

## **Supporting information**

### **Selective removal of zinc from BOF sludge by leaching with mixtures of ammonia and ammonium carbonate**

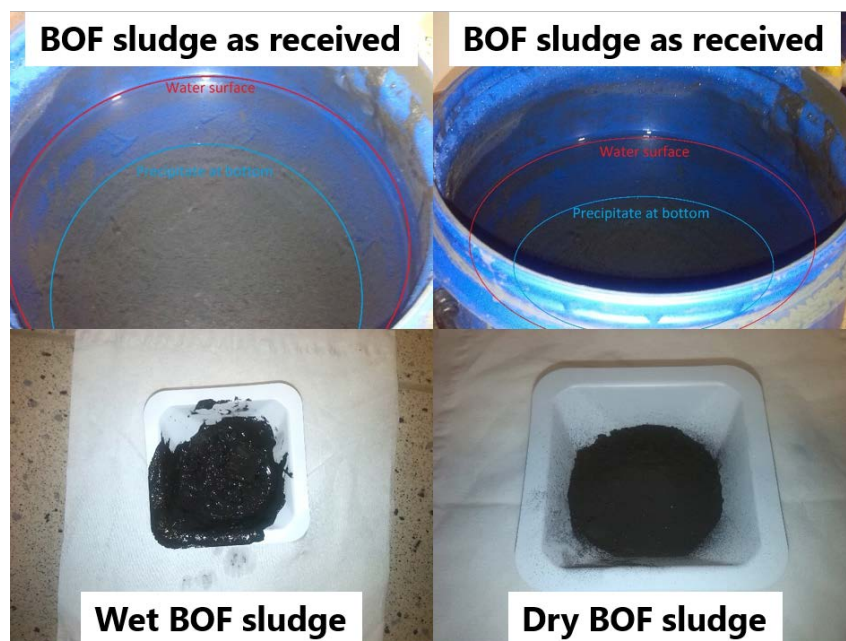
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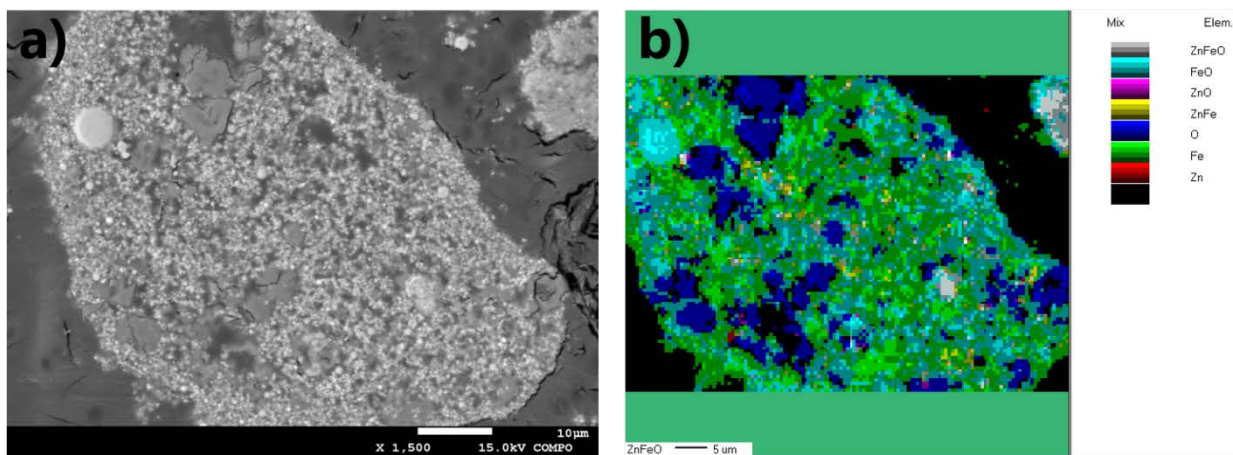
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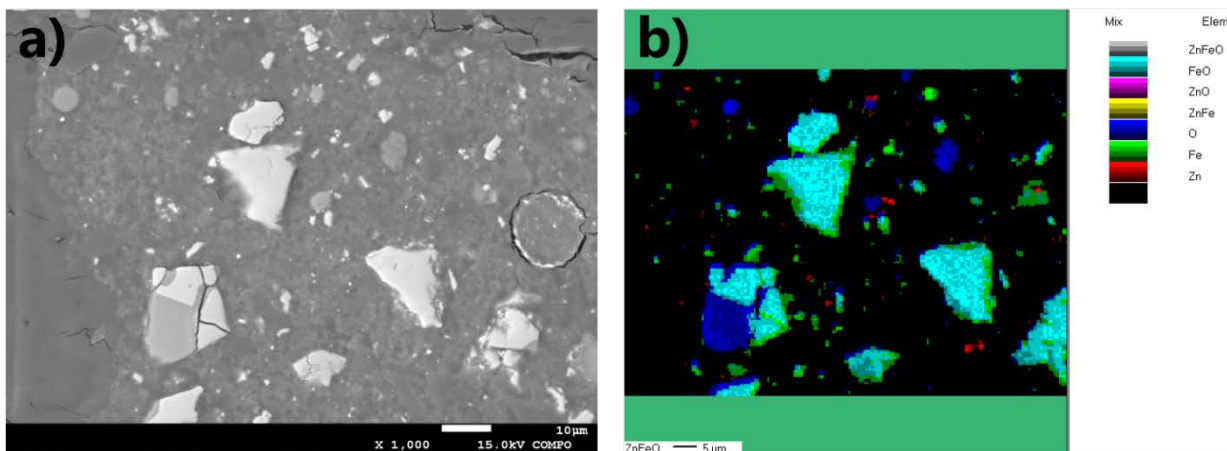
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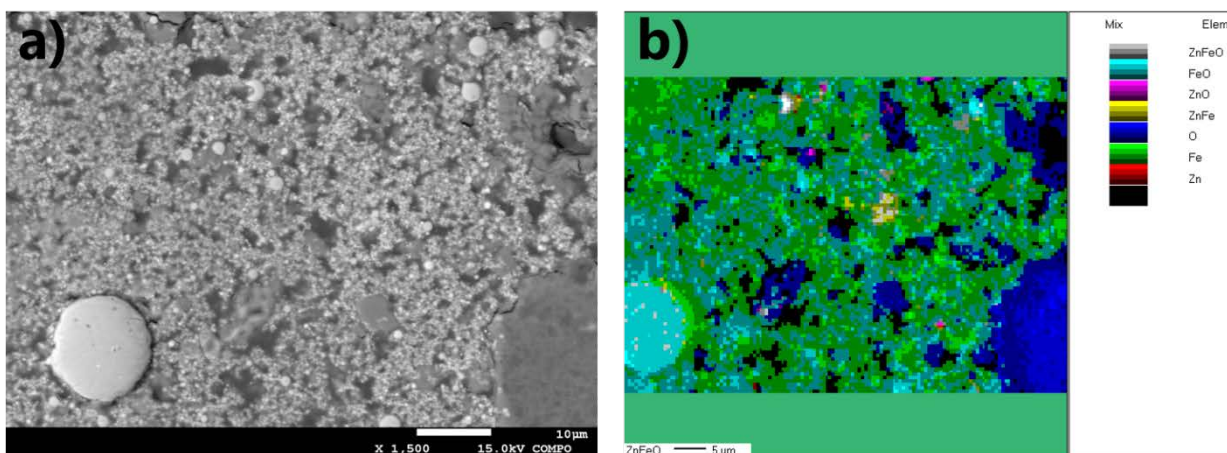
**Fig. S1** Images of the BOF sludge material.



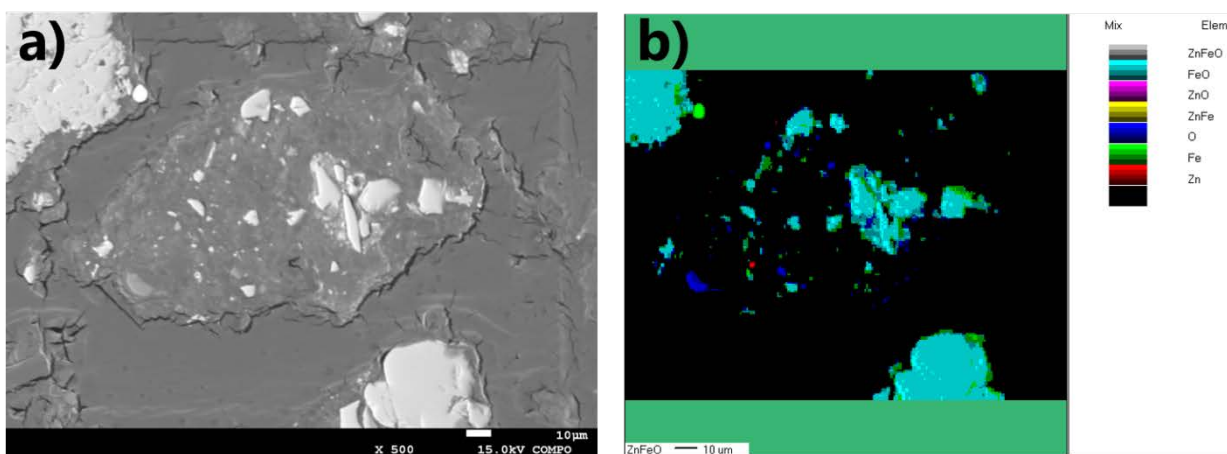
**Fig. S2** EPMA measurements on BOF sample: (a) back-scattered electron (BSE) image and (b) superimposed elemental maps of Zn, O and Fe (scale in color intensity).



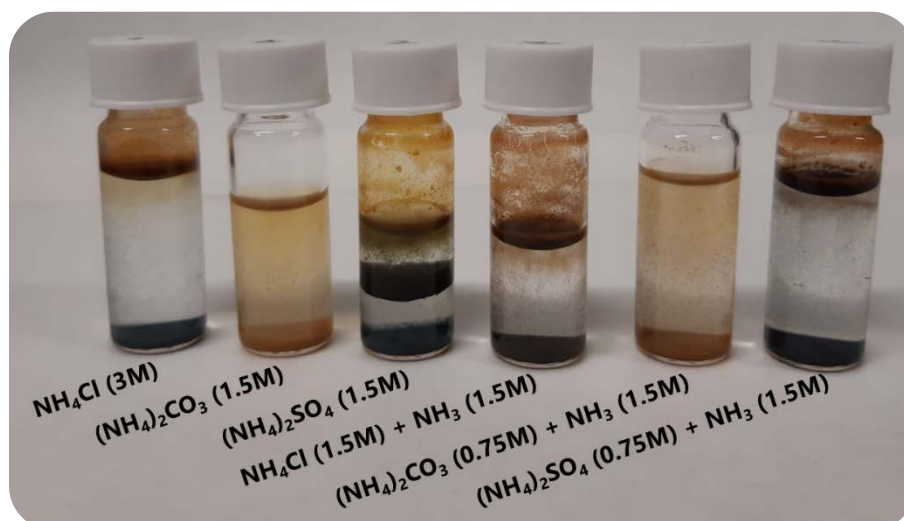
**Fig S3** EPMA measurements on BOF sample: (a) back-scattered electron (BSE) image and (b) superimposed elemental maps of Zn, O and Fe (scale in color intensity).



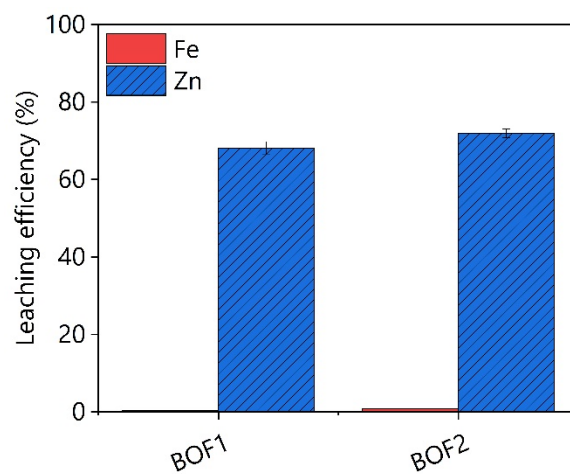
**Fig S4** EPMA measurements on BOF sample: (a) back-scattered electron (BSE) image and (b) superimposed elemental maps of Zn, O and Fe (scale in color intensity)



**Fig S5** EPMA measurements on BOF sample: (a) back-scattered electron (BSE) image and (b) superimposed elemental maps of Zn, O and Fe (scale in color intensity).



**Fig. S6** Filtered BOF pregnant leach solutions after 24 h in a closed vial.



**Fig. S7** Iron and zinc leaching efficiency of the two BOF sludges used in this work under identical leaching conditions:  $T = 60^{\circ}\text{C}$ ;  $t = 3\text{h}$ ;  $L/S = 10$ ;  $[\text{NH}_4^+ + \text{NH}_3] = 3\text{ M}$ ;  $(\text{NH}_4)_2\text{CO}_3:\text{NH}_3 = 1:1$ .

## Optimization of the leaching process

**Table S1** Effect of the liquid-to-solid ratio on the composition of the PLS: Leaching conditions: T = 60 °C; t = 3 h;  $[\text{NH}_4^+ + \text{NH}_3] = 3\text{M}$ ;  $\text{NH}_4^+:\text{NH}_3 = 1:1$

L/S (mL/g)	Fe (ppm)	Zn (ppm)
1	27±7	17972±800
2.5	364±33	11344±239
5	259±8	6640±265
10	433±31	4351±103

**Table S2** Effect of the  $\text{NH}_4^+:\text{NH}_3$  ratio on the composition of the PLS. Leaching conditions: T = 60 °C; t = 3 h; L/S = 10 mL/g;  $[\text{NH}_4^+ + \text{NH}_3] = 3\text{M}$

$\text{NH}_4^+:\text{NH}_3$ ratio	Fe (ppm)	Zn (ppm)
4:1	349±13	2780±108
2:1	264±17	3455±262
1:1	433±31	4351±103
1:2	287±29	4087±44
1:4	793±99	3783±125

**Table S3** Effect of the total ammonia concentration  $[\text{NH}_4^+ + \text{NH}_3]$  on the composition of the PLS. Leaching conditions: T = 60 °C; t = 3 h; L/S = 10 (mL/g);  $\text{NH}_4^+:\text{NH}_3 = 1:2$

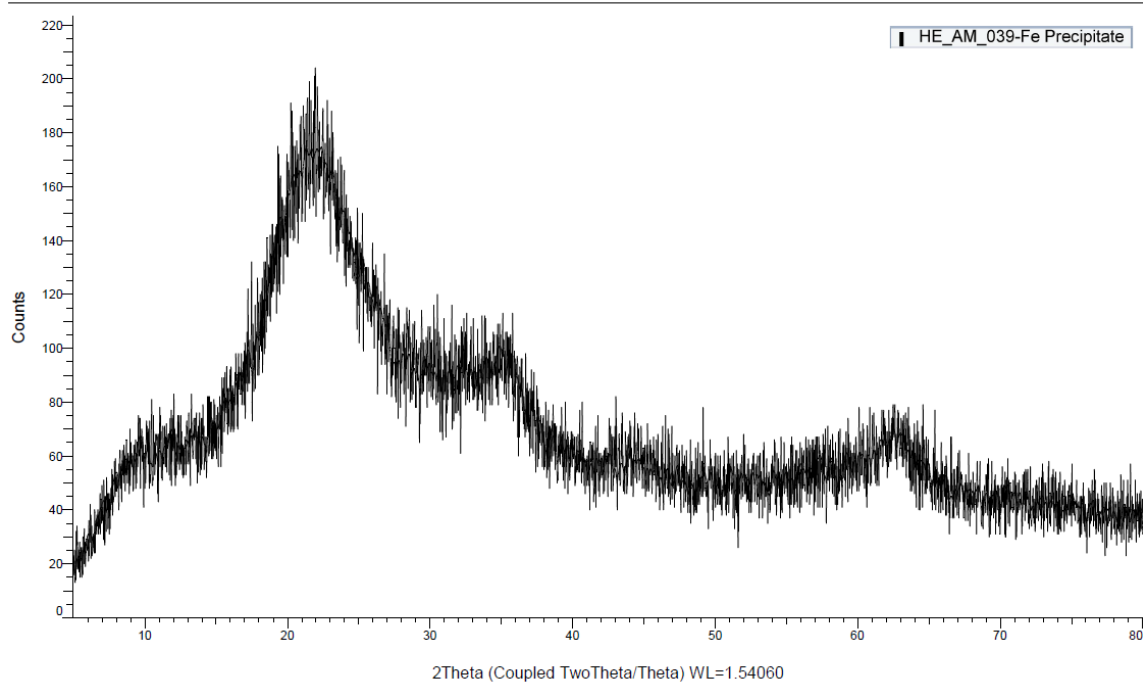
$[\text{NH}_4^+ + \text{NH}_3]$ (mol/L)	Fe (ppm)	Zn (ppm)
2	198±11	3260±75
3	287±29	4087±44
4	1054±48	4415±47
5	2358±242	4549±54
6	3919±259	4773±32



**Fig. S8** Filtration cake of the BOF residue obtained after ammoniacal leaching in the 1 L reactor.



Commander Sample ID (Coupled TwoTheta/Theta)



**Fig. S9** XRD of the precipitate obtained from the iron precipitation step by air addition



**Fig. S10** Precipitate obtained from the iron precipitation step by air addition