Intrinsic five-photon non-linear absorption of two-dimensional BN and its conversion to two-photon absorption in the presence of photo-induced defects

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Abstract: Unlike most two-dimensional materials, h-BN nanoplatelets (BNNPs) exhibit multi-photon absorption. This was previously attributed to a two-photon absorption (2PA) process at 1.16 eV by Kumbhakar *et al.*, *Advanced Optical Materials*, 3, 828, 2015. This is counter-intuitive because a 2PA process at 1.16 eV cannot excite electrons across the wide band gap of BNNPs (~ 5.75 eV). Here, through systematic experimental and theoretical work, we provide compelling evidence that BNNPs exhibit five-photon absorption (5PA) at a low-laser fluence (< 0.6 J/cm²). Furthermore, we show that a high laser fluence (> 0.6 J/cm²) generates photo-induced defects in BNNPs that support a 2PA process. Accordingly, we assert that BNNPs inherently exhibit 5PA at a low laser fluence. In the previous study reported by Kumbhakar *et al.* a high laser fluence was used, which led to the observation of 2PA process.

Keywords: Boron nitride, Two-dimensional materials, non-linear optics, multi-photon absorption, defects, x-ray photoemission spectroscopy

Introduction

Two-dimensional (2D) materials exhibit unique optical properties that are different from their bulk. A testament to this fact is the increasing use of 2D materials in solar cells, light-emitting devices, photodetectors, and ultrafast lasers[1–4]. An interesting feature of light-matter interactions in many 2D materials[5–7] (*e.g.*, graphene, transition metal dichalcogenides, and phosphorene) is their increased transparency at higher light input fluences or the so-called nonlinear saturable absorption (SA)[7–10]. The SA phenomenon is a direct consequence of dynamical interactions between electrons in 2D materials with

the incident laser pulses, which allow for resonant or near-resonant excitation through single-photon absorption[7–10]. In many 2D materials, the excited state absorption cross-section was experimentally found to be lower than that of the ground state absorption[8,11–14]. Consequently, the absorption saturates at an input fluence that is sufficiently high for bleaching the ground state electrons. While this phenomenon is well-known in bulk materials (*e.g.*, Cr:YAG), the threshold SA fluence of 2D materials is significantly higher: a critical requirement for ultrafast laser pulse shaping. Many 2D materials (*e.g.*, graphene and MoS₂) are now being used as SA in mode locking and pulse-shaping in ultrafast lasers[6,13,15,16]. Interestingly, other intriguing nonlinear optical effects such as multi-photon or reverse saturable absorption, often exhibited by other nanomaterials (*e.g.*, C₆₀[7,10], metal-oxide nanoparticles[17–19]) are rarely observed in the nonlinear optical (NLO) properties of undoped 2D materials. Nevertheless, doped 2D materials (*e.g.*, BCN [20]) show multi-photon absorption.

The recent realization of single/few-layer h-BN, which is an isostructural cousin to graphene with a wide band gap of ~ 5.5 -6 eV[21], has opened new vistas for studying multiphoton absorption (MPA) in 2D materials. Previously, Kumbhakar *et al.*[22] reported a strong two photon absorption (2PA) in few-layer h-BN sheets when exposed to nanosecond pulses of a Nd:YAG at 1064 nm (corresponding to a single-photon energy of ~ 1.16 eV). This effect was attributed to the large transition dipole moment resulting from the electronegativity difference between B and N atoms. Furthermore, they reported an anomalously large value for 2PA cross-section at 1064 nm: ~ 52 times larger than the well-known efficient squaraine-bridged porphyrin dimer[23]. From a theoretical standpoint, h-BN should exhibit five-photon absorption (or 5PA) when excited with 1064 nm excitation via the simultaneous absorption of five photons corresponding to an energy ~ 5.80 eV. Given that the band gap of h-BN is ~ 5.5 -6 eV, the conclusions from Kumbhakar *et al.*[22] are peculiar because the energy of two photons from a 1064 nm Nd:YAG laser (two-photon energy ~ 2.32 eV) is insufficient to excite electrons across the band gap. Thus, a fundamental understanding of the nonlinear light-matter interactions in h-BN is still lacking.

Materials exhibiting a 5PA process (either at the bulk or nanoscale) have rarely been investigated in the literature due to the extremely low transition probability associated with a 5PA process[24–26]. Often, the necessary fluence for eliciting 5PA in most materials using nanosecond pulses is well beyond the output of conventional lasers. In this study, we show that pristine h-BN nanoplatelets (BNNPs) inherently exhibit a high transition probability for 5PA when excited with 7 ns pulses of 1064 nm. Thus, BNNPs provide a unique platform for studying exquisite higher-order nonlinearities such as 5PA at low laser fluences (~ 0.2 J/cm²) accessible through a ns pulsed Nd:YAG laser. Moreover, it is well known that MPA at higher fluence can either initiate photo-polymerization at the microscopic scale[27] or induce defects in the material under

study via multiphoton ionization[28]. In case of BNNPs, we observed that a suspension of BNNPs in isopropyl alcohol (IPA) photo-transforms into doped BNNPs with altered configurations of N atoms when excited by a 1064 nm laser at high fluence. While BNNPs inherently exhibited 5PA at a low laser fluence ($< 0.6 \text{ J/cm}^2$), they instantaneously undergo photo-transformation at a high laser fluence leads to 2PA. We used finite-element methods (COMSOL Multiphysics) to uncover that the high laser fluence raises the temperature of BNNPs to $\sim 960 \text{ }^\circ\text{C}$, which consequently promotes doping at the edge sites of BNNPs. In a separate experiment, we prepared O/C doped BNNPs via an alternate route, such as the high-energy ball-milling process, to confirm the influence of laser-induced transformation of the NLO properties in BNNPs, *viz.*, the switching from 5PA to 2PA. Moreover, a clear evidence for the laser-induced doping which promotes the formation of B-O and C-N bonds was elicited through x-ray photoelectron spectroscopy (XPS) of photo-transformed BNNPs. Lastly, the O/C induced electronic states obtained from our density-functional theory (DFT) calculations, provide a rational explanation for the experimentally observed switching in the NLO properties of BNNPs at high laser fluences.

Experimental section

Dispersion of BNNPs: Hexagonal boron nitride powder ($1 \text{ }\mu\text{m}$, 98%) was purchased from Sigma Aldrich. They were exfoliated the using a tip sonicator (Branson, 250 W) with $\sim 3 \text{ mg/ml}$ concentration in N-methyl pyrrolidone (NMP). After exfoliation, the samples were centrifuged at 3000 rpm and the supernatant was collected and washed at least thrice with DI water. The BNNPs for Z-scan experiments ($\sim 1 \text{ mg/ml}$) were dispersed in IPA using a aqueous bath sonicator (Branson Aquasonic) for 5 minutes. Such dispersions were found to be stable for more than an hour, which enabled collection of reliable Z-scans for evaluating the NLO properties of BNNPs

Sample preparation of ball milled BNNPs: For preparing doped BNNPs, two methods were used, *viz.*, ball milling and chemical vapor deposition (CVD). For milling, BNNPs were ball milled with graphite (9:1 weight ratio) in a high-energy ball mill machine (MSK-SFM-3, MTI Corp.) for 2 hours. The CVD process was carried out on a different set of BNNPs in a quartz tube reactor (length 120 cm, and inner diameter 4.5 cm) at atmospheric pressure using CH_4 gas as carbon source. About 50 mg of exfoliated hBN (powder form of hBN kept inside an alumina boat covered with copper foil) was heated to $1000 \text{ }^\circ\text{C}$ for 60 minute in 200 standard cubic centimeters per minute (sccm) Ar/H_2 flow. Next, the CH_4 gas flow is introduced (50 sccm) for 15 minutes in the presence of the Ar/H_2 flow. Subsequently, the CH_4 gas and the CVD furnace were turned off, and the furnace allowed to cool down rapidly (by opening the clam-shell furnace) to room temperature in flowing 200 sccm Ar/H_2 .

Characterization: Raman spectroscopy was performed using a Renishaw in Via Raman microscope equipped with a diode pumped laser source of 532 nm (CrystalLaser, Inc).

High resolution X-ray diffraction (HR-XRD) was performed using a RIGAKU Ultima IV diffractometer, Cu K α radiation, $k \approx 1.5406 \text{ \AA}$ on powder samples that were held in a standard Al sample holder. Quantitative analysis using Rietveld refinement was performed on the XRD peaks using PDXL software. TEM images were obtained using a Hitachi H-9500. While some preliminary XPS studies were performed using a Kratos Axis Ultra DLD instrument (calibrated by the C1s line present at 284.6 eV), more detailed XPS results on CVD-doped BN were obtained using a Kratos Axis Supra XPS employing a monochromated Al K α ($h\nu = 1486.7 \text{ eV}$, 10 mA emission) x-ray source, hybrid (magnetic/electrostatic) optics with a slot aperture, hemispherical analyser, multichannel plate and delay line detector (DLD) with a take-off angle of 90°. High resolution spectra were taken with a pass-energy of 20 eV CVD doped BN was conducting during XPS data acquisition and data were collected with the sample in electrical contact with the spectrometer. Binding energy was referenced to Au4f $_{7/2}$ at 83.96 eV as measured just prior to the sample data acquisition. Quantification was performed using empirical relative sensitivity factors supplied by Kratos Analytical (Manchester, UK).

Z-scan technique: In our Z-scan setup (see **Figure S1**), a linearly polarized 7 ns optical pulsed beam from a Q-switched frequency-doubled 1064 nm Nd:YAG laser was focused by a converging lens (focal length of $\sim 20 \text{ cm}$) to form an optical field of gradually changing laser intensity. A LabVIEW program synchronized the single-shot laser pulses with the moving stage and the resulting repetition rate was $\sim 1 \text{ Hz}$. The linear absorption of BNNP dispersions was quantified using a UV-Vis spectrometer (Perkin Elmer, Lambda) (**Figure S2**), which showed an increased absorption below 400 nm peaked ($\sim 225\text{-}240 \text{ nm}$) concomitant with the band gap of BNNPs $\sim 5.5 \pm 0.5 \text{ eV}$. The linear transmittance (1 mg/ml) at 1064 nm was found to be $\sim 75\%$ at 1064 nm for 1 mg/ml BN dispersions in IPA. The BNNPs/IPA dispersions were held in 1 mm thick quartz cuvettes (100-QS Hellma® Analytics) and translated across the focal plane in the beam direction (Z-direction). With the sample dispersions experiencing different optical intensities at each z position, the corresponding transmittance was recorded by a calibrated photodetector D2 RjP-7620, Laserprobe, Inc.) placed on the sample translation axis (see **Figure S1**). A photodetector was placed off-axis for obtaining nonlinear scattering data. A detailed description of the Z-scan technique can be found elsewhere[7,10,16,29–33].

In our experiments, the open aperture Z-scan MPA coefficients were calculated from spectra under the assumption that only one type of MPA dominates at any given time. The optical intensity I can be described as[21]

$$\frac{dI}{dz'} = -\alpha_N I^{N-1} \quad (1),$$

where z' is propagation distance of light within the sample and α_N is the N-photon absorption coefficient. For example, α_2 represents the 2PA coefficient while α_5 denotes

5PA. The obtained Z-scan data was fitted by numerically solving equation (1) in MATLAB. A fit was deemed good for R^2 (coefficient of determination) values >0.9 .

Results

BNNPs (Sigma Aldrich) were characterized by XRD, Raman spectroscopy, TEM and XPS. The four XRD peaks at $\sim 26^\circ$, $\sim 42^\circ$, $\sim 50^\circ$ and $\sim 55^\circ$ (**Figure 1**) are due to diffraction from the (002), (100), (102) and (004) planes, and the additional diffraction peaks at $\sim 38^\circ$ and $\sim 44^\circ$ stem from the Al sample holder. A lattice constant of 3.29 Å was deduced from the pattern in **Figure 1a** for BNNPs, and its Raman spectrum (**Figure 1b**) showed the characteristic E_{2g} band at 1366 cm^{-1} (FWHM = 11 cm^{-1}). The dynamic light scattering (DLS) measurements (Nanosizer S90, Malvern Instruments) revealed a lateral size of BNNPs as $745 \pm 92\text{ nm}$. HRTEM images of the BNNPs are shown in the **Figures 1c** and **1d**. We found that BNNPs were an average of ~ 20 -25 layers in thickness which exhibited a hexagonal electron diffraction pattern indicating high crystallinity.

We used the open aperture Z-scan method (see experimental section and **Figure S1** for a detailed description) for investigating the NLO properties of BNNPs (suspended in IPA) under varying 1064 nm (1.16 eV) laser fluences. The UV-Vis spectrum of BNNP is shown in **Figure S2**. **Figure 2a** shows a typical normalized transmission (T/T_0 , where T and T_0 are the nonlinear and linear transmittance values) curve for BNNPs when excited with a laser energy of $\sim 220\text{ }\mu\text{J}$ (corresponding to a fluence of $\sim 0.4\text{ J/cm}^2$ at the focal point), which could be best fit to a 5PA process (see solid trace in **Figure 2a** and **Figure S3**). This observation suggests that the electrons at the valence band maximum in BNNPs (band gap of 5.5-6 eV[34], also as observed from **Figure S2**) can be excited to the conduction band through 5PA at input energies $\sim 220\text{ }\mu\text{J}$. It should be mentioned that the possibility of nonlinear light scattering (NLS) leading to the response in **Figure 2a** was ruled out by measuring the off-axis scattered light intensities (see Supplementary Information **Figure S4**). Interestingly, when the laser fluence was increased to $\sim 350\text{ }\mu\text{J}$ (corresponding to a fluence of $\sim 0.6\text{ J/cm}^2$ at the focal point), an abrupt change in the lineshape of BNNPs' normalized transmittance past the focal point ($z = 0\text{ mm}$) became evident (**Figure 2b**). The pre-focus Z-scan data for $350\text{ }\mu\text{J}$ in **Figure 2b**, obtained as the sample approached the focal point (see blue arrow in **Figure S1**), was best fitted to 5PA. However, the post-focus data (receding from the focal point, green arrow in **Figure S1**) showed the presence of two-photon absorption or 2PA.

In reality, BNNPs support 5PA as long as the input fluence is below the threshold value $0.6 \pm 0.012\text{ J/cm}^2$ - an inherent property of BNNPs which was overlooked in the study reported by Kumbhakar *et al* due to the high fluence (3.9 J/cm^2 , 1064 nm) used in their study. Furthermore, BNNPs are also expected to have mid-gap states within the band gap due to inherent defects that could provide real electronic states to support simultaneous 5PA in addition to spontaneous 5PA. However, we found that the 5PA co-efficient was

the same at different intensities suggesting the inherent 5PA is virtual with simultaneous absorption. Given that the laser beam radius at the focal point was $\sim 100\ \mu\text{m}$, only a small amount of BNNPs within the beam path is expected to be photo-transformed. To confirm this, we repeated the Z-scan on the same BNNPs dispersion after gently stirring it. As expected, the inherent 5PA response of BNNPs was recovered (**Figure S5**) implying that photo-transformed BNNPs were no longer present in the beam path. **Also, no discernable changes were observed in the linear absorption spectrum before and after laser exposure.** Furthermore, as mentioned earlier, we ruled out NLS effects arising from micro-bubbles at high laser fluence by using an off-axis detector (see **Figures S1** and **S4**). Based on the data presented in **Figure 2**, we conclude that the much higher laser fluence used in the previous study by Kumbhakar *et al*[22] ($3.9\ \text{J}/\text{cm}^2$, $1064\ \text{nm}$) precluded the authors from uncovering the inherent 5PA in h-BN.

Discussion

We hypothesize that the high laser fluence promotes heteroatomic doping of oxygen/carbon (from IPA) into BNNPs, possibly at the edge sites due to its low formation energy[35,36]. The presence of O/C dopants in BN lattice is known to introduce new electronic states within the band gap[37–41], which could explain the observation of 2PA at high laser fluences[20]. Indeed, recent spectroscopic studies (cathodo- and photo-luminescence, UV absorption, and scanning tunneling spectroscopy) on defected/doped h-BN crystals and nanostructures found clear spectral features at lower energies ($\sim 2\text{--}4\ \text{eV}$) in addition to the band gap $\sim 5.75\ \text{eV}$ due to the presence of dopant-induced electronic states[41,42] within the band gap. Similarly, our theoretical studies using density functional theory (see **Figure S6**) along with earlier investigations employing local density approximation and GW approximation confirmed the presence mid-gap states for C/O dopants at $\sim 1\text{--}4\ \text{eV}$. Based on our Z-scan results and existing spectroscopic studies[7,10,43,44], we posit that the ground state electrons in doped BNNPs when excited with $1064\ \text{nm}$ (or $\sim 1.16\ \text{eV}$) photons are promoted to the C/O-dopant induced states present $\sim 2\text{--}2.5\ \text{eV}$ through a 2PA process (inset in **Figure 2b**).

We used finite-element method (COMSOL Multiphysics) with a heat transfer module to simulate the conditions at high laser fluence that could result in doping of O/C into BNNPs. The BNNPs were modeled as platelets with average dimensions $100 \times 100 \times 10\ \text{nm}^3$ (as gleaned from TEM images such as the one presented in **Figure 1c**) and the laser fluence was matched to threshold values for eliciting a 2PA response, *i.e.*, $\sim 7\ \text{ns}$ pulse with an energy density of $\sim 0.6\ \text{J}/\text{cm}^2$ (*cf.* **Figure 2b**). The temperature distribution of the BNNP platelets after exposure to a single laser pulse is shown in **Figure S7**. Our COMSOL simulation revealed that the laser rapidly ($\sim 7\ \text{ns}$) heats the surface of the BNNP and increases its temperature to $950\text{--}966\ ^\circ\text{C}$. The top surface edges of the BNNP were raised to a higher temperature relative to those on the bottom layer because the heat diffusion rate is much faster in the lateral ($6\ \text{W cm}^{-1}\ ^\circ\text{C}^{-1}$) than in the vertical ($0.3\ \text{W}$

$\text{cm}^{-1} \text{ } ^\circ\text{C}^{-1}$) direction for h-BN[45]. The laser-induced increase in the temperature at the edges of BNNP platelets is much higher than the decomposition temperature of IPA ($\sim 400 \text{ } ^\circ\text{C}$), which possibly leads to IPA carbonization and subsequent doping of BNNPs with carbon at its edge sites.

While conventional XPS is an excellent tool for confirming the presence of dopants in BNNPs, it is plagued by the presence of O/C peak of the IPA solvent used in this study. Moreover, only a small amount of BNNPs that are present within the beam path are photo-transformed. Thus, to confirm our hypothesis that BNNPs switch their NLO properties from 5PA to 2PA due to laser-induced doping of its edges with O/C, we intentionally doped BNNPs with O/C by two different methods: 1) CVD method with temperatures $\sim 1000 \text{ } ^\circ\text{C}$ as observed in the COMSOL simulation and 2) ball milling a mixture of BNNPs and graphite powder.

We performed detailed XPS studies on both CVD doped and ball-milled samples to ascertain different bonding environments (B-N, B-C, C-N, N-O, B-O etc) that could induce mid-gap states responsible for 2PA at higher fluence. In the case of CVD-doped BN (labeled CBN), its B 1s spectrum (**Figure 3a**) exhibits at least four different environments. The lowest binding energy at 190.5 eV is assigned to B bonded to three N atoms. The lack of lower binding energy components in any of the samples is indicative of a lack of boron bonded to carbon and nitrogen alone. The higher binding energy components at 191.0 eV, 191.5 eV and 192 eV are tentatively ascribed to BCNO/ BN_2O / BC_2O , BNO_2 and B(OH)_3 based on previously published XPS data for boron nitride systems.[46–48] The overall shape of the B 1s peak in all of these samples lacks structure and appears slightly asymmetric. Similar to the B 1s peak, the N 1s spectrum for CBN exhibited four different environments (**Figure 3b**). The first environment at 398 eV is assigned to nitrogen bonded to boron. The other environments at 398.5 eV, 399 eV and 400 eV could be assigned to various bonding modes of carbon with nitrogen, referred to as pyridinic and pyrrolic and graphitic respectively. [38,49] These assignments are at slightly lower binding energies compared with literature values for nitrogen doped graphenes, which may be indicative of boron containing environments.[50,51] They are consistent with nitrogen environments reported for other carbon containing boron nitride systems.[48] Overall, our analysis showed that CBN sample had $\sim 35 \text{ at. } \%$ carbon (see supplementary Table 1) suggesting that the sample should exhibit 2PA. Accordingly, we observed that CBN displayed 2PA even at low fluence $\sim 0.2\text{-}0.6 \text{ J/cm}^2$ (**Figure S8**).

We also performed similar studies with ball-milled BN. For ball milling, we mixed BNNPs and graphite in 9:1 weight ratio, which we refer to as “BNNP MIX”. The Z-scan curves for BNNP MIX sample (discussed later in **Figure 4c**), prior to ball milling, showed a clear 5PA signal intrinsic to BNNPs at a low laser fluence $100 \text{ } \mu\text{J}$ ($\sim 0.2 \text{ J/cm}^2$ at

the focus). Next, the BNNP MIX sample was ball milled for 2 hours for doping O/C into the BN lattice. An extensive characterization toolset including XRD, Raman spectroscopy (**Figure S9**) and XPS (**Figure 4**) confirmed doping of both O and C atoms (~1-2 at. %, much lower compared to CBN, as evinced from XPS in **Figures 4a** and **4b**) in ball milled BNNPs[38,40,49], which we henceforth refer to as BNNP BM sample. As shown in **Figures 4a** and **4b**, the B 1s and N 1s spectra from pristine and doped BNNPs show primary binding energies of 190.7 and 398.3 eV, respectively. We observed an extra peak ~192.8 eV in the B 1s spectrum (**Figure 4a**), which is known to arise from B-O bonds, confirming O-doping in BNNP BM samples. Furthermore, we also found that the N 1s spectrum was slightly downshifted from ~398.3 to 398.1 eV, which was previously attributed to C-N bonds. [38,49] As shown in **Figure 4c**, we observed a 2PA response in the Z-scan curve for BNNP BM sample in IPA even at a low laser energy of 100 μJ (~0.2 J/cm^2 at the focus) while BNNP MIX samples (prior to ball milling) exhibited the intrinsic 5PA response. The values of 2PA coefficients for BNNP BM and photo-transformed BNNPs in IPA were found to be very similar (**Table S2**). This implies that the mid-gap states supporting a 2PA response in BNNP BM and photo-transformed BNNPs are similar *viz.*, O/C dopant-induced electronic states present ~2-2.5 eV within the band gap of BN (*cf.* inset of **Figure 2b**).

Conclusions

In summary, we report the discovery of a rarely observed nonlinear phenomenon, *i.e.*, five-photon absorption (5PA) in 2D h-BN nanoplatelets (BNNPs). Through a systematic study we show that the nonlinear optical response of BNNPs dispersed in isopropyl alcohol (IPA) switches from a 5PA to a 2PA response when the laser fluence exceeds 0.6 J/cm^2 (at the focus or $z=0$) due to photoinduced O/C doping (from IPA) of BNNPs. Previous studies on the nonlinear optical properties of BNNPs (Kumbhakar *et al.* [22]) used very high laser fluence (~3.9 J/cm^2) and thus failed to uncover the inherent 5PA phenomenon in BNNPs. Our finite-element calculations using COMSOL Multiphysics showed a rapid increase in the temperature of edge sites in BNNPs (> 960 °C) suitable for carbonizing IPA and doping of O/C atoms at edge sites in BNNPs. We associate the observation of a 2PA response at a high laser fluence in BNNPs with to the NLO properties of O/C doped BNNPs. This correlation was independently confirmed from Z-scan spectra of BNNPs doped with O/C atoms using two-different methods: 1) chemical vapor deposition and 2) ball milling with graphite. While XPS presented clear evidence of heteroatomic bonding in both the CVD and in the ball milled BNNPs, Z-scan spectra at a low laser fluence showed a 2PA response, akin to the Z-scan spectra of photo-transformed BNNPs. Lastly, the 2PA coefficients in ball milled BNNPs and photo-transformed BNNPs were found to be similar, further confirming our hypothesis.

Supporting Information

Supporting Information is available from online or from the author.

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Figures and Figure Captions

Figure 1. a) XRD pattern of BNNPs held on an Al sample holder. b) Raman spectrum of pristine BNNPs using the 532 nm laser excitation. HRTEM images of BNNPs c) side view and d) top view. The inset of d) is the selected area electron diffraction of the

BNNP.

Figure 2. a) A dispersion of pristine BNNPs in IPA exhibits a 5PA response in a Z-scan curve collected with a laser fluence of 220 μJ . b) Upon increasing the laser fluence to 350 μJ , a 5PA response is observed, signaling an onset of photo-induced doping of BNNPs with carbon from the IPA. The coefficients for 2PA and 5PA found from the numerical fits ($R^2 > 0.9$) were 12 cm GW^{-1} and $2500 \text{ cm}^7 \text{ GW}^{-4}$, respectively. A schematic inset showing the O/C dopant levels introduced into the wide band gap of h-BN. The presence of such levels at $\sim 2\text{-}2.5 \text{ eV}$ enable 2PA in the photo-transformed BNNPs.

Figure 3. a) and b) show B 1s and N 1s XPS spectra for CVD doped BN. In both the cases, different environments were found as shown by the deconvoluted peaks. A complete list of peak positions and intensities is provided in Table S1.

Figure 4. a) and b) show the XPS spectra of pristine and ball milled BNNPs. A new peak is observed $\sim 193 \text{ eV}$ in the B 1s spectrum, which is indicative of B-O bonds or the incorporation O atoms into BN lattice. The N 1s spectrum was found to broaden and downshift by $\sim 0.2 \text{ eV}$ possibly due to C doping. c) Z-scan curves for BNNP BM dispersion in IPA, which exhibits a 2PA response (red circles). The mixture of BNNPs and graphite powder, prior to ball milling, exhibited a 5PA response as expected (blue circles).

Figure 1

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Figure 2

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Figure 3

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