

1 **Pectin influences the kinetics of *in vitro* lipid**  
2 **digestion in oil-in-water emulsions**

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11 **Running title:**

12 **Effect of pectin on lipolysis kinetics**

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## 33 **Abstract**

34 Oil-in-water emulsions were prepared with 5% (w/v) carrot-enriched olive oil and stabilized  
35 with Tween 80 (TW), phosphatidylcholine (PC), citrus pectin (CP) or a combination of these  
36 emulsifiers. Additionally, the methylesterification degree (DM) of citrus pectin was modified,  
37 resulting in three different studied pectin structures: CP82, CP38 and CP10. All initial  
38 emulsions presented small initial oil droplet sizes and were submitted to an *in vitro* simulated  
39 gastric and small intestinal phase. The latter was executed in a kinetic way to determine the  
40 time dependency of the lipolysis reaction, micelle formation and carotenoid bioaccessibility.  
41 The results showed that the pectin DM mainly influenced the reaction rate constants, while  
42 the emulsifier (combination) determined the extent of lipolysis and carotenoid  
43 bioaccessibility. Moreover, a direct relation was observed between the lipolysis reaction and  
44 bioaccessibility extent. The presented study showed that targeted emulsion design can be used  
45 to tailor lipid digestion kinetics.

46  
47 **Keywords:** citrus pectin; emulsion; *in vitro* digestion; lipolysis; carotenoids.

48  
49 **Chemical compounds:** Tween 80 (PubChem CID: 5281955); Phosphatidylcholine  
50 (PubChem CID: 16219824); Lipase (PubChem CID: 54603431); Bile salt (PubChem CID:  
51 439520); Calcium (PubChem CID: 5460341);  $\alpha$ -carotene (PubChem CID: 4369188);  $\beta$ -  
52 carotene (PubChem CID: 5280489).

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54

55 **Abbreviation list:**

56 BAC: bioaccessibility

57 CP: citrus pectin

58 CP82; CP38 or CP10: citrus pectin with a methylesterification degree of 82%, 38% and 10%,  
59 respectively

60 CP82 emulsion: 5% (w/v) olive oil-in-water emulsion stabilized with 1% (w/v) citrus pectin  
61 with a methylesterification degree of 82%

62 CP38 emulsion: 5% (w/v) olive oil-in-water emulsion stabilized with 1% (w/v) citrus pectin  
63 with a methylesterification degree of 38%

64 CP10 emulsion: 5% (w/v) olive oil-in-water emulsion stabilized with 1% (w/v) citrus pectin  
65 with a methylesterification degree of 10%

66 DAG: diacylglycerol

67 DM: methylesterification degree

68 FFA: free fatty acids

69 GLY: glycerol

70 HMP: high methoxylated pectin

71 JCR: joint confidence region

72 LMP: low methoxylated pectin

73 MAG: monoacylglycerol

74 MMP: medium methoxylated pectin

75 o/w emulsion: oil-in-water emulsion

76 PC: Phosphatidylcholine

77 PC emulsion: 5% (w/v) olive oil-in-water emulsion stabilized with 1% (w/v)  
78 phosphatidylcholine

79 PCCP82 emulsion: 5% (w/v) olive oil-in-water emulsion stabilized with 1% (w/v)  
80 phosphatidylcholine and 1% (w/v) citrus pectin with a methylesterification degree of 82%

81 PCCP38 emulsion: 5% (w/v) olive oil-in-water emulsion stabilized with 1% (w/v)  
82 phosphatidylcholine and 1% (w/v) citrus pectin with a methylesterification degree of 38%

83 PCCP10 emulsion: 5% (w/v) olive oil-in-water emulsion stabilized with 1% (w/v)  
84 phosphatidylcholine and 1% (w/v) citrus pectin with a methylesterification degree of 10%

85 TAG: triacylglycerol

86 TW: Tween 80

87 TW emulsion: 5% (w/v) olive oil-in-water emulsion stabilized with 0.5% (w/v) Tween 80

88 TWCP82 emulsion: 5% (w/v) olive oil-in-water emulsion stabilized with 0.5% (w/v) Tween  
89 80 and 1% (w/v) citrus pectin with a methylesterification degree of 82%

90 TWCP38 emulsion: 5% (w/v) olive oil-in-water emulsion stabilized with 0.5% (w/v) Tween  
91 80 and 1% (w/v) citrus pectin with a methylesterification degree of 38%

92 TWCP10 emulsion: 5% (w/v) olive oil-in-water emulsion stabilized with 0.5% (w/v) Tween  
93 80 and 1% (w/v) citrus pectin with a methylesterification degree of 10%

94

## 95 **1 Introduction**

96 Lipids are the macronutrients with the highest energy density, which deliver essential fatty  
97 acids and can be carriers of lipophilic micronutrients, such as carotenoids (Golding &  
98 Wooster, 2010). In the human diet, lipids are frequently consumed as oil-in-water (o/w)  
99 emulsions, such as soups and sauces, being natural sources of fibers and micronutrients. Lipid  
100 digestion predominantly occurs in the small intestine, where triacylglycerol (TAG) can be  
101 hydrolyzed by pancreatic lipase into diacylglycerol (DAG), monoacylglycerol (MAG), free  
102 fatty acids (FFA) and glycerol (GLY). Subsequently, these lipid digestion products form  
103 mixed micelles together with bile salts excreted from the liver. These mixed micelles are  
104 amphiphilic structures which can incorporate lipophilic components in the core and easily  
105 migrate in the aqueous intestinal environment towards the intestinal mucosa (McClements &  
106 Decker, 2009). The fraction of ingested lipids and lipophilic components which are  
107 micellarized and consequently are available for absorption into the blood, is called the  
108 'bioaccessible' fraction.

109 Recently, digestion studies were more focusing on controlling the lipid digestion rate as it  
110 plays an important role in satiety control, which in turn is related to food intake and diseases  
111 of nutritional excess (Ohlsson et al., 2014). Previous research has shown that dietary fibers,  
112 such as chitosan, cellulose, guar gum and pectin, may affect lipid digestion based on analysis  
113 of FFA release (Pasquier et al., 1996; Mun et al., 2006; Hur et al., 2013; Espinal-Ruiz et al.,  
114 2014a). It can be postulated that by increasing the intake of natural fibers, the lipid digestion  
115 mechanism can be modified and subsequently lipid intake can be controlled. In addition, lipid  
116 digestion is strongly related to the uptake of lipophilic carotenoids (Borel et al., 1996; Deming  
117 & Erdman, 1999), so modulating the lipolysis extent might also have influence carotenoid  
118 uptake. Therefore, the effect of the presence of fibers in food systems must be investigated on  
119 the kinetics of both lipid digestion as well as carotenoid bioaccessibility.

120 Pectin is one of the most abundant fibers present in the primary cell wall and middle lamella  
121 of all higher plants (Willats et al., 2001). It is a diverse group of polysaccharides rich in  
122 galacturonic acid (GalA) units which recently has shown emulsifying and emulsion-  
123 stabilizing potential (Schmidt, Schwab, & Schuchmann, 2017). Since pectin is a natural  
124 ingredient, it can be interesting to explore its role as emulsifier or when it is present in the  
125 aqueous phase of o/w emulsions on the lipolysis kinetics. It is known that the properties of the  
126 oil droplet, oil-water interface and surrounding medium can have a major effect on the  
127 lipolysis kinetics. In this context, it was proven that initial small oil droplet sizes led to a  
128 faster and higher lipolysis than larger initial oil droplets (Salvia-Trujillo et al., 2017). Not only  
129 initial small oil droplets are of importance, but also the stability of these oil droplets along the  
130 simulated digestive tract, which can be influenced by the emulsifier type used for emulsion  
131 stabilization. The emulsion stability during digestion will determine the available surface area  
132 for lipase adsorption in the small intestinal phase and the subsequent lipolysis kinetics  
133 (Verkempinck et al., 2018b). However, little is known about the specific influence of the  
134 presence of pectin in o/w emulsions on the kinetics of lipid digestion and carotenoid  
135 bioaccessibility. In this sense, pectin can be added in emulsions, being present at the oil-water  
136 interface or pectin can be located in the aqueous phase of these o/w emulsions. In addition,  
137 the structural characteristics of these pectin structures might play an important role during  
138 digestion. For example, the methylesterification degree of pectin influences its charge density  
139 (Celus et al., 2018) and interactions with different components during digestion (such as  
140 calcium, lipase, bile salts and lipophilic components) (Tsujita et al., 2007; McClements,  
141 Decker, & Park, 2008; Verrijssen et al., 2014; Espinal-Ruiz et al., 2014b). Therefore, the  
142 presented work, explored the effect of the presence of citrus pectin with different  
143 methylesterification degree in an emulsified system on the lipid digestion, using the kinetic  
144 digestion approach presented by Verkempinck et al. (2018a). Moreover, not only the effect of

145 citrus pectin as mono-emulsifier was evaluated, but also the combination of citrus pectin and a  
146 conventional emulsifier stabilizing o/w emulsions was studied. Consequently, the link  
147 between lipid digestion and carotenoid bioaccessibility can be quantitatively proven. This  
148 allows to evaluate the proposed hypothesis throughout the whole digestion process based on  
149 the resulting kinetic parameters. So far in this research field, lipid digestion is mostly studied  
150 by evaluation of FFA release in the small intestinal phase. By contrast, this study aimed to  
151 quantify multiple lipid digestion species, namely TAG, DAG, MAG and FFA, both in the  
152 digest as well as in the micellar fraction. The obtained results can eventually contribute to the  
153 development of predictive mathematical (*in silico*) models needed for simulation of lipid  
154 digestion of particular food products and to create products for specific consumer groups.

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## 155 **2 Material and methods**

156 Carotenoid enriched oil-in-water (o/w) emulsions were formulated with different emulsifiers  
157 (Tween 80, phosphatidylcholine or citrus pectin) or combinations of emulsifiers (section 2.4).  
158 Emulsions stabilized with one emulsifier type will be called ‘mono-emulsifier emulsions’,  
159 while emulsions stabilized with pectin and another emulsifier will be called ‘di-emulsifier  
160 emulsions’. More specifically, the effect of pectin present at the oil-water interface or in the  
161 aqueous phase of o/w emulsions on the kinetics of lipid digestion and carotenoid  
162 bioaccessibility was studied. Therefore, all emulsions were *in vitro* digested by simulating a  
163 gastric and small intestinal phase (section 2.5). The latter was performed using a kinetic  
164 approach to evaluate the time dependency of the lipolysis reaction and micellarization of  
165 multiple lipolysis products and carotenoids. Digested emulsions will be further referred to as  
166 ‘digest’. All digests were ultracentrifuged to harvest the aqueous, ‘micellar fraction’  
167 containing the bioaccessible lipid species and carotenoids. Initial emulsions, digests and  
168 micellar fractions were characterized in terms of lipid and carotenoid content (section 2.7 and  
169 2.8). Additionally, the particle charge and size of the initial emulsion, after the gastric phase  
170 (chyme) and after 2 hours of small intestinal phase (digestion end point) were determined to  
171 evaluate the behavior of the oil droplets during gastrointestinal conditions (section 2.6).

### 172 **2.1 Materials**

173 Orange carrots (*Daucus carota* cv. Nerac) were bought in a local shop and stored at 4 °C until  
174 use. Olive oil was purchased in a local shop. Citrus pectin (CP), Tween 80 (TW) and  
175 phosphatidylcholine (PC) were obtained from Sigma Aldrich (Diegem, Belgium). All used  
176 chemicals and reagents were of analytical or HPLC-grade and were purchased from Sigma  
177 Aldrich (Diegem, Belgium) except for KCl, MgCl<sub>2</sub>(H<sub>2</sub>O)<sub>6</sub>, NaOH, heptane, methanol, methyl-  
178 tert-butyl-ether and ethyl acetate (Acros Organics, Geel, Belgium); KH<sub>2</sub>PO<sub>4</sub>; NaHCO<sub>3</sub>, NaCl,

179 H<sub>2</sub>SO<sub>4</sub>, ethanol, acetone and trimethylamine (Fisher Scientific, Merelbeke, Belgium); HCl,  
180 diethylether and iso-propanol (VWR, Leuven, Belgium); acetone (Carlo Erba, Val-de-Reuil,  
181 France); CaCl<sub>2</sub>(H<sub>2</sub>O)<sub>2</sub> (Chem-Lab, Zedelgem, Belgium) and lipid standards (Larodan, Solna,  
182 Sweden). Sample preparations were performed with reagent water (organic free, 18.2 MΩ cm  
183 resistance), supplied by a Simplicity™ 150 water purification system (Millipore, Billerica,  
184 USA).

## 185 **2.2 Pectin preparation**

186 High methylesterified CP (HMP) was enzymatically demethylesterified according to the  
187 procedure described by Ngouémazong et al. (2011). It was opted use citrus pectin as study  
188 vehicle since it is commercially available and has a linear structure, facilitating the  
189 interpretation of the obtained data and attributing that to differences in degree of  
190 methylesterification only.

191 Briefly, 0.8% (w/v) HMP was incubated with purified extract of carrot pectin methylesterase  
192 (PME) at 30 °C for different time periods, resulting in CP with a medium and low degree of  
193 methylesterification (DM) (MMP and LMP, respectively). A thermal treatment was used to  
194 inactivate PME (4 min, 85 °C) and was followed by a 48-hour dialysis to remove present ions  
195 (Spectra/Por®, Molecular weight cut-off = 12-14 kDa). In a final step, pectin samples were  
196 lyophilized (Christ Alpha 2-4 LSC, Germany) and stored in a desiccator at room temperature  
197 until use. Fourier transform infra-red spectroscopy (IRAffinity-1, Shimadzu, Japan) was used  
198 to determine the pectin DM, according to the method described by Kyomugasho et al. (2015).  
199 The resulting values for the DM of HMP, MMP and LMP were 82.2 % (± 1.2), 38.3 % (± 0.9)  
200 and 10.4 % (± 1.0), respectively and these pectin structures will be further indicated as CP82,  
201 CP38 and CP10. In addition, the molecular weight was determined, following the procedure  
202 of Shpigelman et al. (2014) with High Performance Size Exclusion Chromatography. For

203 CP82, CP38 and CP10, a molecular weight was obtained of 58.7 kDa ( $\pm 1.3$ ), 54.5 kDa ( $\pm$   
204 2.0) and 54.1 kDa ( $\pm 1.2$ ), respectively, showing that the enzymatic demethylesterification  
205 reaction had no significant impact on the molecular weight of the pectin structures. The  
206 concentration distribution as function of elution time and the average molecular weights of the  
207 different pectin samples is shown in **Figure A in supplementary material**.

### 208 **2.3 Preparation of carrot-enriched olive oil**

209 Olive oil was enriched with carotenoids from carrot puree according to the method described  
210 by Mutsokoti et al. (2015). Carrots were peeled, cut into small pieces (1 x 1 cm) and mixed  
211 with demineralized water (1:1) for 1 minute with a kitchen blender (Warington Commercial,  
212 Torrington, CT, USA). Carotenoids were released by disrupting their cellular entrapment  
213 through a high pressure homogenization step at 100 MPa (Pony NS2006L, Gea Niro Soavi,  
214 Düsseldorf, Germany). Olive oil was added (1:5 w/w) and subsequently a blending (1 minute)  
215 and a homogenization step (100 MPa, 1 cycle) were performed to allow carotenoid migration  
216 from the natural plant matrix into the oil phase. After centrifugation (Optima XPN-80  
217 Ultracentrifuge, Beckman Coulter, Fullerton, CA, USA) with a speed of 165000 g at 4 °C, the  
218 carotenoid-enriched olive oil was collected and stored at -80 °C, protected from light and  
219 oxygen. The concentration of  $\alpha$ - and  $\beta$ -carotene in the obtained enriched oil were  $86.3 \pm 6.7$   
220 and  $209.2 \pm 16.1$   $\mu\text{g}$  per g of enriched oil, respectively. It was explicitly chosen to incorporate  
221 carotenoids from a food matrix into the olive oil, rather than using pure carotenoids since in  
222 this way a real food system could be approached whereby carotenoids might already be  
223 transferred to the oil phase during food preparation prior to digestion.

### 224 **2.4 Oil-in-water emulsion preparation**

225 First, coarse o/w emulsions were prepared by mixing 5% carrot-enriched oil (w/v) for 10  
226 minutes at 9500 rpm (Ultra-Turrax T25, IKA, Staufen, Germany) to an emulsifier-containing

227 aqueous phase. The aqueous phase consisted of a mono-emulsifier solution of 0.5% w/v  
228 Tween 80 (TW), 1% w/v CP or 1% w/v PC or a di-emulsifier solution of 1% w/v CP mixed  
229 with 0.5% w/v TW or 1% w/v PC. A minimum emulsifier amount was used to create stable  
230 emulsions. The coarse emulsion was kinetically stabilized with high pressure homogenization  
231 at 100 MPa for 1 cycle (Pressure Cell Homogenizer, Stansted Fluid Power LTD., UK).  
232 Hereafter, the acidity of all emulsions was adjusted to pH 3 as a preliminar study showed that  
233 the emulsions were more stable at low pH (data not shown). All emulsions were stored at 4  
234 °C, protected from light and oxygen for maximally 5 days, being the period in which the  
235 emulsions were stable in terms of particle size (data not shown). 'TWCP82' is the  
236 abbreviation used to refer to an emulsion stabilized with 0.5% Tween 80 and 1% citrus pectin  
237 with a DM of 82. In total, 11 different emulsions were prepared and studied, being referred to  
238 as TW; CP82; CP38; CP10; TWCP82; TWCP38; TWCP10; PC; PCCP82; PCCP38 and  
239 PCCP10. TW; CP82; CP38; CP10; PC are the 'mono-emulsifier emulsions', while the  
240 TWCP82; TWCP38; TWCP10; PCCP82; PCCP38 and PCCP10 emulsions will be referred to  
241 as 'di-emulsifier emulsions'.

## 242 **2.5 *In vitro* digestion**

243 A static *in vitro* digestion procedure was used to investigate the isolated effect of time on  
244 lipolysis, micelle formation and carotenoid bioaccessibility. Therefore, the consensus method  
245 described by Minekus et al. (2014) as adjusted by Verkempinck et al. (2018a) was applied.  
246 Briefly, the gastric phase was simulated by mixing 5 mL of o/w emulsion and 5 mL of reagent  
247 water in dark falcon tubes. Subsequently, 7.5 mL simulated gastric fluid, 5 µL  
248 calciumchloride (0.3M); 1.6 mL pepsin solution (2000 U/mL in final chyme); 889 µL reagent  
249 water and 6 µL HCl (2M) (to adjust the pH to 3) were added. The tubes headspace was  
250 flushed with nitrogen, whereafter the chyme was rotated end-over-end (40 rpm) for 2 hours at  
251 37 °C. Hereafter, the small intestinal phase was mimicked by adding 11 mL simulated

252 intestinal fluid, 40  $\mu$ l calciumchloride (0.3M); 1.46 mL demineralized water; 2.5 mL of bile  
253 solution (10 mM in final digest) and 5 mL of pancreatic solution to 10 mL of chyme. All  
254 intestinal fluids were tempered at 37 °C to avoid temperature differences in the digest. The  
255 pancreatic solution contained both pancreatin (100 U/mL based on trypsin activity in final  
256 digest) and porcine pancreas lipase (to obtain 200 U/mL in the final digest). Furthermore,  
257 carotenoid isomerization and degradation during the gastrointestinal conditions was  
258 minimized by addition of pyrogallol (0.6% w/v) and  $\alpha$ -tocopherol (1.4% w/v) (Lemmens et  
259 al., 2011). Nitrogen gas was used to flush the headspaces of the tubes and these tubes were  
260 subsequently rotated end-over-end (40 rpm) at 37 °C. The digestion of each emulsion was  
261 characterized for seven consecutive time moments (0-120 minutes) to study the time effect on  
262 the lipolysis reaction and the micellarization of multiple lipid digestion products and  
263 carotenoids. For each time moment, an individual digestion tube was used, resulting in seven  
264 independent replications of the simulated digestion process. To inactivate lipase at a particular  
265 digestion time in the *in vitro* small intestinal phase, the digest was subjected to a thermal  
266 treatment (10 min, 85 °C) and hereafter immediately cooled to 4 °C. The micellar fraction  
267 was separated from other components (e.g. enzyme residues and undigested oil) after  
268 ultracentrifugation at 165000 g for 68 minutes at 4 °C (Optima XPN-80 Ultracentrifuge,  
269 Beckman Coulter, Fullerton, CA, USA). The aqueous, micellar phase was collected, filtered  
270 (Chromafil PET filter, 0.2  $\mu$ m pore size, 25 mm diameter) and analyzed in terms of lipid  
271 composition and carotenoid content.

272 The lipase activity used in the present study was reduced in comparison to the amount  
273 suggested by the consensus method of Minekus et al. (2014) (2000 U/mL), but was  
274 comparable with the amount used in other, recent digestion studies (63-550 U/mL) (Sarkar et  
275 al., 2016; O'Sullivan et al., 2017; Lin & Wright, 2017). However, as explained in a previous  
276 study of Verkempinck et al. (2018b), all (emulsified) oil droplets were covered with lipase as

277 the added amount per digestion tube was still in excess. It must be noted that the obtained  
278 results might not be considered for *in vivo* human digestion predictions directly, but do gain  
279 more mechanistic insight into the lipolysis reactions.

## 280 **2.6 Physicochemical properties during digestion**

### 281 **2.6.1 Particle electrical charge**

282 The oil droplets  $\zeta$ -potential was measured using a dynamic light scattering electrophoresis  
283 equipment (Zetasizer NanoZS, Malvern Instruments, Worcestershire, UK). Emulsion, chyme  
284 or digest at the digestion end point was diluted 1:100 prior to analysis with reagent water (pH  
285 3), SGF or SIF, respectively. Subsequently, the sample was transferred into a capillary cell  
286 equipped with two electrodes to measure the  $\zeta$ -potential of the studied oil droplets.

### 287 **2.6.2 Particle size distribution and microstructure**

288 Particles size distributions were measured of all initial emulsions, after 2 hours of gastric  
289 phase and after 2 hours of intestinal phase, using a laser diffraction equipment (Beckman  
290 Coulter Inc., LS 13 320, Miami, Florida, USA) as mentioned in previous work (Verkempinck  
291 et al., 2018b). The reported particle sizes are volume-weighted median sizes ( $D[v;0.5]$  value),  
292 defined as the particle diameter at which 50% of the particles are smaller than this size.

293 Volume-weighted particle sizes are a more sensitive indicator for instability of emulsions in  
294 comparison with surface-weighted particle sizes, as the former largely increases with the  
295 presence of a limited number of large particles. In this way, the volume-weighted particle size  
296 can be an indicator for possible destabilization phenomena occurring along the digestive tract.

297 Complementary, microstructures after the gastric phase and at the digestion end point were  
298 visualized by differential interference contrast microscopy (Olympus BX41, Olympus  
299 Corporation, Tokyo, Japan) equipped with a digital camera (Olympus BX51, Olympus  
300 Optical Co. LTD., Tokyo, Japan).

## 301 **2.7 Quantitative analysis of multiple lipid digestion products**

302 Lipid extraction and analysis was exactly executed as described by Verkempinck et al.  
303 (2018b). The approach described by Verkempinck et al. (2018b) was used for analysis of the  
304 obtained lipid data. Stoichiometric reactions and mass balances were used to calculate the  
305 percentage of digested TAG at each time moment in the small intestine (%) and will be  
306 referred to as 'lipolysis extent'. This was based on the difference between the TAG amount in  
307 the emulsion (analytical value) and the sum of multiple lipolysis products released in the  
308 small intestine (MAG+FFA+GLY-H<sub>2</sub>O). MAG and FFA (analytical values) will be expressed  
309 in concentrations (mg/g<sub>emulsion</sub>). It must be noted that DAG were detected, but below the  
310 quantification limit and were therefore not taken into account in the data analysis.

## 311 **2.8 Carotenoid analysis**

312 Carotenoids were extracted and quantified by exactly performing the procedure described  
313 before by Verkempinck et al. (2018a). Carotenoid identification and quantification was done  
314 at the wavelength of maximal absorption for  $\alpha$ - and  $\beta$ -carotene (450 nm). *In vitro* carotenoid  
315 bioaccessibility (BAC) is expressed as a percentage value and is defined as the ratio of  
316 carotenoid amount incorporated into the micelle fraction to the initial carotenoid amount  
317 present in the emulsion, both expressed per g initial emulsion.

## 318 **2.9 Statistical analysis**

319 The statistical software programs JMP (JMP Pro13, SAS Institute Inc., Cary, NC, USA) and  
320 SAS (version 9.4, SAS Institute Inc., Cary, NC, USA) were used to perform all statistical  
321 analysis. Significant differences in particles electrical charge and sizes were conducted by one  
322 way ANOVA and the Tukey's Studentized Range Post-hoc test with a 95% level of  
323 significance (P<0.05). All particle charge and size measurements were performed in  
324 duplicate.

325 Nonlinear regression was used to model the experimental data obtained through the kinetic  
326 study for lipid digestion and carotenoid bioaccessibility. The empirical, most simple model  
327 best describing the experimental data was selected using model discrimination. As described  
328 before by Verkempinck et al. (2018b), a fractional conversion model was chosen and is  
329 characterized by a first order reaction until attainment of a plateau value (equation 1). Hereby  
330  $C$  (%) is the studied parameter at time  $t$  in the simulated intestinal phase;  $C_f$  (%) is the final,  
331 maximum of the studied parameter that can be obtained with the imposed conditions;  $C_0$  (%)  
332 is the initial studied parameter at time 0 in the simulated intestinal phase;  $k$  ( $\text{min}^{-1}$ ) is the  
333 reaction rate constant of the studied process and  $t$  (min) is time in the simulated intestinal  
334 phase. The TAG concentration at the start of the simulated intestinal phase was assumed to be  
335 equal to the TAG concentration present in the initial emulsion.

$$336 \quad C = C_f + (C_0 - C_f) \cdot e^{(-kt)} \quad (\text{equation 1})$$

337 This fractional conversion model could be simplified as lipid digestion at the start of the  
338 simulated intestinal phase could be neglected (equation 2). Therefore, the concentrations of  
339 MAG; FFA and GLY, and carotenoid concentration in the micellar fraction were assumed to  
340 be zero ( $C_0=0$ ) at this initial stage of digestion.

$$341 \quad C = C_f \cdot (1 - e^{(-kt)}) \quad (\text{equation 2})$$

342 The jointly estimated parameters ( $C_f$  and  $k$ ) were determined and visualized by calculating the  
343 90% joint confidence regions (JCRs). If regions of different samples overlap, it could be  
344 stated that the overall kinetics of these samples are not significantly different.



### 345 **3 Results and discussion**

346 Different carrot-enriched o/w emulsions were submitted to an *in vitro* digestion procedure,  
347 consisting on a simulated gastric and small intestinal phase. The emulsions were stabilized  
348 with different (i) mono-emulsifier types (Tween 80 (TW); phosphatidylcholine (PC) or citrus  
349 pectin (CP)) or (ii) di-emulsifier solutions (TWCP or PCCP) to evaluate the role of the pectin  
350 fiber in both the lipolysis as well as the carotenoid bioaccessibility reaction. The intestinal  
351 phase was studied in a kinetic way in order to gain more mechanistic insight in the process of  
352 lipolysis and micellarization of multiple lipid digestion products and carotenoids (sections 3.2  
353 and 3.3). Particles electrical charge and size were determined during digestion as the different  
354 emulsifier types and combinations might have a specific influence on the emulsion behavior  
355 under simulated gastrointestinal conditions (sections 3.1.1 and 3.1.2, respectively).

#### 356 **3.1 Evolution of physicochemical changes during *in vitro* digestion**

##### 357 **3.1.1 Particle electrical charge**

358 The particle electrical charge was measured for the emulsion, after the gastric phase and at the  
359 small intestinal digestion end point. This can provide more information about the organization  
360 of the emulsifier(s) at the oil-water interface and the phenomena that occur during simulated  
361 gastrointestinal conditions. The obtained particle charges are shown in **Table 1**.

362 The  $\zeta$ -potential of all initial emulsions with a pH of 3 varied between 3 and -31 mV  
363 depending on the emulsifier type and combination used (**Table 1**). The emulsion stabilized  
364 with the non-ionic TW had an initial  $\zeta$ -potential of  $-7.84 \pm 1.19$  mV. These negative charges  
365 might be attributed to the presence of impurities in the oil phase (e.g. free fatty acids; FFA) or  
366 the preferential adsorption of hydroxyl ions from the continuous phase to the hydrophilic head  
367 of the surfactant (McClements, 2016). By contrast, the PC emulsion was slightly positively  
368 charged ( $2.43 \pm 0.24$  mV), which could be explained by the shielding of the negatively

369 charged phosphate groups by hydrogen ions at low pH (Lin et al., 2014). The  $\zeta$ -potential of  
370 the CP emulsions was significantly influenced by the pectin DM and was more negative for  
371 CP10 compared to CP38 and CP82. This was attributed to the presence of more negatively  
372 chargeable carboxylic groups in LMP compared to MMP and HMP, even at low pH. By  
373 contrast, the TWCP emulsions had similar charges as the TW emulsion (**Table 1**) and no  
374 effect of DM was observed. Therefore, it could be hypothesized that the small molecule  
375 surfactant TW ( $\pm 1200$ - $1350$  Da) adsorbed much faster at the oil-water interface than the large  
376 pectin structures ( $\pm 55000$  Da). Consequently, the latter remained in the aqueous phase and  
377 did not attribute to the particle electrical charge. Oppositely, the charge of the PCCP  
378 emulsions showed an effect of pectin DM, being more negative when LMP was used for  
379 emulsion stabilization in comparison with MMP or HMP. In the case of these PCCP  
380 emulsions, probably both PC and CP adsorbed at the interface. More specifically, it can be  
381 postulated that a multilayer emulsion was formed with first phosphatidylcholine surrounding  
382 the oil droplet, followed by the oppositely charged pectin structures. As suggested by Guo et  
383 al. (2017), this multilayer emulsion might be more resistant to lipolysis than the monolayer  
384 emulsions.

385 After simulation of the gastric phase, an increase in particle charge was measured (**Table 1**).  
386 This was attributed to the interactions between the negatively charged oil-water surfaces and  
387 the added ions (such as  $\text{Ca}^{2+}$ ,  $\text{K}^+$  and  $\text{H}^+$ ). In general, similar trends were observed as in the  
388 initial emulsions. In this sense, the TW and TWCP emulsions had similar particle charges  
389 initially as well as after the gastric phase. The emulsions having CP at the interface (i.e. CP  
390 and PCCP emulsions) still showed a more negative charge with lower pectin DM (charge  
391  $\text{HMP} > \text{MMP} > \text{LMP}$ ).

392 More negatively charged particles were observed in all cases at the end of the intestinal phase  
393 (**Table 1**). Among all added fluids used to simulate the small intestinal phase, lipase was

394 responsible for hydrolysis of triacylglycerol (TAG) into diacylglycerol (DAG),  
395 monoacylglycerol (MAG), FFA and glycerol (GLY). According to Singh, Ye & Horne  
396 (2009), the more negative charge could be ascribed to the presence of structures formed  
397 containing bile salts and phospholipids, and as well to FFA release which are negatively  
398 charged at pH 7. An effect of pectin DM was no longer observed in the pectin-containing  
399 samples.

### 400 **3.1.2 Particle size distribution and microstructure**

401 The particle size distribution and microstructure were studied before as well as during  
402 digestion to evaluate the stability of the oil droplets along the simulated digestive tract. The  
403 particle size distributions are shown in **Figure 1**, while all microstructures are depicted in  
404 **Figure B** (supplementary material).

405 All initial emulsions had a particle size in the sub-micron range, with median oil droplet sizes  
406 between 1.01 and 1.82  $\mu\text{m}$  (**Table A, supplementary material**). The TW emulsion had the  
407 smallest median droplet size ( $1.27 \pm 0.10 \mu\text{m}$ ) of all mono-emulsifier emulsions (i.e.  
408 emulsions stabilized with one emulsifier type) and the PC emulsion had the largest droplet  
409 size ( $1.82 \pm 0.07 \mu\text{m}$ ). This might indicate that PC is not that efficient in stabilizing o/w  
410 emulsions compared to TW. McClements & Gumus (2016) described that PC is mostly used  
411 in combination with other emulsifiers, as PC has only an intermediate hydrophobicity. In case  
412 of the CP emulsions, the CP38 emulsion had significant smaller oil droplets than the CP82  
413 and CP10 emulsions. Similar results were observed by Verrijssen et al. (2015), whereby  
414 MMP exhibited smaller oil droplets. This observation was explained by the block-wise  
415 organization of the present hydrophilic and hydrophobic galacturonic acid molecules in  
416 MMP. TWCP emulsions had smaller particle sizes compared to both the TW and CP  
417 emulsion. As was mentioned in section 3.1.1, it was hypothesized that for the TWCP  
418 combination, the small molecule TW covered the complete interface, while pectin remained

419 in the aqueous phase. However, the presence of pectin might have created a competitional  
420 effect for the interface, whereby TW moved slightly faster to the interface and eventually led  
421 to smaller particle sizes for the TWCP emulsions. A similar trend of smaller particle sizes was  
422 observed in the di-emulsifier emulsion PCCP, in comparison with their mono-emulsifier  
423 emulsions. Additionally, an effect of pectin DM was observed in pectin-containing emulsions.  
424 More specifically, slightly smaller oil droplets were created using MMP as emulsifier  
425 compared to LMP or HMP. It is known that the initial oil droplet size of emulsions is an  
426 important characteristic which can have a big impact upon lipolysis kinetics (Zhang et al.,  
427 2016; Salvia-Trujillo et al., 2017). Nevertheless, the differences in initial oil droplet in the  
428 presented study are rather small and the emulsifier structure and organization at the interface  
429 might play a more dominating and so crucial role on the emulsions behavior during digestion  
430 and subsequently lipolysis kinetics.

431 Simulation of the gastric conditions influenced the particle size of the studied emulsions  
432 (**Figure 1 and Table A in supplementary material**). Two different situations were observed.  
433 On the one hand, the particle size distributions of the TW and PCCP emulsions remained  
434 similar, while on the other hand bimodal particle size distributions were observed for the  
435 TWCP, PC and CP emulsions (**Figure 1**). TW is known to be an acid stable emulsifier  
436 (Marciani et al., 2006; Verkempinck et al., 2018b). However TWCP emulsions were unstable  
437 in the gastric phase, which could be due to the presence of pectin in the aqueous phase,  
438 inducing coalescence and flocculation phenomena. The PCCP combination resulted in acid  
439 stable emulsions, which could be explained by the multilayer organization at the oil-water  
440 interface, being more resistant to the gastric conditions. By contrast, the emulsions stabilized  
441 with PC or CP only were acid unstable. This could be attributed to the addition of ions (such  
442 as  $\text{Ca}^{2+}$ ,  $\text{K}^{+}$  and  $\text{H}^{+}$ ) and consequently shift to more neutral particle charge (**Table 1**) which  
443 could have compromised the emulsion stability. In case of the acid unstable emulsifier

444 (combination), coalescence phenomena caused the observed particle size increase and were  
445 also visualized by microscopic pictures (**Figure B, supplementary material**).

446 Multimodal particle size distributions were detected at the digestion end point (**Figure 1**). The  
447 first intensity peak (1-3  $\mu\text{m}$ ) might be related to the formed micellar fraction and other  
448 vesicles, containing bioaccessible lipolysis products and carotenoids. The second main  
449 intensity peak (10-30  $\mu\text{m}$ ) might be related to the presence of undigested oil. For the TW  
450 emulsion, the first intensity peak is more pronounced than the second, which might be related  
451 to a relatively high lipolysis extent. In all other cases, the intensity peak related to undigested  
452 oil, was equally or more intense than the first peak. In fact, the presence of undigested oil at  
453 the digestion end point was also confirmed by microscopic pictures (**Figure B,**  
454 **supplementary material**). These observations showed that on the one hand the lipolysis  
455 reaction was incomplete for all the emulsions studied, but that on the other hand structures  
456 with sizes of 1-3  $\mu\text{m}$  were present which can contain bioaccessible lipolysis products and  
457 carotenoids.

458 In conclusion, the emulsions can be divided into two categories: acid stable emulsifier(s)  
459 (combinations) (i.e. TW and PCCP) and acid unstable emulsifier(s) (combinations) (i.e.  
460 TWCP, PC and CP). The increased particle size in the latter case compared to the initial  
461 emulsion, might have a direct impact upon lipolysis kinetics as the particle size at the end of  
462 the gastric phase determines the available area for digestive enzymes to adsorb in the  
463 intestinal phase. At the end of the small intestinal phase, multimodal particle size distributions  
464 were observed for all emulsions. The two main intensity peaks can be associated with (i)  
465 (colloidal) structures which can contain multiple lipid digestion products and carotenoids or  
466 (ii) large, undigested oil droplets.

## 467 **3.2 Study of lipid digestion kinetics as influenced by the presence of citrus** 468 **pectin**

469 Emulsion lipid digestion mainly occurs in the small intestine, where lipase hydrolyses TAG  
470 into DAG, MAG, FFA and GLY. Simultaneously, bile salts remove the formed multiple  
471 lipolysis products from the oil interface to build mixed micelles. In the present study, small  
472 intestinal lipid digestion of several emulsions was executed in a kinetic way. In this sense, the  
473 effect of time on TAG hydrolysis and formation of multiple lipid digestion products in the  
474 small intestinal phase can be studied. The obtained experimental data were modelled using  
475 nonlinear regression. A fractional conversion model (equation 2) best described the data. It is  
476 characterized by a linear increase ( $k$ -value) until attainment of a final plateau value ( $C_f$ ).  
477 Consequently, differences among the studied emulsions could be quantitatively evaluated. In  
478 a next step, the jointly estimated parameters ( $k$  and  $C_f$ ) were used as input for joint confidence  
479 region (JCR) (90%) analysis to determine whether these parameters were significantly  
480 different among all emulsions studied. The kinetic digestion approach and corresponding data  
481 analysis already showed large potential in clarifying differences in lipolysis kinetics and  
482 micelle formation in previous studies performed at our research unit (Salvia-Trujillo et al.,  
483 2017; Verkempinck et al., 2018b). The obtained kinetic data for the small intestinal lipolysis  
484 and the mixed micelle formation kinetics will be discussed in sections 3.2.1 and 3.2.2,  
485 respectively.

### 486 ***3.2.1 Lipid digestion kinetics in the small intestinal phase***

487 In all cases, a steady-state condition was reached within the 2 hours of simulated small  
488 intestinal digestion (**Figure 2**). However, depending on the emulsifier type and combination  
489 used for emulsion stabilization, major differences were observed regarding (*i*) the rate with  
490 which the steady-state condition was reached and (*ii*) the final concentration of the different  
491 lipid species (**Table 2**).

492 Evaluating the  $k$ -values of the TAG hydrolysis reaction of the pectin-based emulsions (i.e. all  
493 emulsions containing pectin) revealed that the pectin DM had a major effect on the reaction  
494 rate constants (**Table 2**). CP10-based emulsions (i.e. CP10; TWCP10 and PCCP10  
495 emulsions) reached the steady-state conditions much faster than CP38- or CP82-based  
496 emulsions. This suggests that the interface of CP10-based emulsions is of good quality for  
497 lipase adsorption and led to a fast lipolysis reaction. This can be attributed to a less compact  
498 organization of CP10 at the oil-water interface due to more present charged carboxylic  
499 groups, thus creating more inter-molecular repulsive forces and probably leave accessible  
500 spaces at the interface. Consequently, the interface can be more easily reached by digestive  
501 components. However, LMP is known to interact in a high extent with calcium ions (Espinal-  
502 Ruiz et al., 2014a). These ions are needed to remove the lipid digestion products from the oil  
503 surface. Consequently, it can be hypothesized that in the case of CP10-based emulsions, the  
504 lipolysis reaction reached its steady-state condition fast due to a relatively accessible oil-water  
505 interface which is thereafter quickly covered by lipolysis products since these were no longer  
506 removed by the calcium ions due to pectin-calcium interactions. When combining TW with  
507 CP to stabilize an emulsion, an increased  $k$ -value of the lipolysis reaction was observed  
508 (ranging from 0.092 to 0.253  $\text{min}^{-1}$ ) in comparison with the TW emulsion (0.086  $\text{min}^{-1}$ ). An  
509 opposite trend was observed when comparing the  $k$ -values of the PC versus PCCP emulsions  
510 as in the case of the PC emulsion, the  $k$ -value was 0.233  $\text{min}^{-1}$ , while it was ranging from  
511 0.052 to 0.147  $\text{min}^{-1}$  for the PCCP di-emulsifier emulsions. For the PCCP emulsion it was  
512 assumed that both emulsifier types were adsorbed at the oil-water interface. The presence of  
513 pectin at the interface might have led to interactions between pectin and the digestive  
514 components, slowing down the TAG hydrolysis. In addition, presence of the large pectin  
515 structures at the interface might hinder or retard the adsorption of lipase, bile and calcium at  
516 the oil droplet surface.

517 By contrast, the final percentage of digested TAG was mainly determined by the emulsifier  
518 (combination) used for emulsion stabilization (**Table 2**). TW emulsions presented the highest  
519 amount of digested TAG (83%), followed by the TWCP (72-77%), PCCP (6.5-7.3%), CP  
520 (4.9-5.5%) and PC (4.6%) emulsions. In case of the TWCP di-emulsion, the addition of pectin  
521 to the emulsion lowered the lipolysis extent in comparison with the TW mono-emulsion. The  
522 data of the particle charge (section 3.1.1) suggested that in case of these TWCP di-emulsions,  
523 TW was adsorbed at the oil-water interface, while the CP structures were located in the  
524 continuous phase. Consequently, it can be postulated that the CP structures present in the  
525 aqueous phase interact with components within the intestinal juices, such as lipase and bile  
526 salts, inhibiting the lipid digestion extent. A study of Espinal-Ruiz et al. (2014a) evaluated the  
527 interactions between pectin and individual components present during digestion and showed  
528 that pectin interacts with bile, lipase and calcium. When comparing the lipolysis extent of the  
529 PC, CP and PCCP emulsions a different trend could be observed. Namely, the combination of  
530 PC and CP at the interface improved the lipolysis extent in comparison with the mono-  
531 emulsifier emulsions. This could be attributed to presence of a double layer in the PCCP  
532 emulsions, improving the gastric stability. The PC and CP emulsions, unlike the PCCP  
533 emulsions, were unstable under gastric conditions (section 3.1.2). Consequently, coalescence  
534 phenomena occurred, resulting in oil droplets with a lower active surface area at the beginning  
535 of the small intestinal phase in case of the PC and CP emulsions.

536 Significant differences in the correlated parameters among all emulsions were visualized by  
537 JCR analysis, depicted in **Figure C in supplementary material**. These results confirm the  
538 observations described above: the  $k$ -values of the TAG hydrolysis reaction are predominantly  
539 determined by the DM of the present CP structures, while the final digested TAG percentages  
540 were mainly affected by the emulsifier (combination) used for emulsion formulation.



541 It could also be observed that the TAG hydrolysis was incomplete in all cases (**Table 2**). In  
542 case of the acid unstable emulsions (section 3.1.2), this could be partially attributed to the  
543 increased particle size at the end of the gastric phase. Consequently, a lower active surface  
544 area is available for lipase to adsorb. Similar observations were conducted in a study of  
545 Salvia-Trujillo (2017), showing that emulsions with large emulsified oil droplet sizes ( $\approx 15$   
546  $\mu\text{m}$ ) were digested in a lower extent compared to emulsions with small emulsified oil droplet  
547 sizes ( $\approx 2 \mu\text{m}$ ). The lipolysis reaction was also incomplete for acid stable emulsions. It can be  
548 hypothesized that in the case where pectin is adsorbed at the oil-water interface (i.e. CP and  
549 PCCP emulsions), pectin can inhibit the adsorption of digestive components (such as lipase,  
550 bile and calcium). As a result, these digestive components can no longer fulfill their function,  
551 leading to a diminished TAG hydrolysis reaction. Moreover, in the case of the PCCP  
552 emulsions, it was postulated that a multilayer emulsions was formed, which is assumed to be  
553 more resistant to lipase hydrolysis (section 3.1.1) (Guo et al., 2017). From **Table 2**, it can also  
554 be observed that emulsions containing CP82 led to a lower lipolysis extent in comparison to  
555 emulsions composed with CP10. The more hydrophobic HMP is known to interact in a higher  
556 extent with lipase, bile and phospholipids, in comparison with the more hydrophilic MMP and  
557 LMP (Tsujita et al., 2007; Espinal-Ruiz et al., 2014b) which might have diminished the  
558 lipolysis extent of the CP82-based emulsions.

559 In addition, it could be noted that the emulsions where pectin is located at the interface,  
560 present a relatively low TAG hydrolysis extent. Consequently, it can be stated that the  
561 presence of pectin at the oil-water interface has a negative effect on the TAG hydrolysis  
562 extent. By contrast, the emulsions stabilized by TW showed a more complete lipolysis  
563 reaction so the presence of pectin structures in the continuous aqueous phase does not  
564 significantly change the lipolysis reaction extent (**Figure C1, supplementary material**).

565 By digesting TAG molecules, a simultaneous release of MAG and FFA in the digest was  
566 detected (**Figure 2**). It must be noted that DAG were detected but below the quantification  
567 limit so were not taken into account in the discussion. GLY concentrations present in the  
568 digest were relatively low and no time effect could be observed, so these observations were  
569 omitted from the data analysis. Regarding the kinetics of MAG and FFA, similar observations  
570 were conducted as in the TAG hydrolysis reaction. In this sense, the release rate constant was  
571 mainly influenced by the pectin DM, while the extent of release MAG and FFA was  
572 determined by the emulsifier(s) used for emulsion stabilization (**Figure 2, Table 2 and**  
573 **Figure C2 and C3 in supplementary material**). Additionally, for all pectin containing  
574 emulsions, higher  $k$ -values were observed for MAG release in comparison to FFA release. In  
575 other words, the steady-state condition was reached slower for FFA release. This might be  
576 attributed to a further conversion of MAG into FFA and GLY (Mu & Høy, 2004; Tokle, Mao,  
577 & McClements, 2013).

578 In conclusion, it can be stated that the correlated parameters  $k$  and  $C_f$  can be influenced by  
579 different emulsion characteristics. More specifically,  $k$ -values were mainly influenced by the  
580 pectin DM, while the extent of TAG hydrolysis and subsequent FFA and MAG release was  
581 determined by the emulsifier type and combination present in the emulsion. These results also  
582 showed that specific emulsion characteristics might be selected with the aim of tailoring lipid  
583 digestion kinetics.

### 584 **3.2.2 Composition and formation kinetics of the micellar fraction**

585 The hydrolysis of TAG molecules led to the release of MAG and FFA in the digest and a  
586 simultaneous micellarization of these lipid digestion products could be observed. This is in  
587 line with the mechanism of lipid digestion and micelle formation described in literature (Mun,  
588 Decker, & McClements, 2007; Singh, Ye, & Horne, 2009). The micellarization kinetics were

589 strongly related to the kinetics of FFA and MAG release in the digest (section 3.2.1) and are  
590 summarized below.

591 Similarly to the observations in the digest, the reaction rate constant  $k$  for MAG and FFA  
592 micellarization was mainly effected by the pectin DM (**Table 2**). In most cases, the  $k$ -values  
593 for TAG hydrolysis, MAG and FFA release in the digest were of the same magnitude for a  
594 certain emulsion type, showing that the lipolysis reaction and ensuing release of MAG and  
595 FFA are not the rate limiting steps in the lipid digestion process. The final concentration lipid  
596 digestion products present in the micellar fraction was predominantly determined by the  
597 emulsifier (combination) used for emulsion stabilization as was also observed for the lipolysis  
598 reaction. It can be stated that the presence of pectin in emulsions does not have an  
599 unambiguous effect on the lipolysis and subsequent micelle formation kinetics. On the one  
600 hand, pectin can interact with lipolysis components (such as lipase, bile salts and calcium),  
601 diminishing the final lipolysis extent. On the other hand, the presence of pectin can improve  
602 the quality of the oil-water interface or improve the emulsion stability during gastrointestinal  
603 simulation, leading to an increased lipid digestion extent and subsequent release and  
604 micellarization of MAG and FFA.

605 The obtained data were also evaluated by JCR (90%) analysis and the results are depicted in  
606 **Figure C (supplementary material)**. These graphs confirm the above described trends and  
607 clearly show that the TW and TWCP emulsions present higher MAG and FFA concentrations  
608 in the micellar fraction than the PC, CP and PCCP emulsions. Moreover, it can be observed  
609 that not all MAG and FFA were micellarized which could be attributed to interactions  
610 between the lipid digestion products and pectin, calcium and other salts present in the  
611 simulated intestinal fluid.

612 In conclusion, the emulsifier (combination) used for emulsion stabilization had a significant  
613 effect on the final micellarised concentrations of MAG and FFA. By contrast, the reaction rate

614 constant of the incorporation of MAG and FFA into mixed micelles was mainly determined  
615 by the pectin DM. These differences in composition of these mixed micelles will most likely  
616 have an important effect on the carotenoid bioaccessibility kinetics, which will be further  
617 discussed in section 3.3.1.

### 618 **3.3 Study of carotenoid bioaccessibility kinetics as influenced by the** 619 **presence of citrus pectin**

620 Carotenoid bioaccessibility (BAC) was studied to evaluate (i) the time dependency of  
621 carotenoid micellarization and (ii) its relationship with lipolysis kinetics. Therefore,  
622 carotenoid BAC kinetics were determined in parallel with the lipolysis kinetics, so using the  
623 same kinetic approach as described in section 3.2. Briefly, the small intestinal phase was  
624 simulated in a kinetic way, whereafter the experimental data were modelled with a fractional  
625 conversion model. This resulted in the obtainment of a reaction rate constant  $k$  and final  $BAC_f$   
626 value of the process. Subsequently, these jointly estimated parameters were used as input to  
627 determine 90% JCR. In this way, significant differences among all studied emulsions could be  
628 determined and visualized.

#### 629 **3.3.1 Carotenoid bioaccessibility kinetics**

630 Micellarisation of  $\alpha$ - and  $\beta$ -carotene increased linearly as function of small intestinal digestion  
631 time until a steady-state condition was reached (**Figure 3**). This steady-state condition was  
632 reached in all cases within two hours of small intestinal digestion. Similar to lipid digestion  
633 and micelle formation, the kinetics of carotenoid bioaccessibility (BAC) were influenced by  
634 the emulsifier type, emulsifier combination and pectin DM stabilizing the o/w emulsions  
635 (**Figure 3, Table 3 and Figure D in supplementary material**) and will be summarized  
636 below.

637 The highest carotenoid micellarization rate constant was observed in case of the PC emulsion  
638 ( $0.283 \text{ min}^{-1}$  for  $\alpha$ -carotene and  $0.134 \text{ min}^{-1}$  for  $\beta$ -carotene), while the other emulsions showed  
639 similar  $k$ -values, varying between  $0.013$  and  $0.073 \text{ min}^{-1}$ . The carotenoid micellarization rate  
640 constant of emulsion having pectin adsorbed at the oil-water interface (i.e. CP and PCCP  
641 emulsions) were lower for CP82 than for CP38 and CP10. By contrast, when pectin was  
642 present in the aqueous phase (i.e. TWCP emulsions) a reverse trend was observed (CP82 >  
643 CP38 > CP10). It can be hypothesized that in the former case, the more hydrophobic CP82  
644 interacts in a higher extent with the lipophilic carotenoids compared to CP38 and CP10,  
645 retarding the incorporation of carotenoids into the micellar fraction. By contrast in the case  
646 where CP is present in the aqueous phase, CP10 might have led to an increased viscosity of  
647 the digest by interactions with calcium, which in turn can slow down the migration of  
648 carotenoids towards the mixed micelles (Verrijssen et al., 2015).

649 The final carotenoid BAC was predominantly determined by the emulsifier(s) used for  
650 emulsions stabilization. In this sense, the TW emulsion presented the highest carotenoid BAC  
651 values (69-82%), followed by the TWCP emulsions (37.8-64.3%), the PCCP emulsions (19.4-  
652 25.4%) and finally the PC (13.8-17.1%) and CP (11.1-19.4%) emulsions, which presented the  
653 lowest  $BAC_f$ -values. The addition of pectin in the case of TW emulsions, led to a diminished  
654 amount of micellized carotenoids, namely from 82% to 40-64% for  $\alpha$ -carotene and from  
655 69% to 38-56% for  $\beta$ -carotene. This can be explained by the interactions occurring between  
656 pectin structures and the lipophilic carotenoids. In addition, the presence of pectin at the  
657 interface does not improve the carotenoid bioaccessibility. More specifically, the final  
658 carotenoid BAC values obtained in case of the CP and PCCP emulsions were rather low and  
659 ranged from 11% to 25%. This could be related to the low amount of digested TAG molecules  
660 (section 3.2.1). In this way, carotenoids remained entrapped in the undigested oil and could  
661 not be transferred to mixed micelles.

### 662 **3.3.2 Relation between the kinetics of lipid digestion and carotenoid bioaccessibility**

663 The estimated parameters  $k$  and  $C_f$  of the micellarization of MAG, FFA and  $\beta$ -carotene were  
664 used to evaluate the relation between the two micellarization phenomena (**Figure E,**  
665 **supplementary material**) as previous research has indicated that there is a strong link  
666 between the lipid digestion reaction and carotenoid bioaccessibility (Salvia-Trujillo et al.,  
667 2017; Verkempinck et al., 2018a).

668 For the  $k$ -values, it can be seen that no direct link was observed between the rate constants of  
669 lipid digestion and carotenoid bioaccessibility. This can be attributed to the presence of pectin  
670 with different DM, which effects the reaction differently depending on the emulsion  
671 composition. Regarding the final micellarized amounts of MAG, FFA and carotenoids, an  
672 strong interrelation could be observed between the MAG and FFA concentrations  
673 incorporated into micelles and the extent of carotenoid bioaccessibility. In this sense, a higher  
674 concentration of incorporated MAG and FFA into mixed micelles, eventually led to a higher  
675 carotenoid bioaccessibility.

676 In the context of satiety and human health, some emulsion compositions can be of interest.  
677 For example, the TW mono-emulsion and TWCP di-emulsion present a relatively slow lipid  
678 digestion rate and finally result in a high lipolysis rate and relatively high carotenoid  
679 bioaccessibility. A slow lipid digestion increases the hormone-induced satiety and slows  
680 down gastric emptying, which consequently reduces energy intake (Ohlsson et al., 2014).  
681 Additionally, a high lipolysis extent is desired as undigested lipids in the large intestine are  
682 associated with several negative effects, such as a bloated feeling, steatorrhoea and an  
683 increased risk of tumor development in the colon (Hoyles & Wallace, 2010).

684

## 685 **4 Conclusions**

686 The presented work aimed to explore the effect of the presence of pectin in emulsions on the  
687 kinetics of lipid digestion and carotenoid bioaccessibility. The emulsifier(s) (combination)  
688 used for emulsion stabilization determined the composition of the oil-water interface and  
689 surrounding medium, but however presented similar small initial particle sizes. Subsequently,  
690 the different emulsion compositions affected the particle size and charge during digestion and  
691 moreover, had a significant impact on the lipid digestion reaction and micellarization of  
692 multiple lipolysis products and carotenoids. The reaction rate constant  $k$  for lipid hydrolysis  
693 was mainly influenced by the pectin DM. By contrast, the final percentage digested TAG,  
694 released MAG and FFA, and micellarized MAG, FFA and carotenoids was predominantly  
695 influenced by the emulsifier(s) (combination) used for emulsion stabilization. An interrelation  
696 was observed between the extent of micellarization of MAG, FFA and  $\beta$ -carotene. Namely, a  
697 higher concentration of incorporated MAG and FFA into the micellar fraction is associated  
698 with a higher final carotenoid bioaccessibility. Finally, it can be stated that the intake of fibers  
699 alters lipid digestion extent which was hypothesized before the study. Moreover, the results  
700 proved that targeted emulsion design can be used as a tool for tailoring lipid digestion and/or  
701 carotenoid bioaccessibility.

702

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709 ***Declaration of interests***

710 The authors of the present work declare no conflict of interests.

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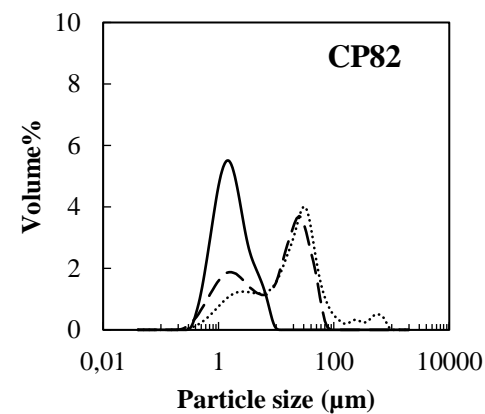
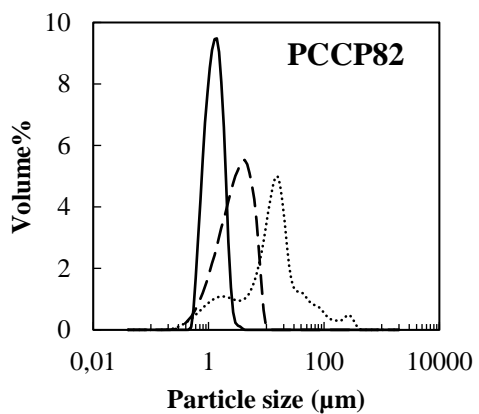
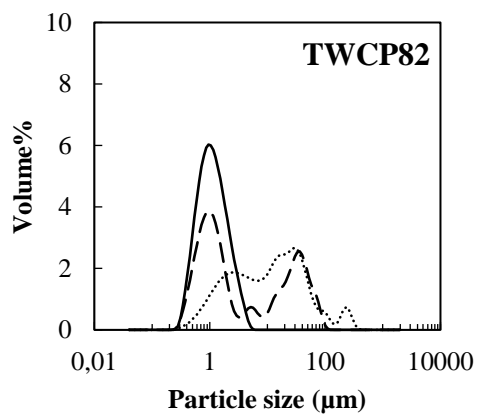
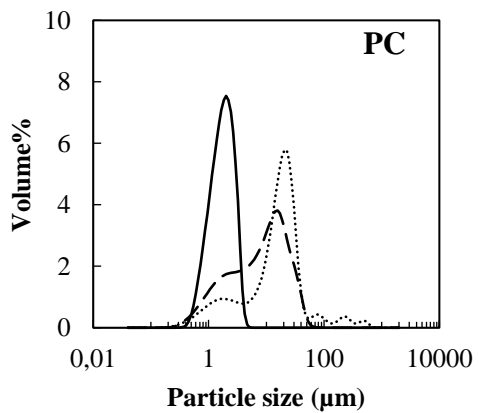
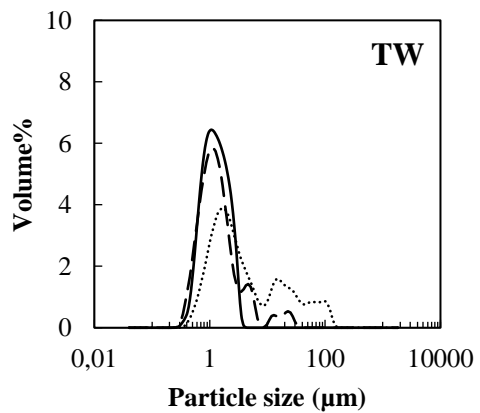
833 **Table 1: Changes on the  $\zeta$ -potential of the initial emulsion, chyme and digest. Different letters in the same**  
 834 **column indicate significant differences (Tukey test:  $p < 0.05$ ) among the samples. For interpretation of the**  
 835 **abbreviations is referred to the abbreviation list.**

	Emulsion		Gastric phase		Small intestinal phase	
TW	$-7.84 \pm 1.19$	BC	$-3.18 \pm 0.70$	AB	$-31.15 \pm 0.26$	ABC
PC	$2.43 \pm 0.24$	A	$-1.35 \pm 0.29$	A	$-33.80 \pm 1.70$	BCD
CP82	$-14.05 \pm 1.13$	D	$-5.07 \pm 0.25$	B	$-38.95 \pm 2.59$	D
CP38	$-23.13 \pm 0.13$	F	$-10.00 \pm 0.49$	C	$-33.50 \pm 2.69$	CD
CP10	$-30.95 \pm 0.54$	G	$-15.13 \pm 0.59$	D	$-36.53 \pm 2.56$	CD
TWCP82	$-5.60 \pm 0.65$	B	$-3.34 \pm 0.61$	AB	$-31.05 \pm 2.43$	ABC
TWCP38	$-7.53 \pm 0.02$	BC	$-3.16 \pm 0.21$	AB	$-26.38 \pm 1.80$	A
TWCP10	$-9.59 \pm 3.42$	C	$-3.07 \pm 0.24$	AB	$-28.70 \pm 1.56$	AB
PCCP82	$0.03 \pm 0.06$	A	$-4.08 \pm 3.57$	AB	$-39.10 \pm 1.27$	D
PCCP38	$-19.53 \pm 0.42$	E	$-4.85 \pm 0.47$	B	$-37.68 \pm 5.50$	D
PCCP10	$-23.93 \pm 1.06$	F	$-11.07 \pm 1.03$	C	$-35.37 \pm 2.88$	CD

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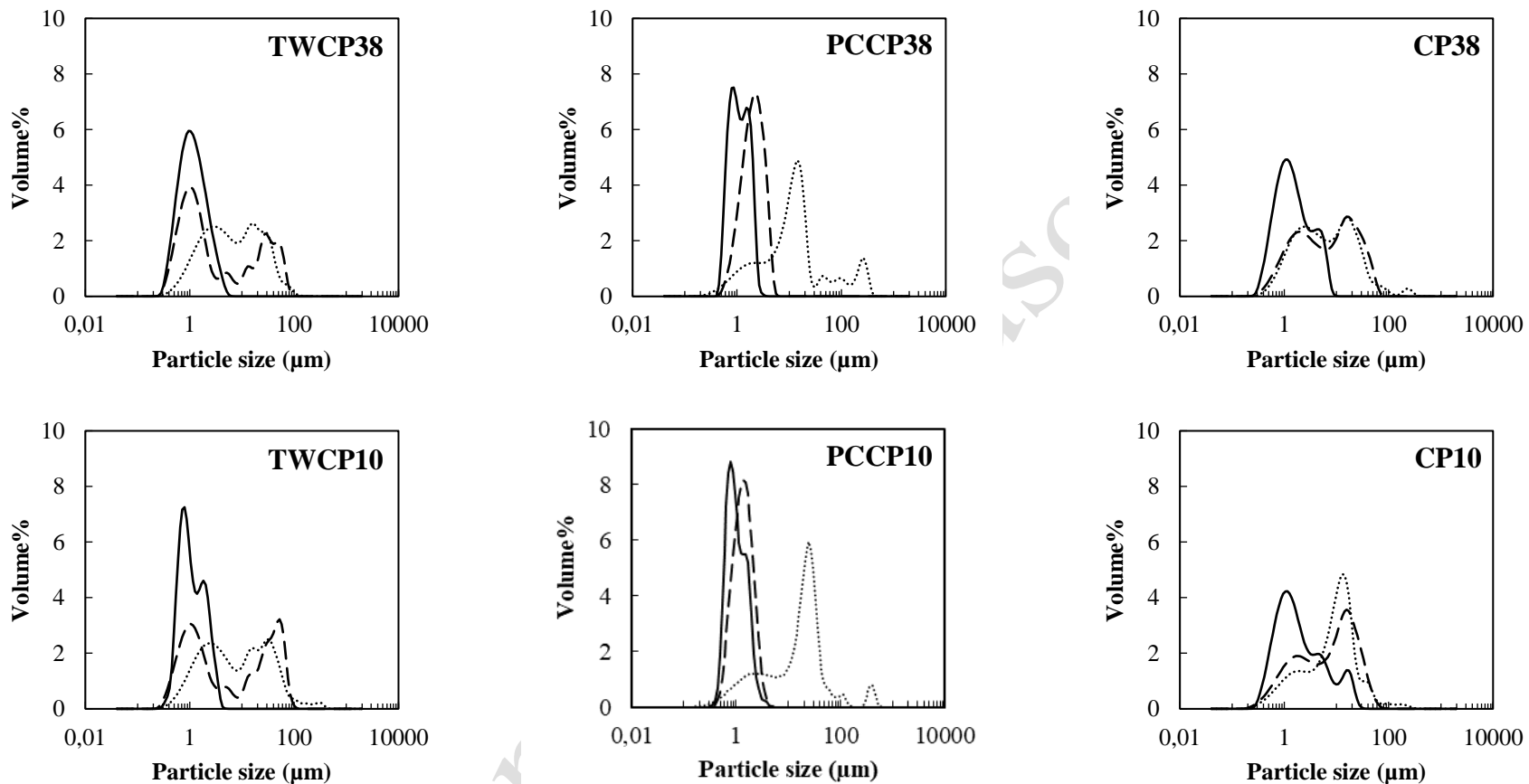
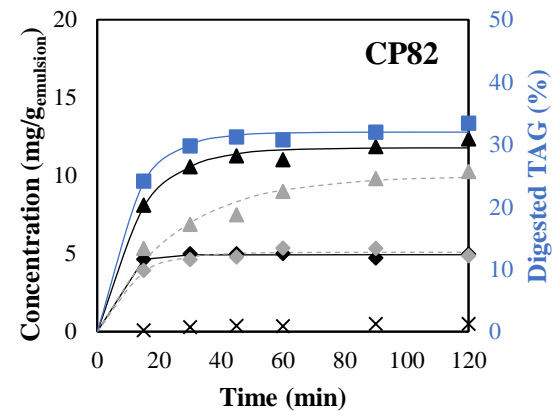
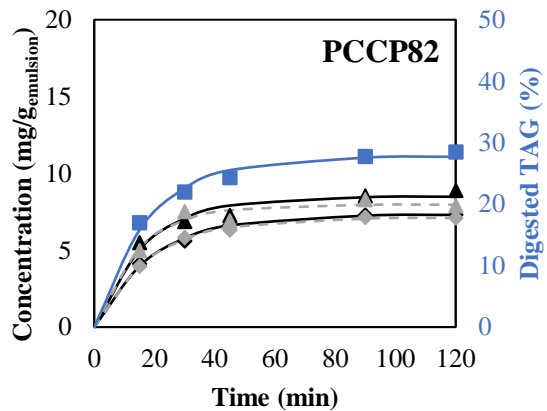
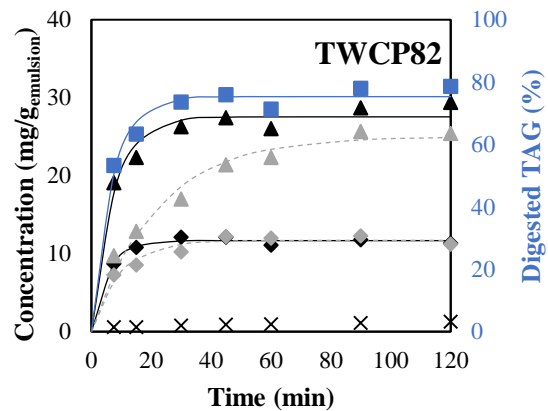
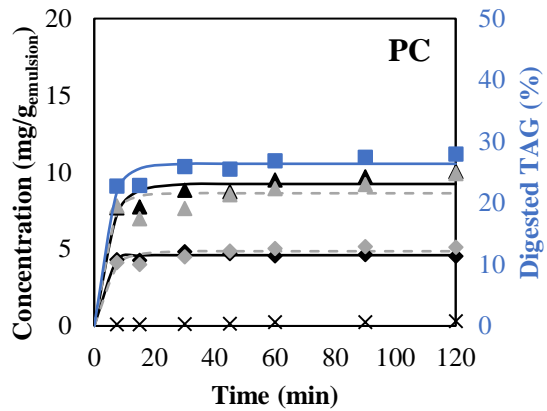
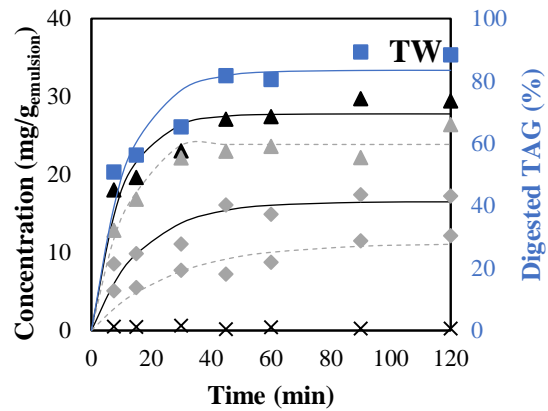


Figure 1: Particle size distribution of all carrot-enriched oil-in-water emulsions during digestion (full line: initial emulsion; dashed line: chyme (2h); dotted line: digest (2h)). For interpretation of the abbreviations is referred to the abbreviation list.



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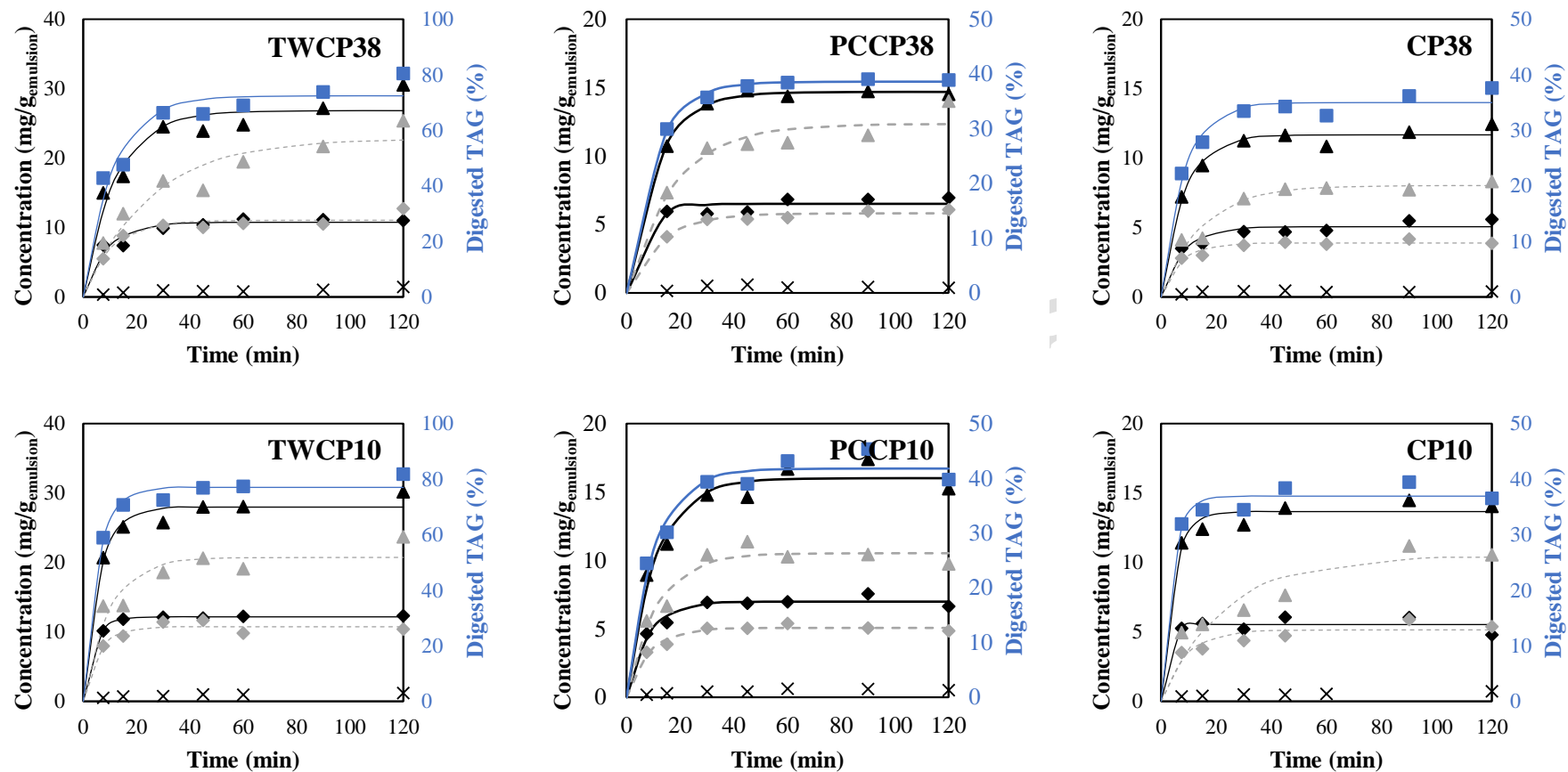


Figure 2: *In vitro* lipid digestion of carrot-based emulsions as function of small intestinal digestion. The symbols indicate the experimental data while the full and dotted lines show the predicted values of the corresponding fractional conversion model in the digest and micellar fraction, respectively (■ TAG<sub>digest</sub>; ◆ MAG<sub>digest</sub>; ◆ MAG<sub>micelle</sub>; ▲ FFA<sub>digest</sub>; ▲ FFA<sub>micelle</sub>; × GLY<sub>digest</sub>). For representation in color is referred to the online version and for the interpretation of the abbreviations is referred to the abbreviation list.

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**Table 2: Estimated kinetic parameters of the different carrot-based emulsions modelled by a fractional conversion model through a nonlinear regression procedure.  $C_f$  is the final lipid concentration estimated by the model (expressed in % in case of TAG and mg/g emulsion in all other cases) and  $k$  is the incorporation rate constant of the lipids into micelles ( $\text{min}^{-1}$ ). For interpretation of the abbreviations is referred to the abbreviation list.**

	TAG <sub>digest</sub>	MAG <sub>digest</sub>	FFA <sub>digest</sub>	MAG <sub>micelle</sub>	FFA <sub>micelle</sub>
<b>TW</b>					
$k \pm \text{stdev}$	0.086 $\pm$ 0.018	0.060 $\pm$ 0.014	0.100 $\pm$ 0.020	0.038 $\pm$ 0.012	0.090 $\pm$ 0.012
$C_f \pm \text{stdev}$	83.40 $\pm$ 3.93	16.51 $\pm$ 1.01	27.77 $\pm$ 1.16	11.17 $\pm$ 1.16	23.86 $\pm$ 0.70
$R^2_{\text{adj}}$	0.923	0.906	0.933	0.851	0.962
<b>PC</b>					
$k \pm \text{stdev}$	0.233 $\pm$ 0.045	0.344 $\pm$ 0.068	0.203 $\pm$ 0.045	0.209 $\pm$ 0.049	0.254 $\pm$ 0.105
$C_f \pm \text{stdev}$	26.38 $\pm$ 0.64	4.60 $\pm$ 0.07	9.24 $\pm$ 0.29	4.86 $\pm$ 0.16	8.63 $\pm$ 0.41
$R^2_{\text{adj}}$	0.969	0.988	0.950	0.945	0.889
<b>CP82</b>					
$k \pm \text{stdev}$	0.092 $\pm$ 0.007	0.188 $\pm$ 0.028	0.076 $\pm$ 0.007	0.092 $\pm$ 0.013	0.040 $\pm$ 0.006
$C_f \pm \text{stdev}$	31.98 $\pm$ 0.43	4.93 $\pm$ 0.05	11.78 $\pm$ 0.22	5.09 $\pm$ 0.12	9.98 $\pm$ 0.43
$R^2_{\text{adj}}$	0.994	0.995	0.990	0.980	0.893
<b>CP38</b>					
$k \pm \text{stdev}$	0.121 $\pm$ 0.014	0.124 $\pm$ 0.025	0.120 $\pm$ 0.012	0.139 $\pm$ 0.022	0.068 $\pm$ 0.010
$C_f \pm \text{stdev}$	34.99 $\pm$ 0.82	5.06 $\pm$ 0.20	11.67 $\pm$ 0.23	3.88 $\pm$ 0.11	8.02 $\pm$ 0.31
$R^2_{\text{adj}}$	0.976	0.934	0.983	0.963	0.958
<b>CP10</b>					
$k \pm \text{stdev}$	0.253 $\pm$ 0.049	0.402 $\pm$ 0.265	0.223 $\pm$ 0.039	0.113 $\pm$ 0.030	0.044 $\pm$ 0.013
$C_f \pm \text{stdev}$	36.94 $\pm$ 0.88	5.53 $\pm$ 0.22	13.66 $\pm$ 0.34	5.14 $\pm$ 0.31	10.44 $\pm$ 0.98
$R^2_{\text{adj}}$	0.977	0.933	0.976	0.892	0.872
<b>TWCP82</b>					
$k \pm \text{stdev}$	0.148 $\pm$ 0.016	0.181 $\pm$ 0.017	0.138 $\pm$ 0.018	0.104 $\pm$ 0.015	0.046 $\pm$ 0.006
$C_f \pm \text{stdev}$	75.34 $\pm$ 1.41	11.67 $\pm$ 0.17	27.54 $\pm$ 0.67	11.70 $\pm$ 0.36	25.00 $\pm$ 0.98
$R^2_{\text{adj}}$	0.984	0.989	0.973	0.962	0.973
<b>TWCP38</b>					
$k \pm \text{stdev}$	0.089 $\pm$ 0.016	0.112 $\pm$ 0.021	0.082 $\pm$ 0.016	0.096 $\pm$ 0.017	0.040 $\pm$ 0.009
$C_f \pm \text{stdev}$	72.42 $\pm$ 2.90	10.75 $\pm$ 0.41	26.83 $\pm$ 1.20	11.04 $\pm$ 0.42	22.79 $\pm$ 1.71
$R^2_{\text{adj}}$	0.943	0.940	0.932	0.947	0.915
<b>TWCP10</b>					
$k \pm \text{stdev}$	0.185 $\pm$ 0.021	0.240 $\pm$ 0.007	0.169 $\pm$ 0.024	0.170 $\pm$ 0.033	0.101 $\pm$ 0.027
$C_f \pm \text{stdev}$	77.03 $\pm$ 1.48	12.17 $\pm$ 0.05	27.97 $\pm$ 0.70	10.74 $\pm$ 0.37	20.73 $\pm$ 1.28
$R^2_{\text{adj}}$	0.986	0.999	0.977	0.959	0.895
<b>PCCP82</b>					
$k \pm \text{stdev}$	0.056 $\pm$ 0.005	0.052 $\pm$ 0.002	0.060 $\pm$ 0.009	0.054 $\pm$ 0.002	0.069 $\pm$ 0.012
$C_f \pm \text{stdev}$	27.74 $\pm$ 0.69	7.32 $\pm$ 0.09	8.49 $\pm$ 0.31	7.11 $\pm$ 0.08	7.97 $\pm$ 0.31
$R^2_{\text{adj}}$	0.989	0.997	0.973	0.998	0.968
<b>PCCP38</b>					
$k \pm \text{stdev}$	0.097 $\pm$ 0.004	0.147 $\pm$ 0.050	0.089 $\pm$ 0.004	0.080 $\pm$ 0.009	0.057 $\pm$ 0.012
$C_f \pm \text{stdev}$	38.59 $\pm$ 0.24	6.51 $\pm$ 0.23	14.69 $\pm$ 0.12	5.81 $\pm$ 0.12	12.35 $\pm$ 0.62
$R^2_{\text{adj}}$	0.999	0.951	0.998	0.987	0.941
<b>PCCP10</b>					
$k \pm \text{stdev}$	0.099 $\pm$ 0.013	0.126 $\pm$ 0.016	0.091 $\pm$ 0.013	0.120 $\pm$ 0.014	0.088 $\pm$ 0.015
$C_f \pm \text{stdev}$	41.78 $\pm$ 1.20	6.99 $\pm$ 0.17	16.01 $\pm$ 0.50	5.07 $\pm$ 0.12	10.53 $\pm$ 0.39
$R^2_{\text{adj}}$	0.968	0.973	0.964	0.977	0.952

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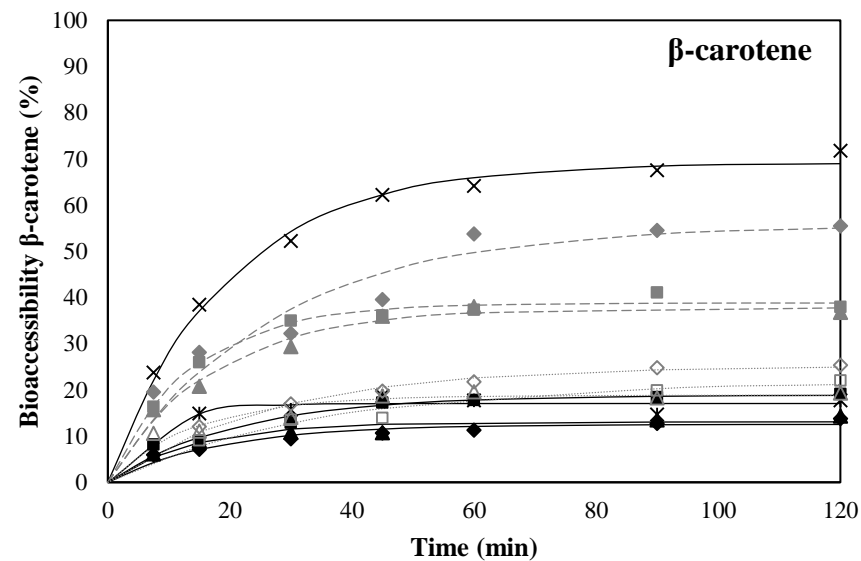
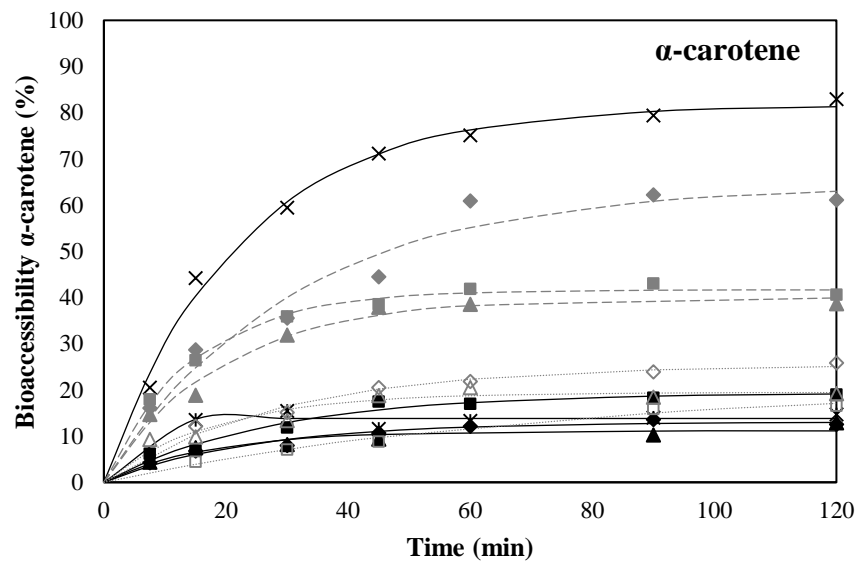


Figure 3: Time dependent evolution as function of small intestinal digestion time of  $\alpha$ - and  $\beta$ -carotene. The symbols indicate the experimental data, while the lines show the predicted values of the corresponding fractional conversion model. Full lines represent mono-emulsifier emulsions ( $\times$  TW;  $*$  PC;  $\blacksquare$  CP82;  $\blacklozenge$  CP38 and  $\blacktriangle$  CP10); dashed lines represent the tween-pectin emulsions ( $\blacksquare$  TWCP82;  $\blacklozenge$  TWCP38 and  $\blacktriangle$  TWCP10) and dotted lines represent the phosphatidylcholine-pectin emulsions ( $\blacksquare$  PCCP82;  $\blacklozenge$  PCCP38 and  $\blacktriangle$  PCCP10). For interpretation of the abbreviations is referred to the abbreviation list.

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**Table 3: Estimated kinetic parameters ( $\pm$  standard deviation) of the different carotenoid-enriched emulsions modelled by a fractional conversion model through a nonlinear regression procedure ( $k$ : incorporation rate constant of carotenoids into micelles and  $BAC_f$ : final bioaccessibility estimated by the model). For interpretation of the abbreviations is referred to the abbreviation list.**

	$\alpha$ -carotene		$\beta$ -carotene	
	<b>TW</b>			
$k \pm \text{stdev} (\text{min}^{-1})$	0.045 $\pm$	0.003	0.051 $\pm$	0.003
$BAC_f \pm \text{stdev} (\%)$	81.66 $\pm$	1.58	69.12 $\pm$	1.21
$R^2_{\text{adj}}$	0.993		0.994	
	<b>PC</b>			
$k \pm \text{stdev} (\text{min}^{-1})$	0.283 $\pm$	0.491	0.134 $\pm$	0.047
$BAC_f \pm \text{stdev} (\%)$	13.82 $\pm$	0.60	17.06 $\pm$	0.69
$R^2_{\text{adj}}$	0.922		0.937	
	<b>CP82</b>			
$k \pm \text{stdev} (\text{min}^{-1})$	0.037 $\pm$	0.006	0.048 $\pm$	0.007
$BAC_f \pm \text{stdev} (\%)$	19.36 $\pm$	0.98	18.87 $\pm$	0.85
$R^2_{\text{adj}}$	0.965		0.961	
	<b>CP38</b>			
$k \pm \text{stdev} (\text{min}^{-1})$	0.041 $\pm$	0.006	0.056 $\pm$	0.010
$BAC_f \pm \text{stdev} (\%)$	13.05 $\pm$	0.60	12.54 $\pm$	0.64
$R^2_{\text{adj}}$	0.967		0.940	
	<b>CP10</b>			
$k \pm \text{stdev} (\text{min}^{-1})$	0.058 $\pm$	0.014	0.071 $\pm$	0.014
$BAC_f \pm \text{stdev} (\%)$	11.14 $\pm$	0.78	13.09 $\pm$	0.70
$R^2_{\text{adj}}$	0.905		0.935	
	<b>TWCP82</b>			
$k \pm \text{stdev} (\text{min}^{-1})$	0.069 $\pm$	0.004	0.073 $\pm$	0.005
$BAC_f \pm \text{stdev} (\%)$	41.65 $\pm$	0.61	38.81 $\pm$	0.67
$R^2_{\text{adj}}$	0.994		0.991	
	<b>TWCP38</b>			
$k \pm \text{stdev} (\text{min}^{-1})$	0.032 $\pm$	0.005	0.037 $\pm$	0.008
$BAC_f \pm \text{stdev} (\%)$	64.32 $\pm$	3.76	55.67 $\pm$	3.75
$R^2_{\text{adj}}$	0.962		0.938	
	<b>TWCP10</b>			
$k \pm \text{stdev} (\text{min}^{-1})$	0.052 $\pm$	0.006	0.058 $\pm$	0.006
$BAC_f \pm \text{stdev} (\%)$	39.99 $\pm$	1.39	37.79 $\pm$	1.22
$R^2_{\text{adj}}$	0.982		0.982	
	<b>PCCP82</b>			
$k \pm \text{stdev} (\text{min}^{-1})$	0.013 $\pm$	0.003	0.029 $\pm$	0.005
$BAC_f \pm \text{stdev} (\%)$	21.37 $\pm$	2.18	21.83 $\pm$	1.40
$R^2_{\text{adj}}$	0.984		0.963	
	<b>PCCP38</b>			
$k \pm \text{stdev} (\text{min}^{-1})$	0.035 $\pm$	0.004	0.038 $\pm$	0.003
$BAC_f \pm \text{stdev} (\%)$	25.44 $\pm$	0.97	25.20 $\pm$	0.65
$R^2_{\text{adj}}$	0.981		0.990	
	<b>PCCP10</b>			
$k \pm \text{stdev} (\text{min}^{-1})$	0.055 $\pm$	0.011	0.070 $\pm$	0.014
$BAC_f \pm \text{stdev} (\%)$	19.45 $\pm$	1.11	18.85 $\pm$	0.95
$R^2_{\text{adj}}$	0.926		0.925	

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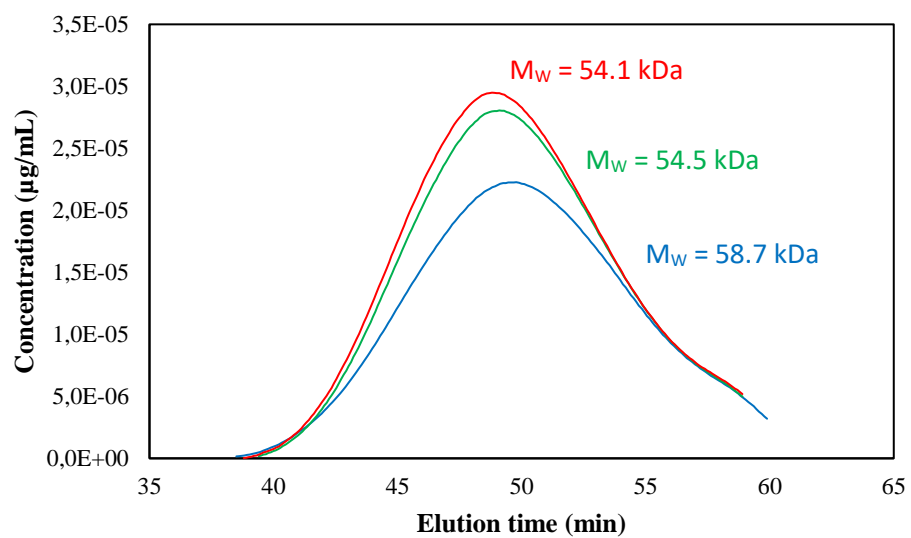


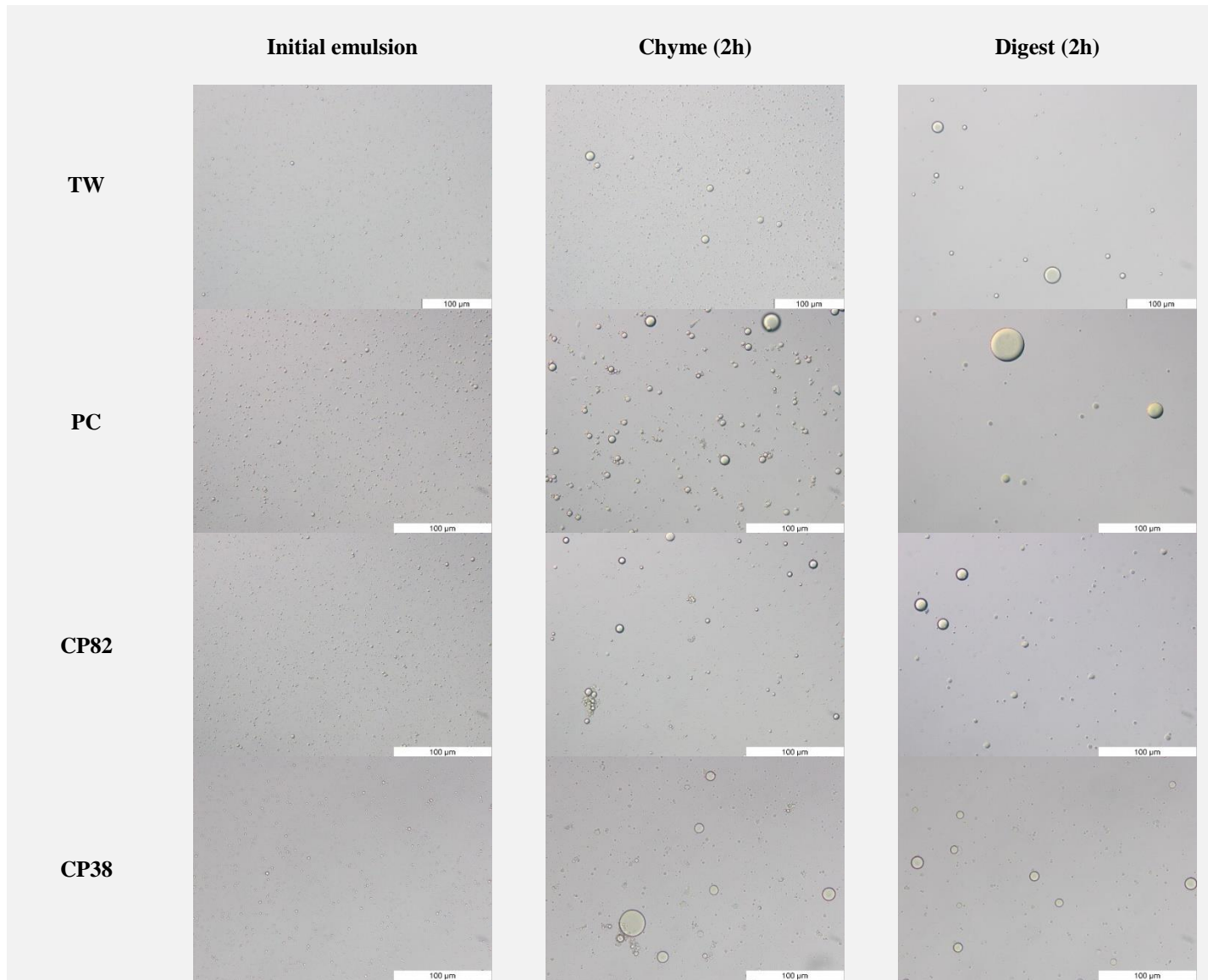
Figure A: Concentration profile and corresponding weight average molecular weight ( $M_w$ ) of the pectin samples (■ CP82; ◆ CP38 and ▲ CP10).

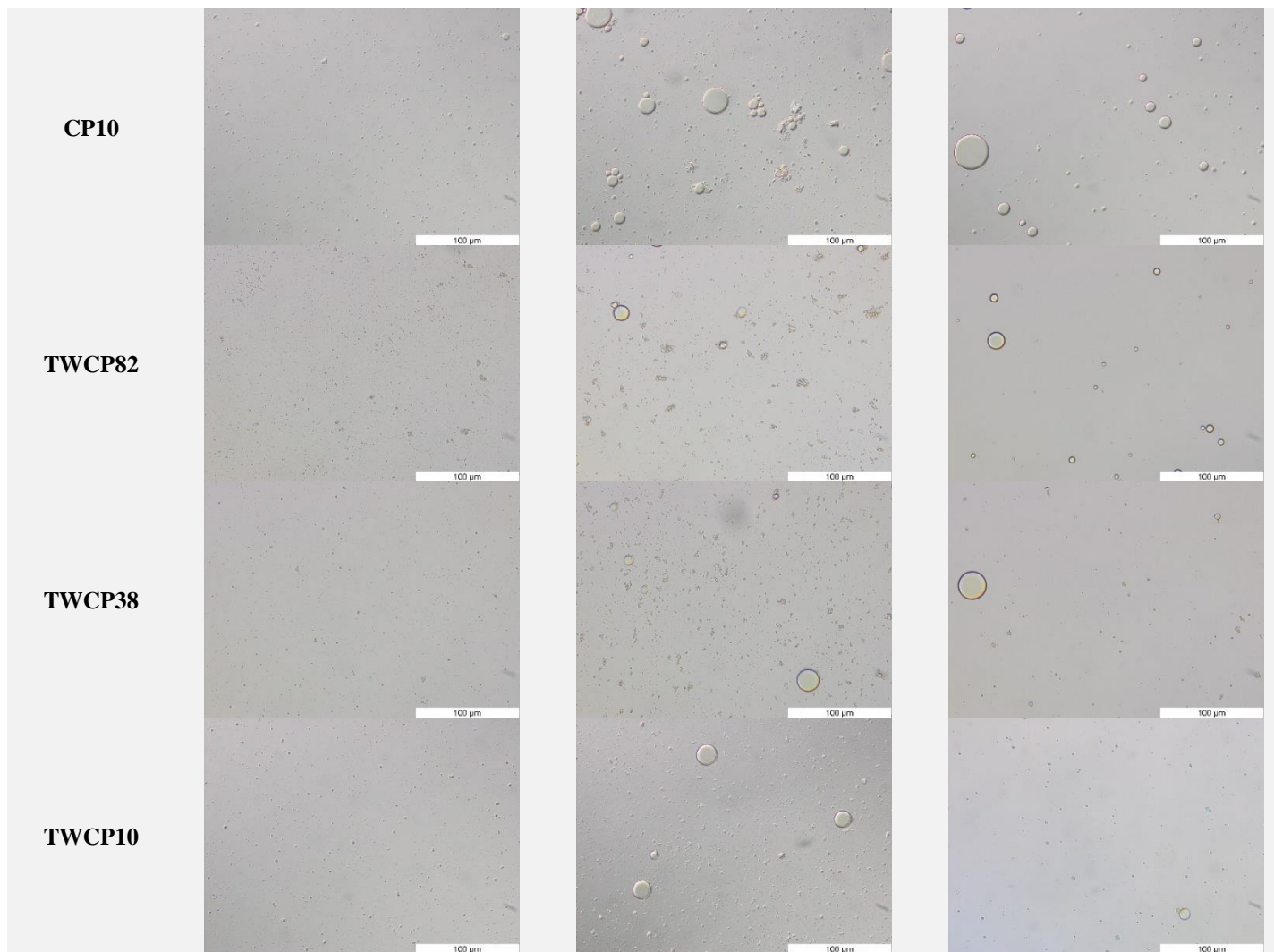
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**Table A (supplementary material): Changes on the particle size of the initial emulsion, chyme and digest represented by the median volume-weighted particle size. Different letters in the same column indicate significant differences (Tukey test:  $p < 0.05$ ) among the samples. For interpretation of the abbreviations is referred to the abbreviation list.**

	Emulsion		Gastric phase		Small intestinal phase	
TW	1.27 ± 0.10	E	1.29 ± 0.02	E	2.89 ± 0.19	G
PC	1.65 ± 0.01	D	12.71 ± 5.68	A	21.22 ± 4.80	A
CP82	1.42 ± 0.02	B	6.75 ± 0.89	BCD	6.48 ± 1.04	F
CP38	1.67 ± 0.08	A	8.30 ± 0.10	BC	9.99 ± 0.32	DE
CP10	1.12 ± 0.01	F	1.96 ± 0.30	E	11.90 ± 1.71	CD
TWCP82	1.14 ± 0.01	F	1.86 ± 0.08	E	6.70 ± 0.51	F
TWCP38	1.06 ± 0.05	FG	4.33 ± 0.52	CDE	8.01 ± 1.11	EF
TWCP10	1.82 ± 0.07	C	9.78 ± 3.34	AB	17.62 ± 1.44	B
PCCP82	1.30 ± 0.01	E	3.19 ± 0.03	DE	13.52 ± 0.51	C
PCCP38	1.13 ± 0.02	F	2.13 ± 0.01	E	12.11 ± 0.28	CD
PCCP10	1.01 ± 0.03	G	1.41 ± 0.01	E	19.30 ± 0.99	AB

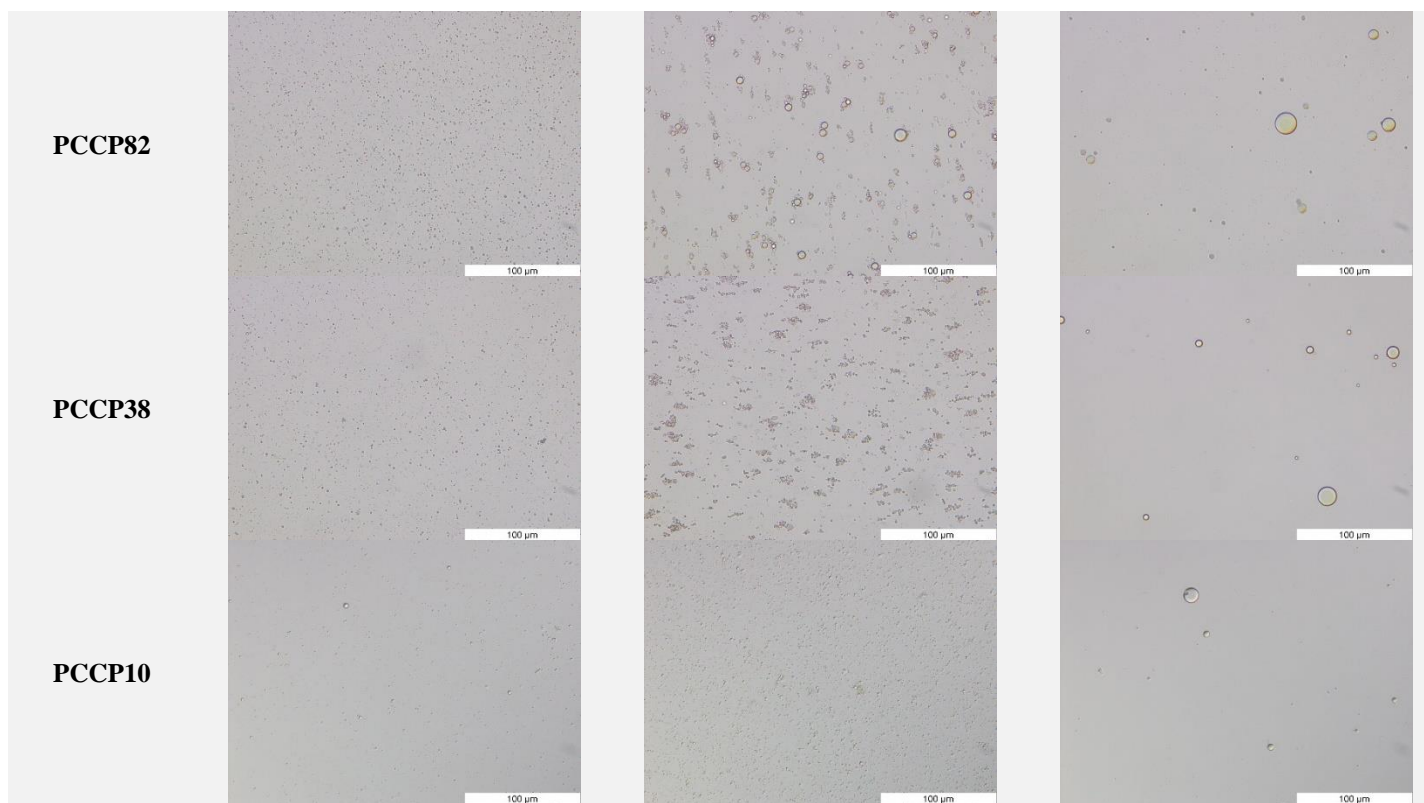
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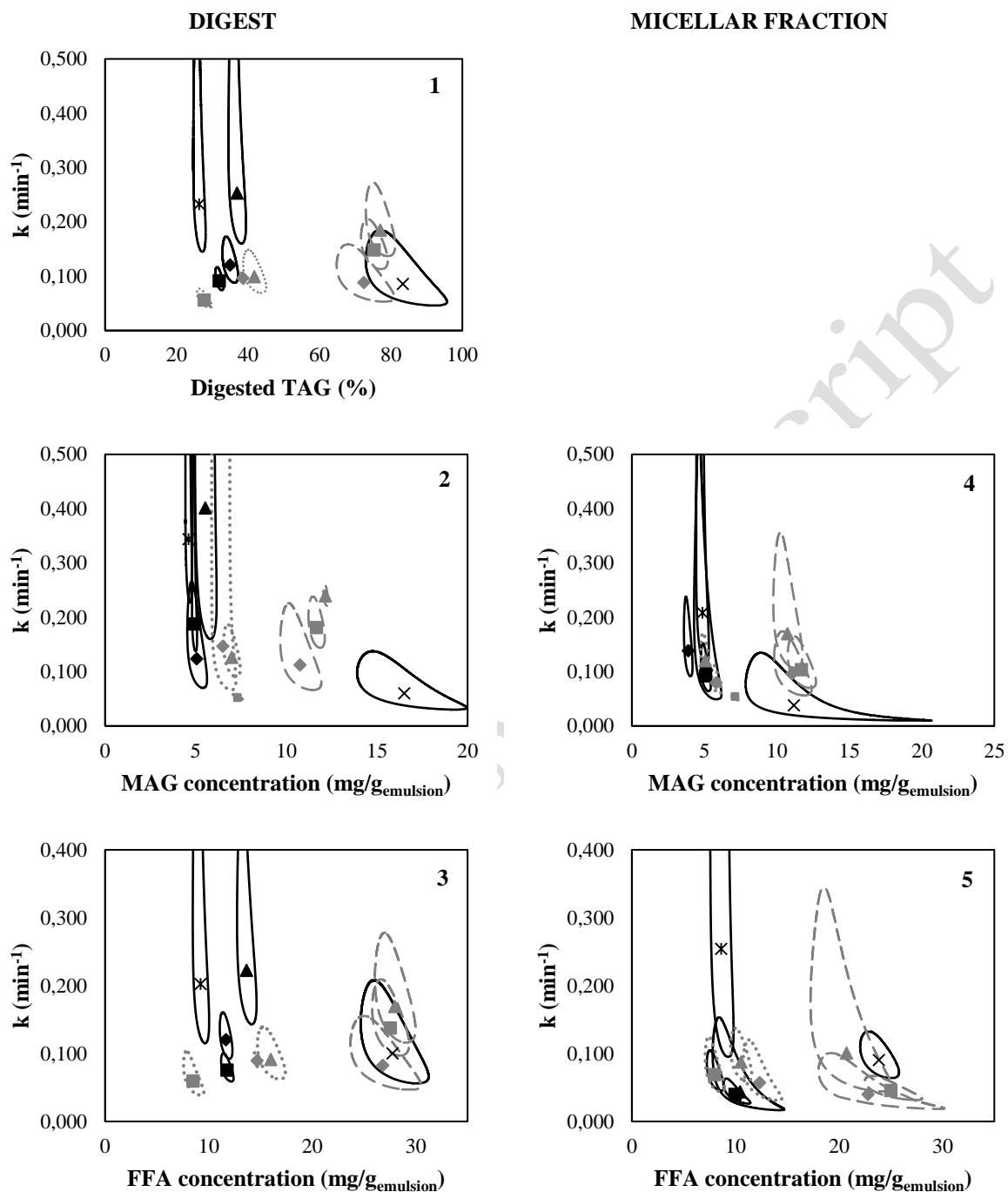
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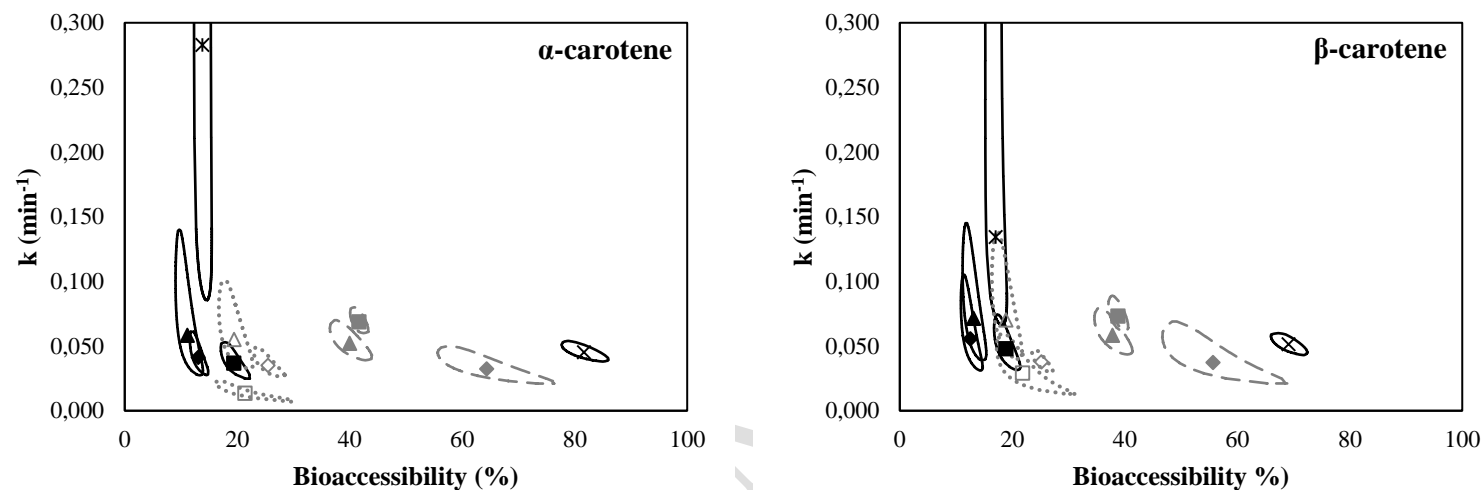


**Figure B: Changes in microstructure of all carrot-based oil-in-water emulsions during digestion (scale bar represents a length of 100 μm). For interpretation of the abbreviations is referred to the abbreviation list.**

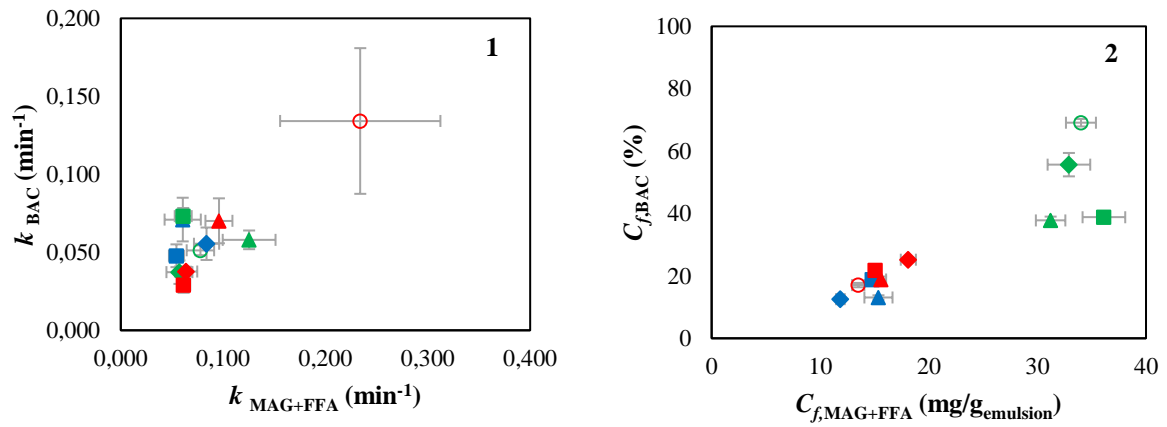
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**Figure C:** Joint confidence region analysis (90%) of the jointly estimated parameters (Table 2) for *in vitro* small intestinal lipid digestion: (1) TAG in the digest; (2) MAG in the digest; (3) FFA in the digest; (4) MAG in the micellar fraction and (5) FFA in the micellar fraction. The symbols indicate the estimated value of the corresponding fractional conversion model, while the lines show the joint confidence region. Full lines represent mono-emulsifier emulsions (× TW; \* PC; ■ CP82; ◆ CP38 and ▲ CP10); dashed lines represent tween-pectin combined emulsions (■ TWCP82; ◆ TWCP38 and ▲ TWCP10) and dotted lines represent phosphatidylcholine-tween combined emulsions (■ PCCP82; ◆ PCCP38 and ▲ PCCP10). For interpretation of the abbreviations is referred to the abbreviation list.



**Figure D: Joint confidence region analysis (90%) of the jointly estimated parameters (Table 3) for carotenoid bioaccessibility of  $\alpha$ - and  $\beta$ -carotene. The symbols indicate the experimental data, while the lines show the joint confidence region. Full lines represent mono-emulsifier emulsions ( $\times$  TW;  $*$  PC;  $\blacksquare$  CP82;  $\blacklozenge$  CP38 and  $\blacktriangle$  CP10); dashed lines represent the tween-pectin emulsions ( $\blacksquare$  TWCP82;  $\blacklozenge$  TWCP38 and  $\blacktriangle$  TWCP10) and dotted lines represent the phosphatidylcholine-pectin emulsions ( $\blacksquare$  PCCP82;  $\blacklozenge$  PCCP38 and  $\blacktriangle$  PCCP10).**



**Figure E: Relation between the (1)  $k$ -values (min<sup>-1</sup>) and (2)  $C_f$ -values of MAG and FFA micellarisation, and carotenoid incorporation into mixed micelles (○ TW; ○ PC; ■ CP82; ◆ CP38; ▲ CP10; ■ TWCP82; ◆ TWCP38; ▲ TWCP10; ■ PCCP82; ◆ PCCP38; ▲ PCCP10). For interpretation of the abbreviations is referred to the abbreviation list.**

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