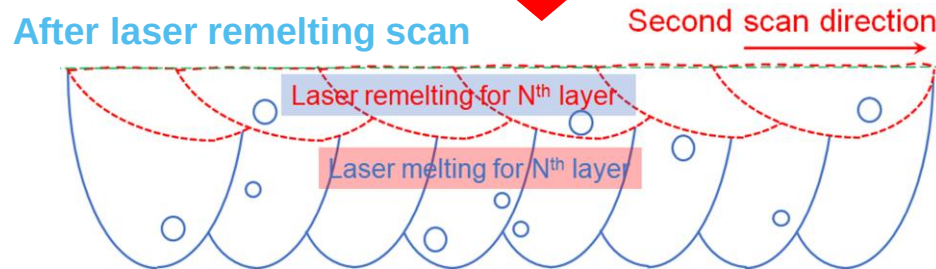
$98.2 \pm 0.55 \%$
Increment of 13 %
 $87.1 \pm 0.70 \%$
No remelting!

Densification | Mechanism behind densification

- Laser remelting smoothened the top surface and closed gaps layer by layer

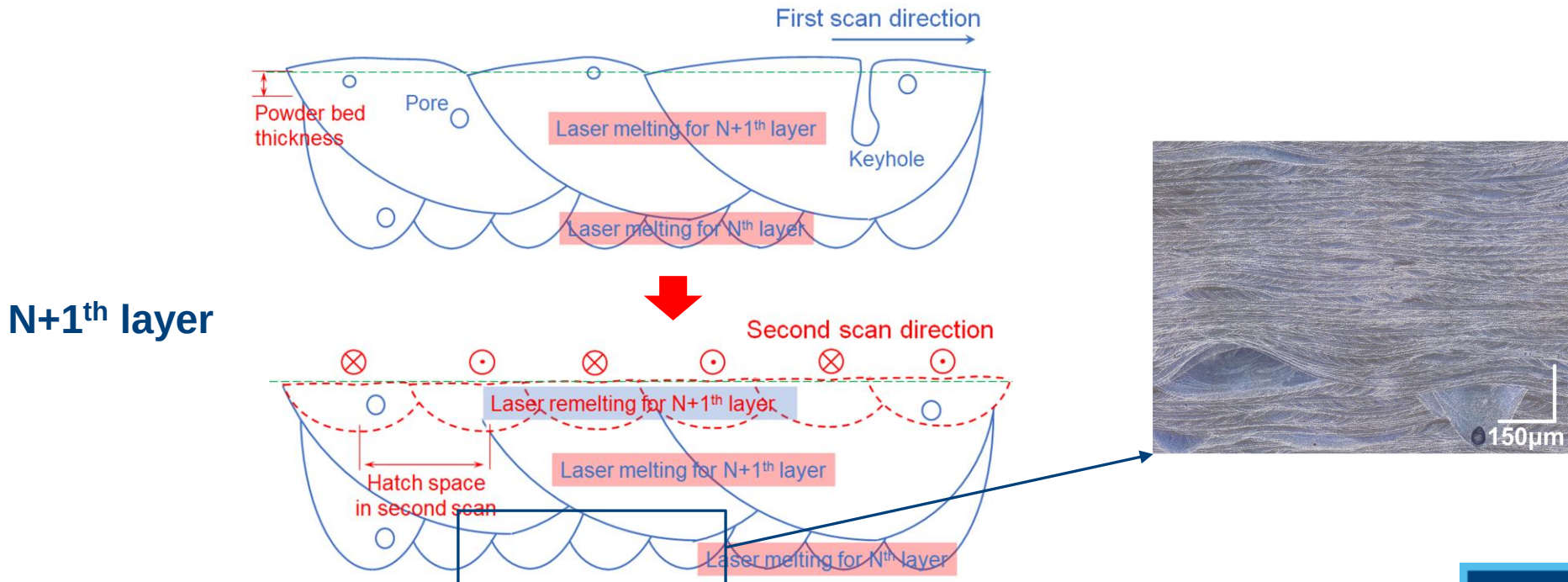


N^{th} layer



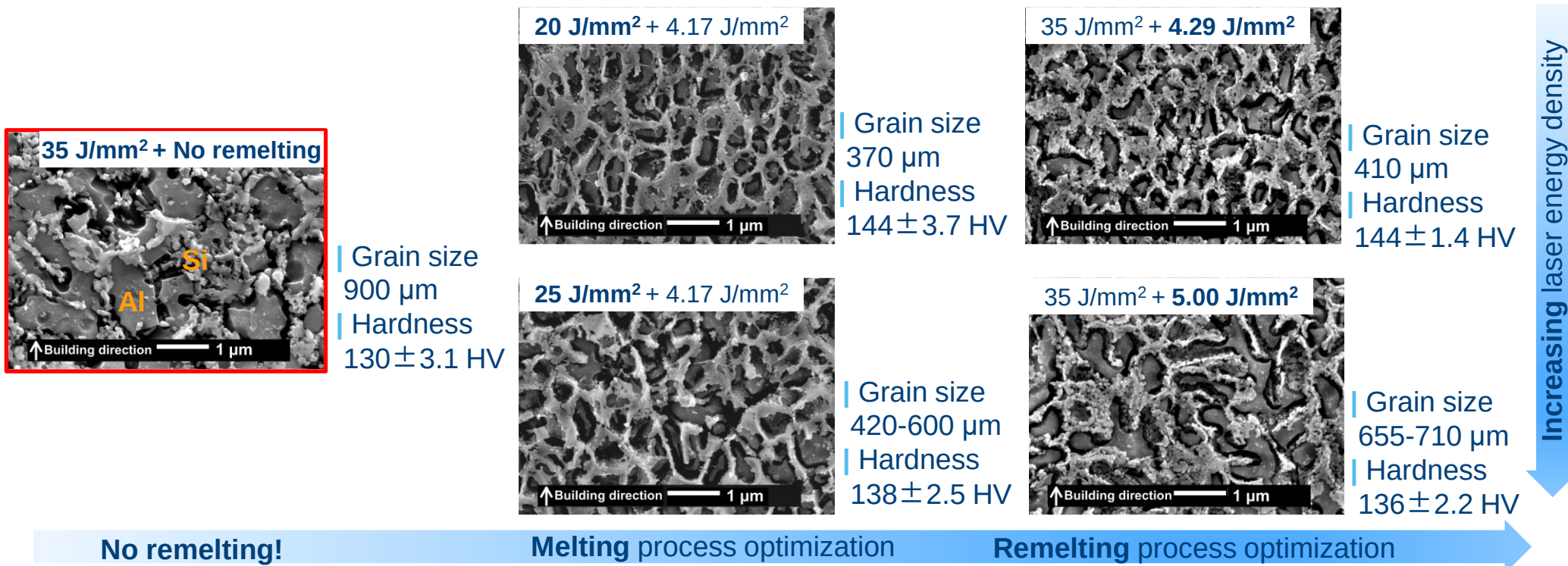
Densification | Mechanism behind densification

- Repeated laser melting + remelting produced fine horizontally layered solidified structure



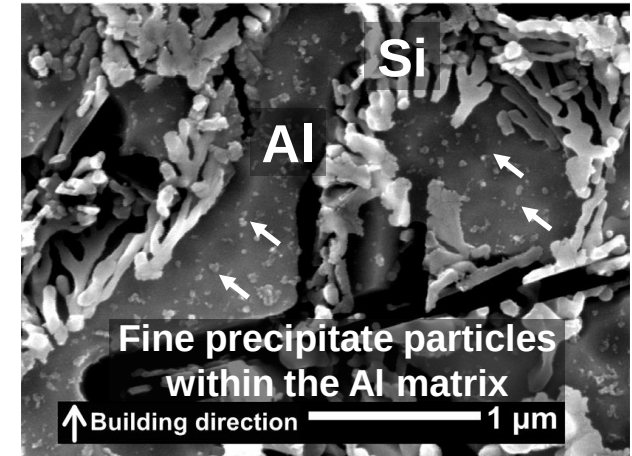
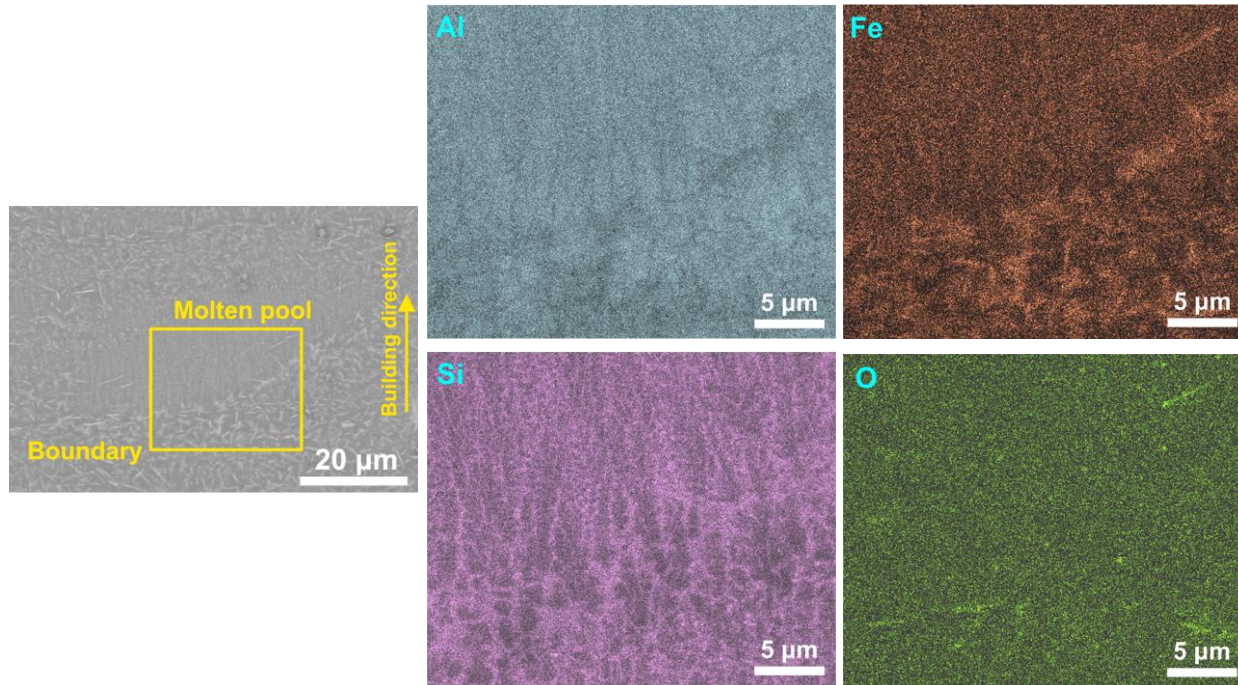
Microstructure & Phase | Microstructural evolution

- Laser melting + remelting did not coarsen the microstructure



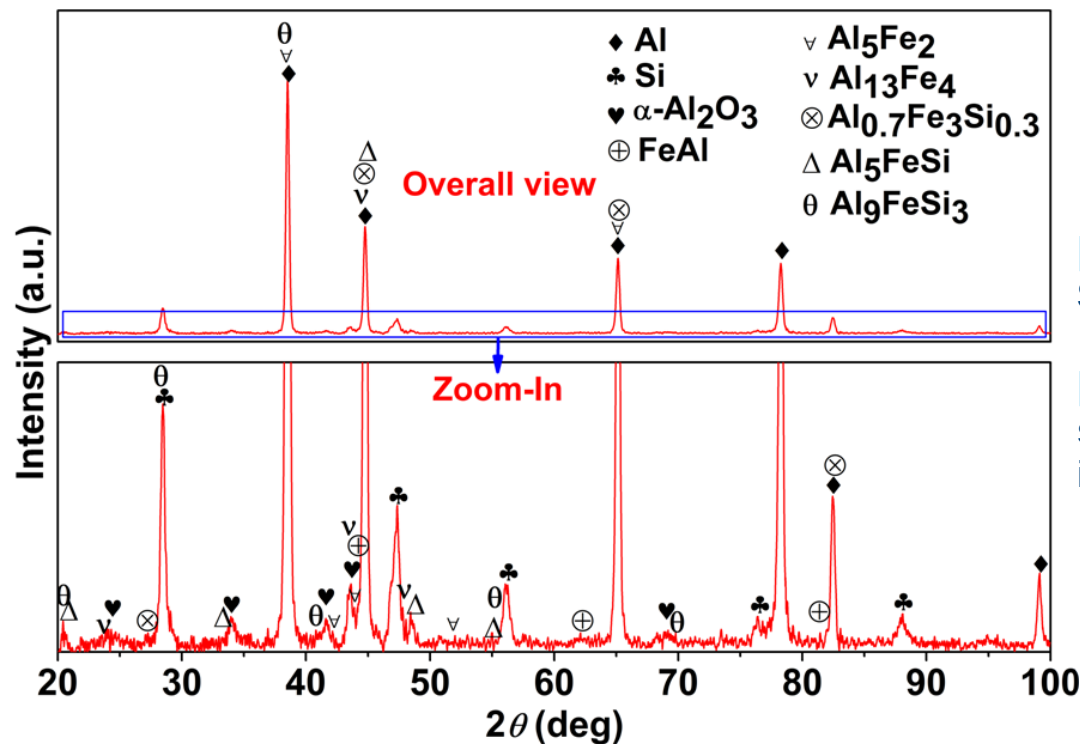
Microstructure & Phase | Precipitate distribution

- EDS maps confirm the distribution of precipitate-related elements (Al, Si, Fe and O) across the whole material



Microstructure & Phase | Phase evaluation

- Laser melting + remelting produced multiple precipitate phases

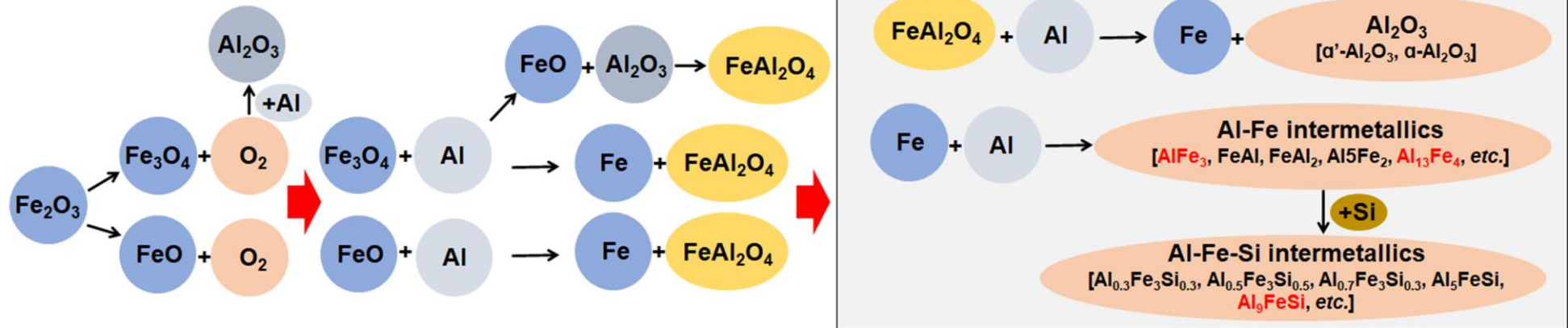


| Main phases of **Al matrix** and **Si particles**

| Minor phases of **Al₂O₃**, and **stable/ metastable Al-Fe(-Si) intermetallics**

Microstructure & Phase | Phase evaluation

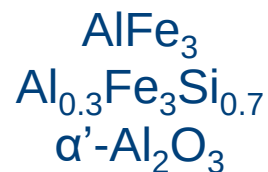
- In-situ $\text{Fe}_2\text{O}_3/\text{Al}$ reaction includes multiple steps, providing enough possibilities of forming precipitates [1,2]



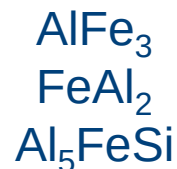
Microstructure & Phase | Phase evaluation

- Rapid cooling process of LPBF froze intermediate metastable precipitates

**Primary
metastable phases**



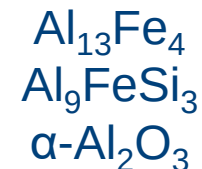
**Intermediate
metastable phases**



...

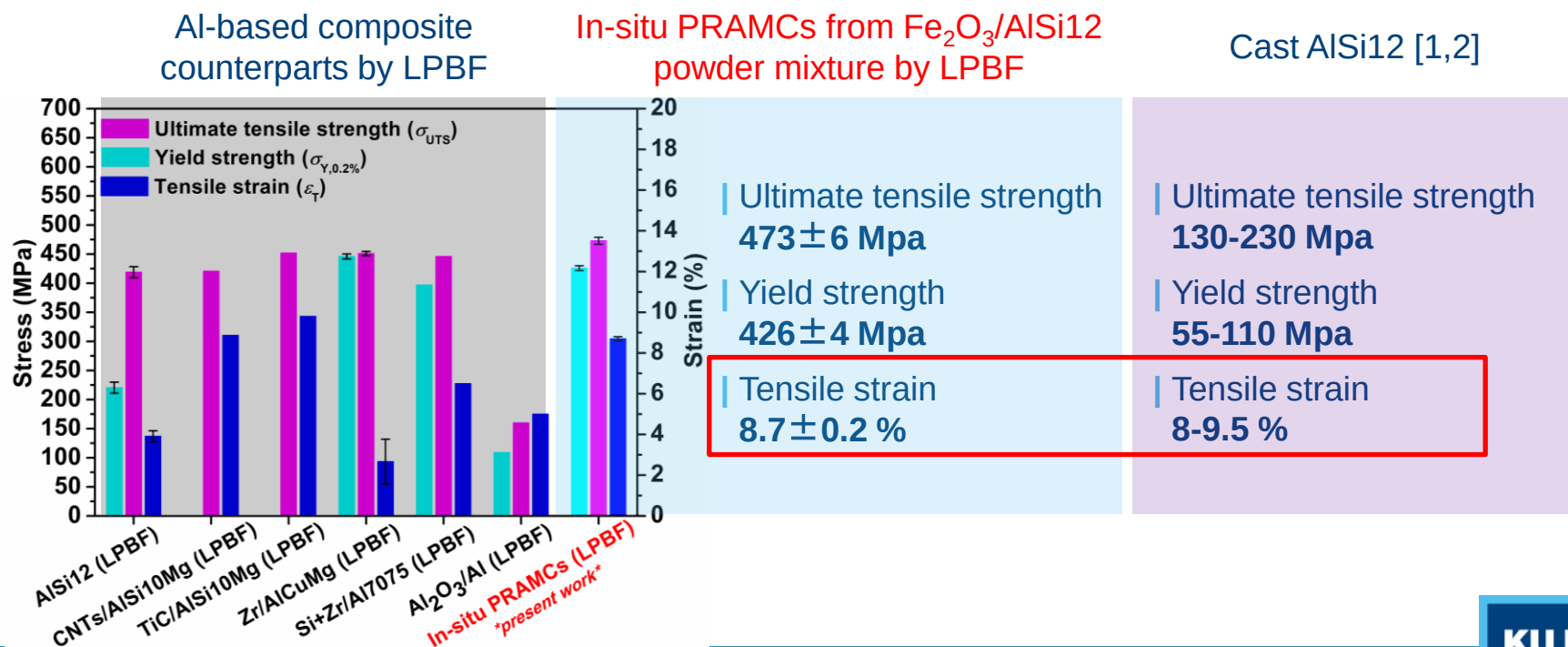


**Final
stable phases**



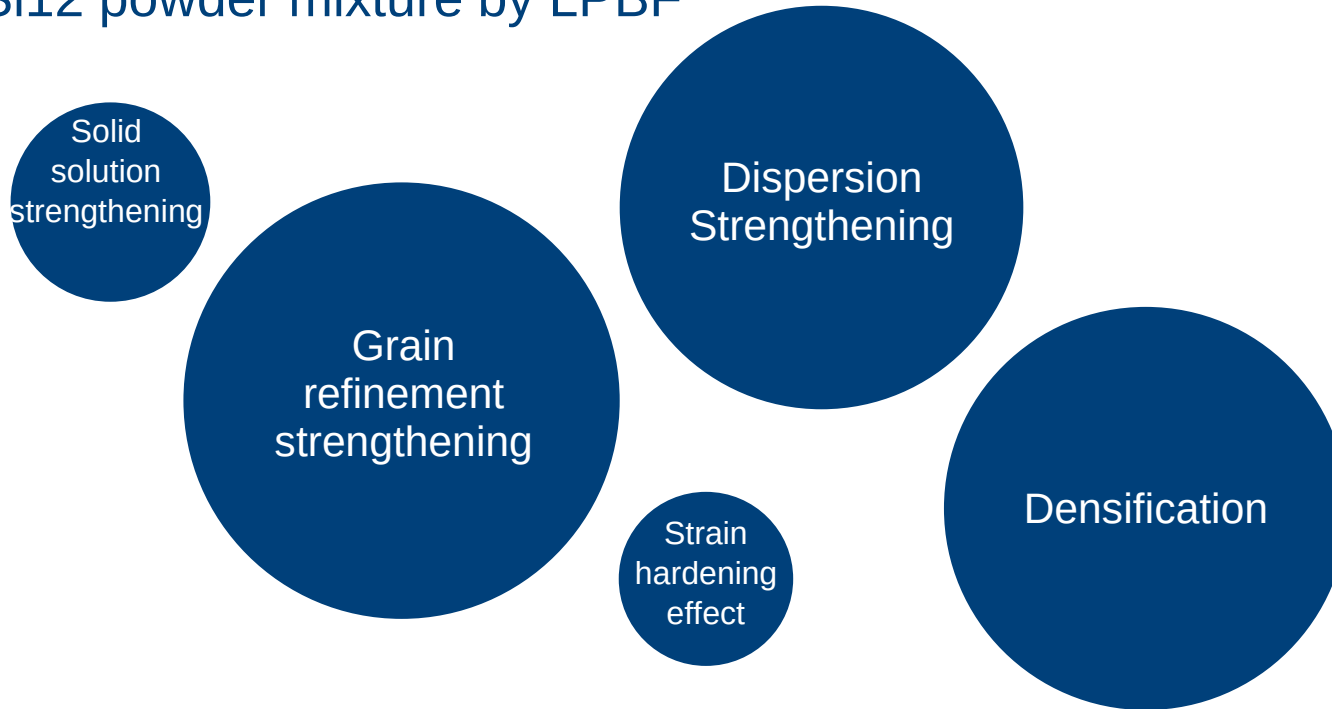
Mechanical performance | Tensile properties

- In-situ PRAMCs by LPBF have **comparable/ superior mechanical performance** to LPBF-fabricated Al-based composite counterparts/ cast AlSi12.



Mechanical performance | Strengthening mechanism

- Multiple contributions strengthened the in-situ PRAMCs made from the $\text{Fe}_2\text{O}_3/\text{AlSi12}$ powder mixture by LPBF



Conclusions

- The consecutive laser melting + remelting scan strategy can prepare **dense PRAMCs** from $\text{Fe}_2\text{O}_3/\text{AlSi12}$ powder mixture.
- The PRAMCs made from $\text{Fe}_2\text{O}_3/\text{AlSi12}$ powder mixture by LPBF achieve good mechanical properties.
- Laser melting + remelting could be practical to process other powder mixtures with severe porosity/ oxide-film formation potential

For further details

- Qimin Shi, *et al.*, In-situ formation of particle reinforced Aluminium matrix composites by laser powder bed fusion of $\text{Fe}_2\text{O}_3/\text{AlSi12}$ powder mixture using laser melting/remelting strategy, *Journal of Materials Processing Technology* 299 (2022) 117357.





Thank you for your attention

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Material characterization

- **OM:** Optical Microscope (Keyence VHX6000 microscope, Japan)
- **Etching:** Keller's reagent (95 mL H₂O, 2.5 mL HNO₃, 1.5 mL HCl and 1 mL HF for 60s)
- **XRD:** Seifert 3003 T/T X-Ray diffractometer, at 40 kV and 40 mA, step-scan type with a step width of 0.02° and exposure time of 1 s for each step, $2\theta = 20\text{-}100^\circ$
- **SEM:** Scanning Electron Microscope (Philips XL30 FEG, Eindhoven, Netherlands)
- **EDS:** Energy Dispersive Spectrometer (EDS; EDAX Inc., USA)
- **Hardness:** FV-700 microhardness tester, load of 500 g during 15 s
- **Tensile:** Instron 4467 testing facility, tensile rate of 0.2 mm/min