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Unravelling the structure of serum pectin originating from thermally and mechanically processed carrot-based suspensions

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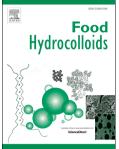
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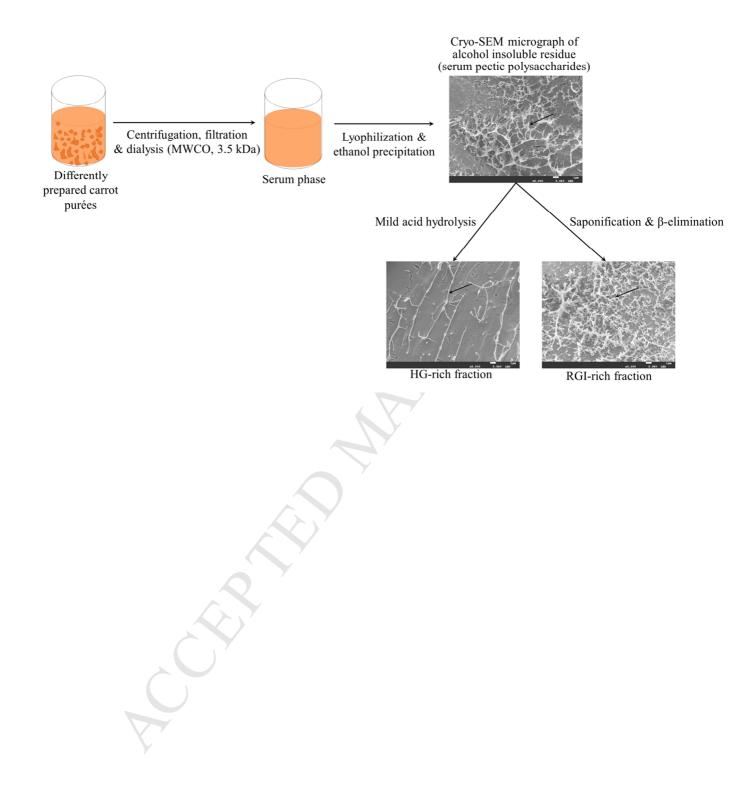
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Highlights

- Serum pectin subdomains were isolated using hot acid and alkaline conditions.
- The AIR and derived RGI-rich fraction exhibited branch-like microstructures.
- Homogalacturonan-rich fractions displayed strand-like microstructures.

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21 Abstract

The molecular structure of pectin, a cell wall polysaccharide in fruits and vegetables, 22 greatly determines its functionality. Owing to its solubility characteristics, pectin is naturally 23 present in the particle and liquid (serum) phase of plant-based dispersions (e.g. purées) that 24 can influence product functionality. The objective of the present study was to investigate the 25 structure of serum pectin obtained from carrot purées, differently processed by a combination 26 of mechanical tissue disintegration (blending and high pressure homogenization) and heat 27 treatments (high and low temperature). Serum phases, isolated from the purées, were 28 lyophilized and served as the starting material for isolation of the pectic polysaccharides as 29 alcohol insoluble residues. Physico-chemical characterization revealed that different 30 processing combinations of carrot purées can generate serum pectins with specific structural 31 properties. Differences in the molecular weight of carrot serum pectic polysaccharides were 32 observed depending on the sequence of heat and mechanical treatments. Subsequent 33 34 partitioning of carrot serum pectic polysaccharides into rhamnogalacturonan I (RGI) and homogalacturonan (HG)-rich fractions using hot alkaline and mild acid conditions, 35 respectively, revealed the composition and molecular characteristics of the serum pectin 36 subdomains. The RGI-rich fractions were mainly characterized by high concentrations of 37 neutral sugars (arabinose, galactose, and rhamnose) and exhibited branch-like structures in 38 solution under cryo-SEM. HG-rich fractions, on the other hand, were characterized by highly 39 linear galacturonic acid-rich pectin and exhibited strand-like structures under cryo-SEM. 40 Knowledge of these structural changes can be useful in exploring the functionalities of carrot 41 42 serum pectic polysaccharides.

43 Keywords

- 44 Carrot purée, serum pectin, homogalacturonan-rich fraction, rhamnogalacturonan I-rich
- 45 fraction, microstructure

46 **1. Introduction**

47 The production of fruit and vegetable-based dispersions such as soups, smoothies, sauces and purées involves tissue disintegration and heat treatments, which alters the 48 microstructure of the plant matrix resulting in a complex multi-scale food system structured 49 by particles dispersed in a continuous (serum) phase (Moelants, et al., 2014). Pectin, due to its 50 solubility characteristics, is present in both the particle and serum phase, and can exhibit 51 different structural properties. Moreover, changes in the molecular structure of pectin in both 52 phases can occur during processing and influence the functionality (e.g. flow behavior) of 53 low-starch containing plant-based dispersions (Christiaens et al., 2012; Houben et al., 2014). 54 Pectin solubilized in the serum phase, a water-soluble pectin termed as serum pectin, is 55 suggested to be an important constituent of plant dispersions. Depending on its molecular 56 characteristics, serum pectin is suggested to interact within the aqueous phase or with the 57 58 particles and contribute to the overall structure organization of plant-based dispersions (Kyomugasho, Willemsen, Christiaens, Van Loey, & Hendrickx, 2015; Moelants et al., 2014; 59 Moelants et al., 2012). However, in-depth knowledge and understanding of the specific 60 molecular characteristics of serum pectin naturally present in plant-based food dispersions is 61 still limited. An insight into the molecular properties of serum pectin and how these properties 62 change under specific processing conditions might prompt the exploration of serum pectin's 63 potential as a functional component. 64

To date, pectin is a well-recognized biopolymer comprising several polysaccharide
domains with at least 17 different monosaccharides interconnected through more than 20
different linkages (Ridley, O'Neill, & Mohnen, 2001). Homogalacturonan (HG) and
rhamnogalacturonans (mainly rhamnogalacturonans I and II) are the most abundant pectic
polysaccharides. On the one hand, HG is a linear chain of 1,4 linked α-D-galacturonic acid
(GalA) residues known as the "smooth" region. Its GalA residues can be methyl-esterified at

C-6 up to 70-80 % and O-acetylated at C-3 or C-2 depending on the plant source (Voragen, 71 Coenen, Verhoef, & Schols, 2009). On the other hand, rhamnogalacturonan I (RGI) consists 72 of a backbone of the repeating disaccharide $[-\alpha$ -D-GalA-1,2- α -L-Rha-1-4-]n and represents 73 20-35 % of pectin. RGI, the "hairy" region, is ramified with side chains of individual, linear 74 or branched oligosaccharide residues attached at the C-4 of rhamnose residues (Voragen et al., 75 2009). Linear arabinan and (arabino)galactan are the predominant RGI side chains (Caffall & 76 Mohnen, 2009). Recently, the galacturonic acid and rhamnose residues of the RGI subdomain 77 of pectin extracted from okra pods were found to be O-acetylated (Alba, Laws, & 78 Kontogiorgos, 2015; Sengkhamparn et al., 2009). The third substructure, rhamnogalacturonan 79 II (RGII) accounts for about 10 % of total pectin and, is the most complex and conserved 80 structure among the pectic polysaccharides. This substructure comprises of a backbone 81 consisting of at least eight GalA residues with side branches of either structurally distinct 82 83 disaccharides or oligosaccharides (Caffall et al., 2009; Mohnen, 2008). The structure of individual pectin subdomains can be revealed by partitioning pectin 84 chemically and/or enzymatically followed by a physico-chemical characterization. In this 85 regard, Thibault, Renard, Axelos, Roger, & Crepeau (1993) isolated the homogalacturonan 86 fraction of apple, beet and citrus pectins using mild acid hydrolysis at 80 °C for 72 h and its 87 characterization revealed different sensitivities of the glycosidic linkages of these pectins to 88 acid hydrolysis. Recently, mild acid hydrolysis at 80 °C for 24 h was shown to be sufficient in 89 isolating the homogalacturonan fraction of different citrus pectins (Kaya, Sousa, Crepeau, 90 Sorensen, & Ralet, 2014). Similarly, the homogalacturonan subdomain of tomato pectin from 91 unripe pericarp was selectively isolated using mild acid hydrolysis at 80 °C for 24 h and then 92 investigated using atomic force microscopy (Round, Rigby, MacDougall, & Morris, 2010). 93 Moreover, to characterize the molecular properties of sugar beet pectin (SBP), Morris, Ralet, 94 Bonnin, Thibault, & Harding (2010) selectively isolated the homogalacturonan and 95

rhamnogalacturonan I subdomains by different means. On the one hand, the 96 97 homogalacturonan subdomain of SBP was isolated using mild acid hydrolysis at 80 °C for 72 h, while on the other hand, the rhamnogalacturonan subdomain was obtained by 98 enzymatically degrading the homogalacturonan using fungal pectin methyl-esterase and 99 different polygalacturonases (Morris et al., 2010). Given that mild acid hydrolysis for 72 h 100 can degrade and influence the molecular weight of homogalacturonan, a combination of 101 different enzymes including pectin methylesterase, rhamnogalacturonan hydrolase, 102 galactanase and arabinanase were used to isolate the homogalacturonan of sugar beet, apple 103 and lime pectins (Bonnin, Dolo, Le Goff, & Thibault, 2002). However, fragments of 104 rhamnogalacturonan side chains were still present in the enzymatically isolated 105 homogalacturonan fraction (Bonnin et al., 2002). Therefore, using either chemical or 106 enzymatic means of partitioning the pectin subdomains might often be associated with 107 108 residual amounts of one fraction in another.

As a first step in gaining insight into the molecular properties of serum (solubilized) 109 pectin, the chemical means of partitioning the pectin subdomains was used in the present 110 study. Serum pectin polysaccharides inherently present in thermally and mechanically 111 processed fruits and vegetables (e.g. carrot purée) can be structurally more complex than 112 commercially-prepared, purified or chemically extracted pectins. Different serum pectin 113 molecular structures can be obtained, depending on the sequence of thermal and mechanical 114 treatments or the action of endogenous pectin-related enzymes (e.g. pectin methylesterase) 115 (Santiago, Christiaens, Van Loey, & Hendrickx, 2016; Santiago, Jamsazzadeh Kermani, Xu, 116 Van Loey, & Hendrickx, 2017). Therefore, the present work aimed to gain an in-depth 117 understanding of the molecular structure of serum pectin polysaccharides, isolated from 118 differently processed carrot purées, by selectively partitioning the homogalacturonan and 119 rhamnogalacturonan subdomains. These subdomains were subsequently characterized for 120

121	their degree of methyl-esterification, molecular weight and neutral sugar composition. On the
122	one hand, mild acid hydrolysis at 80 °C for 24 h was applied to obtain the homogalacturonan
123	(HG) subdomain while on the other hand, a hot alkaline condition (90 $^{\circ}$ C for 2 h) was
124	employed to isolate the rhamnogalacturonan I subdomain. To visualize the microstructure of
125	serum pectin polysaccharides (as alcohol insoluble residue) and the partitioned carrot serum
126	pectin subdomains in solution, cryo-scanning electron microscopy (cryo-SEM) was used.
127	2. Materials and methods
128	2.1 Materials
129	A batch of fresh carrots (Daucus carota cv. Nerac) from a local shop was washed,
130	peeled and cut into slices of approximately 0.5 cm thick. All chemical reagents used were of
131	analytical grade.
132	2.2 Preparation of carrot purées and isolation of the serum phases
133	A schematic overview of the purée preparation is provided in Figure 1. The
134	combinations of thermal and mechanical treatments, applied in producing three differently-
135	prepared purées, were aimed at creating distinct pectin molecular structures. High temperature
136	treatment (HT) at 95 °C for 30 min was applied to inactivate the enzymes and/or solubilize
137	pectin into the serum phase (Houben et al., 2014; Moelants et al., 2012), while low
138	temperature treatment (LT) at 60 $^{\circ}$ C for 40 min was aimed at stimulating the action of
139	endogenous pectin methyl-esterase (PME) to generate low methyl-esterified pectin
140	(Christiaens et al., 2012; Santiago et al., 2016). High pressure homogenization (HPH) at 100
141	MPa was applied to enhance pectin solubilization into the serum phase of the purées
142	(Moelants et al., 2012).
143	Briefly, carrot slices were vacuum-packed in polyethylene bags (DaklaPack®
144	Lamigrip Stand-up Pouch Transparent; 220 mm \times 300 mm + 65 mm bottom fold) and then

either heat treated at 95 °C for 30 min or left untreated (raw). De-mineralized water in a 1:1

(w/w) ratio was then added to the raw or heat treated carrot slices followed by mechanical 146 disruption for 20 s at low speed then 40 s at high speed using a kitchen blender (Waring 147 Commercial, Torrington, Connecticut, USA). Afterwards, a high pressure homogenizer 148 (Panda 2 K, Gea Niro, Parma, Italy) was used at 100 MPa (single pass) to intensively disrupt 149 the tissues. The purée obtained from the raw carrot slices was immediately vacuum-packed in 150 polyethylene bags and either immediately subjected to HT at 95 °C for 30 min to inactivate 151 pectin-modifying enzymes or first subjected to LT at 60 °C for 40 min to activate PME prior 152 to an inactivation step (HT) as shown in Figure 1. After HT, samples were cooled to ambient 153 temperature in an ice-water bath. In order to obtain and isolate the serum phase from the 154 particles, the purées were centrifuged at 12,400 x g for 30 min at 20 °C (J2-HS centrifuge, 155 Beckman, CA, USA). The serum phase, the supernatant, was vacuum-filtered, extensively 156 dialyzed (3.5 kDa, MWCO) against de-mineralized water, lyophilized using Christ alpha 2-4 157 158 freeze dryer (Osterode, Germany) and then stored in a desiccator over P₂O₅ until further use. Hereto, the dry material obtained after freeze-drying the serum is referred as the lyophilized 159 serum and served as the starting material for isolation of the carrot serum pectic 160 polysaccharides, as alcohol insoluble residue. 161 2.3 Isolation of carrot serum pectic polysaccharides as alcohol insoluble residue (AIR) 162 Approximately 1.2 g of lyophilized serum, from each of the differently-prepared carrot 163 purées, was suspended in 600 mL of 95 % (ν/ν) ethanol and mixed for 30 min at 4 °C. The 164 resulting suspension was vacuum-filtered (Machery-Nagel MN 615, Ø 90 mm) and re-165 suspended in 300 mL of 95 % (ν/ν) ethanol. After a second vacuum filtration step, the alcohol 166 insoluble residue (AIR) was suspended and stirred in acetone. A final vacuum filtration 167 generated the AIR which was dried overnight at 40 °C, thoroughly mixed and stored in a 168

169 desiccator over P_2O_5 until further analysis.

170 2.3.1 Isolation of the rhamnogalacturonan I (RGI)-rich fraction from AIR using hot alkaline
171 conditions

AIRs of lyophilized sera, from differently-prepared carrot purées, were subjected to 172 alkaline pH and high temperature conditions. Approximately 60 mg AIR was suspended in 12 173 mL de-mineralized water and stirred at 4 °C for 16 h. Subsequently, the samples were heated 174 at 90 °C for 30 min and hot sodium hydroxide (1 M) was gradually added until pH 12 (pH 175 meter Lab 860, Schott Instruments Analytics GmbH, Germany). The samples were kept at 90 176 °C for 2 h with mild stirring and afterwards cooled to ambient temperature. Subsequently, 177 ethanol was added until a final concentration of 80 % (v/v) and the suspension was vacuum-178 filtered using 0.1 µm (Durapore® membrane filters, Merck Millipore ltd. Ireland) glass unit 179 filter. The residue/precipitate was dissolved in 10 mL de-mineralized water, neutralized to pH 180 6 using 0.1 M hydrochloric acid and consequently dialyzed (MWCO, 3.5 kDa) against de-181 182 mineralized water for 48 h. Finally, the suspensions/RGI-rich fractions were lyophilized and then stored in a desiccator over P₂O₅ until further analysis. 183

184 2.3.2 Isolation of the homogalacturonan (HG)-rich fraction from de-esterified AIR using mild
185 acid hydrolysis

AIRs were first subjected to chemical de-esterification and then acid hydrolysis based 186 on the work of Morris et al. (2010) and Thibault et al. (1993) with minor modifications. 187 De-esterification/saponification of serum pectic polysaccharides was aimed at increasing the 188 resistance of the linkages of galacturonic acid residues to hydrolysis under the high 189 temperature and acid conditions (Thibault et al., 1993). To this extent, approximately 60 mg 190 AIR was suspended in 12 mL de-mineralized water and stirred at 4 °C for 16 h. Cold sodium 191 hydroxide (1 M) was slowly added to the mixture until a pH of 12 was obtained and then 192 maintained for 6 h at 4 °C. After chemical de-esterification, the pH was adjusted to 1.0 using 193 hydrochloric acid (1 M) and the samples were subsequently heated in a temperature-194

controlled oil bath at 80 °C for 24 h. After cooling the samples to ambient temperature, 195 196 ethanol was added until a final concentration of 80 % (ν/ν). The suspension was subsequently vacuum-filtered using a 0.1 µm (Durapore® membrane filters, Merck Millipore ltd. Ireland) 197 glass unit filter. About 3 mL of 0.1 M HCl was gradually added to the residues obtained 198 followed by 10 mL de-mineralized water. Finally, the samples/HG-rich fractions were 199 neutralized to pH 6 using 0.5 M lithium hydroxide, extensively dialyzed (MWCO, 3.5 kDa) 200 201 against de-mineralized water and lyophilized prior to being stored in a desiccator over P_2O_5 until further. 202

203 2.4 Analysis of the physico-chemical characteristics of carrot serum pectic polysaccharides
204 and partitioned serum pectin subdomains

205 2.4.1 Monosaccharide composition

Monosaccharides including neutral sugars, galacturonic and glucuronic acids were 206 207 determined as described by De Ruiter, Schols, Voragen & Rombouts (1992) and Nagel, Sirisakulwat, Carle, & Neidhart (2014) with minor modifications. Methanolysis followed by 208 acid hydrolysis of each sample was carried out to obtain and analyze the constituting 209 monosaccharides. The resulting monosaccharides were identified and quantified using high 210 performance anion exchange chromatography (HPAEC) combined with pulsed amperometric 211 detector (PAD). Briefly, 1 mg sample was dissolved in 1 mL de-mineralized water and an 212 aliquot of 20 µL was dried in a screw cap test-tube under a N₂ evaporator at 45 °C. After 213 adding 0.5 mL anhydrous 2 M methanolic HCl, methanolysis was carried out at 80 °C for 16 214 h in an oil bath. The samples were then cooled to ambient temperature and dried under N₂ at 215 30 °C. Subsequently, acid hydrolysis using 0.5 mL of 2 M trifluoroacetic acid (TFA) was 216 performed for 1 h at 121 °C. The acid hydrolyzed samples were cooled to ambient 217 temperature, dried under N2 evaporator at 45 °C to remove the TFA and then dissolved in 218 ultrapure water (organic free, 18 M Ω cm resistance) to a final concentration of 0.002 % (w/v). 219

 standards (L-fucose, L rhamnose, L-arabinose, D-galactose, D-glucose, D-mannose, D-xylose, galacturonic acid and glucuronic acid) with known concentrations were subjected to methanolysis and TFA hydrolysis. All the samples were filtered through 0.45 µm (Chromafil A-45/25, 0.45 µm, Macherey-Nagel Gmbh, Duren, Germany) prior to injection onto HPAEC. A Dionex HPLC system (DX600), equipped with a GS50 gradient pump, a CarboPacTM PA20 column (150 x 3 mm, pH range = 0-14), a CarboPacTM PA20 guard column (30 x 3 mm), and an ED50 electrochemical detector (Dionex, Sunnyvale, USA) was used. The detector was equipped with a reference pH electrode (Ag/AgCl) and a gold electrode that was used in the PAD mode, performing a quadruple potential waveform. The elution gradients were adapted from the method described by (Arnous & Meyer, 2008). After equilibration (-10 → -5 min: 100 mM NaOH, -5 → 0 min; elution gradient), 10 µL hydrolysate was injected and eluted at 30 °C with a flow rate of 0.5 mL/min. From 0 to 20 min, elution gradients of 0.5 mM NaOH to obtain better peak resolution of xylose and mannose or 15 mM NaOH to separate the peaks of rhamnose and arabinose were applied (Jamsazzadeh Kermani et al., 2014). The column was regenerated by using 500 mM NaOH (20 → 30 min). To quantify the monosaccharides, mixtures of un-hydrolyzed sugar standards were used at varying concentrations (<i>i.e.</i> 5-25 ppm galacturonic acid and 1-10 ppm for neutral sugars and glucuronic acid). Peak areas of un-hydrolyzed and hydrolyzed sugar standards were compared and the recovery values were taken into account in quantifying the monosaccharides. All hydrolysis and chromatographic measurements were carried out in duplicate. 	220	To correct the possible degradation of monosaccharides, 100 μ L aliquots of a mixture of sugar
to methanolysis and TFA hydrolysis. All the samples were filtered through 0.45 μ m (Chromafil A-45/25, 0.45 μ m, Macherey-Nagel Gmbh, Duren, Germany) prior to injection onto HPAEC. A Dionex HPLC system (DX600), equipped with a GS50 gradient pump, a CarboPac TM PA20 column (150 x 3 mm, pH range = 0-14), a CarboPac TM PA20 guard column (30 x 3 mm), and an ED50 electrochemical detector (Dionex, Sunnyvale, USA) was used. The detector was equipped with a reference pH electrode (Ag/AgCl) and a gold electrode that was used in the PAD mode, performing a quadruple potential waveform. The elution gradients were adapted from the method described by (Arnous & Meyer, 2008). After equilibration (-10 \rightarrow -5 min: 100 mM NaOH, -5 \rightarrow 0 min; elution gradient), 10 μ L hydrolysate was injected and eluted at 30 °C with a flow rate of 0.5 mL/min. From 0 to 20 min, elution gradients of 0.5 mM NaOH to obtain better peak resolution of xylose and mannose or 15 mM NaOH to separate the peaks of rhamnose and arabinose were applied (Jamsazzadeh Kermani et al., 2014). The column was regenerated by using 500 mM NaOH (20 \rightarrow 30 min). To quantify the monosaccharides, mixtures of un-hydrolyzed sugar standards were used at varying concentrations (<i>i.e.</i> 5-25 ppm galacturonic acid and 1-10 ppm for neutral sugars and glucuronic acid). Peak areas of un-hydrolyzed and hydrolyzed sugar standards were compared and the recovery values were taken into account in quantifying the monosaccharides. All hydrolysis and chromatographic measurements were carried out in	221	standards (L-fucose, L rhamnose, L-arabinose, D-galactose, D-glucose, D-mannose,
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used. The detector was equipped with a reference pH electrode (Ag/AgCl) and a gold electrode that was used in the PAD mode, performing a quadruple potential waveform. The elution gradients were adapted from the method described by (Arnous & Meyer, 2008). After equilibration $(-10 \rightarrow -5 \text{ min: } 100 \text{ mM NaOH}, -5 \rightarrow 0 \text{ min: elution gradient}), 10 \ \mu\text{L}$ hydrolysate was injected and eluted at 30 °C with a flow rate of 0.5 mL/min. From 0 to 20 min, elution gradients of 0.5 mM NaOH to obtain better peak resolution of xylose and mannose or 15 mM NaOH to separate the peaks of rhamnose and arabinose were applied (Jamsazzadeh Kermani et al., 2014). The column was regenerated by using 500 mM NaOH $(20 \rightarrow 30 \text{ min})$. To quantify the monosaccharides, mixtures of un-hydrolyzed sugar standards were used at varying concentrations (<i>i.e.</i> 5-25 ppm galacturonic acid and 1-10 ppm for neutral sugars and glucuronic acid). Peak areas of un-hydrolyzed and hydrolyzed sugar standards were compared and the recovery values were taken into account in quantifying the monosaccharides. All hydrolysis and chromatographic measurements were carried out in	226	CarboPac TM PA20 column (150 x 3 mm, pH range = 0-14), a CarboPac TM PA20 guard
electrode that was used in the PAD mode, performing a quadruple potential waveform. The elution gradients were adapted from the method described by (Arnous & Meyer, 2008). After equilibration (-10 \rightarrow -5 min: 100 mM NaOH, -5 \rightarrow 0 min: elution gradient), 10 µL hydrolysate was injected and eluted at 30 °C with a flow rate of 0.5 mL/min. From 0 to 20 min, elution gradients of 0.5 mM NaOH to obtain better peak resolution of xylose and mannose or 15 mM NaOH to separate the peaks of rhamnose and arabinose were applied (Jamsazzadeh Kermani et al., 2014). The column was regenerated by using 500 mM NaOH (20 \rightarrow 30 min). To quantify the monosaccharides, mixtures of un-hydrolyzed sugar standards were used at varying concentrations (<i>i.e.</i> 5-25 ppm galacturonic acid and 1-10 ppm for neutral sugars and glucuronic acid). Peak areas of un-hydrolyzed and hydrolyzed sugar standards were compared and the recovery values were taken into account in quantifying the monosaccharides. All hydrolysis and chromatographic measurements were carried out in	227	column (30 x 3 mm), and an ED50 electrochemical detector (Dionex, Sunnyvale, USA) was
elution gradients were adapted from the method described by (Arnous & Meyer, 2008). After equilibration (-10 \rightarrow -5 min: 100 mM NaOH, -5 \rightarrow 0 min: elution gradient), 10 µL hydrolysate was injected and eluted at 30 °C with a flow rate of 0.5 mL/min. From 0 to 20 min, elution gradients of 0.5 mM NaOH to obtain better peak resolution of xylose and mannose or 15 mM NaOH to separate the peaks of rhamnose and arabinose were applied (Jamsazzadeh Kermani et al., 2014). The column was regenerated by using 500 mM NaOH (20 \rightarrow 30 min). To quantify the monosaccharides, mixtures of un-hydrolyzed sugar standards were used at varying concentrations (<i>i.e.</i> 5-25 ppm galacturonic acid and 1-10 ppm for neutral sugars and glucuronic acid). Peak areas of un-hydrolyzed and hydrolyzed sugar standards were compared and the recovery values were taken into account in quantifying the monosaccharides. All hydrolysis and chromatographic measurements were carried out in	228	used. The detector was equipped with a reference pH electrode (Ag/AgCl) and a gold
equilibration (-10 \rightarrow -5 min: 100 mM NaOH, -5 \rightarrow 0 min: elution gradient), 10 µL hydrolysate was injected and eluted at 30 °C with a flow rate of 0.5 mL/min. From 0 to 20 min, elution gradients of 0.5 mM NaOH to obtain better peak resolution of xylose and mannose or 15 mM NaOH to separate the peaks of rhamnose and arabinose were applied (Jamsazzadeh Kermani et al., 2014). The column was regenerated by using 500 mM NaOH (20 \rightarrow 30 min). To quantify the monosaccharides, mixtures of un-hydrolyzed sugar standards were used at varying concentrations (<i>i.e.</i> 5-25 ppm galacturonic acid and 1-10 ppm for neutral sugars and glucuronic acid). Peak areas of un-hydrolyzed and hydrolyzed sugar standards were compared and the recovery values were taken into account in quantifying the monosaccharides. All hydrolysis and chromatographic measurements were carried out in	229	electrode that was used in the PAD mode, performing a quadruple potential waveform. The
hydrolysate was injected and eluted at 30 °C with a flow rate of 0.5 mL/min. From 0 to 20 min, elution gradients of 0.5 mM NaOH to obtain better peak resolution of xylose and mannose or 15 mM NaOH to separate the peaks of rhamnose and arabinose were applied (Jamsazzadeh Kermani et al., 2014). The column was regenerated by using 500 mM NaOH ($20 \rightarrow 30$ min). To quantify the monosaccharides, mixtures of un-hydrolyzed sugar standards were used at varying concentrations (<i>i.e.</i> 5-25 ppm galacturonic acid and 1-10 ppm for neutral sugars and glucuronic acid). Peak areas of un-hydrolyzed and hydrolyzed sugar standards were compared and the recovery values were taken into account in quantifying the monosaccharides. All hydrolysis and chromatographic measurements were carried out in	230	elution gradients were adapted from the method described by (Arnous & Meyer, 2008). After
20 min, elution gradients of 0.5 mM NaOH to obtain better peak resolution of xylose and mannose or 15 mM NaOH to separate the peaks of rhamnose and arabinose were applied (Jamsazzadeh Kermani et al., 2014). The column was regenerated by using 500 mM NaOH ($20 \rightarrow 30$ min). To quantify the monosaccharides, mixtures of un-hydrolyzed sugar standards were used at varying concentrations (<i>i.e.</i> 5-25 ppm galacturonic acid and 1-10 ppm for neutral sugars and glucuronic acid). Peak areas of un-hydrolyzed and hydrolyzed sugar standards were compared and the recovery values were taken into account in quantifying the monosaccharides. All hydrolysis and chromatographic measurements were carried out in	231	equilibration (-10 \rightarrow -5 min: 100 mM NaOH, -5 \rightarrow 0 min: elution gradient), 10 µL
mannose or 15 mM NaOH to separate the peaks of rhamnose and arabinose were applied (Jamsazzadeh Kermani et al., 2014). The column was regenerated by using 500 mM NaOH ($20 \rightarrow 30$ min). To quantify the monosaccharides, mixtures of un-hydrolyzed sugar standards were used at varying concentrations (<i>i.e.</i> 5-25 ppm galacturonic acid and 1-10 ppm for neutral sugars and glucuronic acid). Peak areas of un-hydrolyzed and hydrolyzed sugar standards were compared and the recovery values were taken into account in quantifying the monosaccharides. All hydrolysis and chromatographic measurements were carried out in	232	hydrolysate was injected and eluted at 30 °C with a flow rate of 0.5 mL/min. From 0 to
(Jamsazzadeh Kermani et al., 2014). The column was regenerated by using 500 mM NaOH ($20 \rightarrow 30$ min). To quantify the monosaccharides, mixtures of un-hydrolyzed sugar standards were used at varying concentrations (<i>i.e.</i> 5-25 ppm galacturonic acid and 1-10 ppm for neutral sugars and glucuronic acid). Peak areas of un-hydrolyzed and hydrolyzed sugar standards were compared and the recovery values were taken into account in quantifying the monosaccharides. All hydrolysis and chromatographic measurements were carried out in	233	20 min, elution gradients of 0.5 mM NaOH to obtain better peak resolution of xylose and
$(20 \rightarrow 30 \text{ min}). \text{ To quantify the monosaccharides, mixtures of un-hydrolyzed sugar standards}$ were used at varying concentrations (<i>i.e.</i> 5-25 ppm galacturonic acid and 1-10 ppm for neutral sugars and glucuronic acid). Peak areas of un-hydrolyzed and hydrolyzed sugar standards were compared and the recovery values were taken into account in quantifying the monosaccharides. All hydrolysis and chromatographic measurements were carried out in	234	mannose or 15 mM NaOH to separate the peaks of rhamnose and arabinose were applied
 were used at varying concentrations (<i>i.e.</i> 5-25 ppm galacturonic acid and 1-10 ppm for neutral sugars and glucuronic acid). Peak areas of un-hydrolyzed and hydrolyzed sugar standards were compared and the recovery values were taken into account in quantifying the monosaccharides. All hydrolysis and chromatographic measurements were carried out in 	235	(Jamsazzadeh Kermani et al., 2014). The column was regenerated by using 500 mM NaOH
 sugars and glucuronic acid). Peak areas of un-hydrolyzed and hydrolyzed sugar standards were compared and the recovery values were taken into account in quantifying the monosaccharides. All hydrolysis and chromatographic measurements were carried out in 	236	$(20 \rightarrow 30 \text{ min})$. To quantify the monosaccharides, mixtures of un-hydrolyzed sugar standards
 were compared and the recovery values were taken into account in quantifying the monosaccharides. All hydrolysis and chromatographic measurements were carried out in 	237	were used at varying concentrations (i.e. 5-25 ppm galacturonic acid and 1-10 ppm for neutral
240 monosaccharides. All hydrolysis and chromatographic measurements were carried out in	238	sugars and glucuronic acid). Peak areas of un-hydrolyzed and hydrolyzed sugar standards
	239	were compared and the recovery values were taken into account in quantifying the
241 duplicate.	240	monosaccharides. All hydrolysis and chromatographic measurements were carried out in
	241	duplicate.

242 2.4.2 Degree of methyl-esterification

The degree of methyl-esterification (DM) was analyzed based on the method of
Kyomugasho, Christiaens, Shpigelman, Van Loey, & Hendrickx (2015) using Fourier

245	transform infra-red (FT-IR) spectroscopy (ATR-FTIR, Shimadzu FTIR-8400S, Japan). The
246	transmittance of the samples was recorded at wavenumbers from 4000 cm^{-1} to 400 cm^{-1} at
247	resolution 4 cm ⁻¹ , peak intensities at 1600-1630 cm ⁻¹ due to carboxylate group (COO ⁻) and
248	at 1740 cm ⁻¹ due to methyl-ester carbonyl group (C=O) stretching being of interest. The ratio
249	(<i>R</i>) between the peak intensity at 1740 cm ⁻¹ and the sum of the peak intensities at 1740 cm ⁻¹
250	and 1600-1630 cm ⁻¹ was used to predict the DM of the samples based on the calibration line:
251	DM (%) = 136.86 x R + 3.987 (Kyomugasho, Christiaens, et al., 2015). The calibration line
252	was constructed using citrus pectin with different DMs (0-94 %) obtained
253	enzymatically/chemically as well as another set of DMs (0-94 %) generated by mixing
254	polygalacturonic acid and high DM citrus pectin in different ratios (Kyomugasho, Christiaens,
255	et al., 2015).
256	2.4.3 Protein content
257	The total nitrogen content of the samples was measured using an EA 1110 CHNS-O
258	elemental analyzer (CE-Instruments/Thermo Fisher Scientific). About 1 mg of the sample was
259	placed in a crimped tin capsules (8 mm \times 5 mm) prior to combustion in the elemental
260	analyzer. A conversion factor of 6.25 was used to calculate the amount of proteins in the
261	sample (Immerzeel, Eppink, de Vries, Schols, & Voragen, 2006).
262	2.4.4 Molecular weight distribution and intrinsic viscosity
263	The molecular weight distribution and intrinsic viscosity of the samples were analyzed
264	based on the work of Shpigelman, Kyomugasho, Christiaens, Van Loey, & Hendrickx (2014).
265	A high performance size exclusion chromatography (HPSEC) integrated to 4 detectors,
266	namely: PN3621 multi-angle laser light scattering (MALLS) from Postnova analytics,
267	Germany; Shodex RI refractive index (RI) from Showa Denko K.K., Kawazaki, Japan; a PN
268	3310 viscometer from Postnova analytics, Germany and a G1316A diode array detector
269	(DAD) from Agilent technologies, Diegem, Belgium that qualitatively detects the presence of

270	UV absorbing molecules, was used. Prior to HPSEC analysis, the samples were dissolved and
271	prepared differently as they showed different filterability and recovery after filtration.
272	However, all the samples were eluted using the same buffer and analytical conditions. The
273	AIRs and RGI-rich fractions (0.2 % w/v) were stirred overnight in a pH 4.4 buffer (0.1 M
274	acetic acid and NaNO ₃) followed by filtration through a 0.45 μ m syringe filter (Miller-HV).
275	For the de-esterified AIRs and HG-rich fractions, 0.2 % (w/v) sample was first dissolved in
276	ultrapure (organic free, 18 M Ω cm resistance) water overnight followed by dialysis (MWCO,
277	3.5 kDa) against 0.05 M NaNO ₃ solution for 48 h. To facilitate the filterability, these samples,
278	were heated at 50 °C for 10 min prior to filtration (Thibault et al., 1993). Exactly 100 μ L of
279	filtered sample was then injected using an auto-sampler onto a series of three Waters column
280	(Waters, Milford, MA), namely, Ultrahydrogel 250, 1000 and 2000 with exclusion limits of
281	8×10^4 , 4×10^6 , and 1×10^7 g/mol, respectively. A constant flow rate of 0.5 mL/min of the
282	elution buffer (0.1 M acetic acid and NaNO ₃ buffer at pH 4.4) was applied and the columns
283	were kept at 35 °C. A dn/dc value of 0.146 mL/g was used to calculate the concentration. The
284	molecular weight was calculated using the Debye fitting method (up to 2 nd order) by the
285	software provided by the manufacturer of MALLS detector (NovaMals, version 1.2.0.0,
286	Postnova analytics, Germany). The intrinsic viscosity was obtained from the viscometer
287	detector signals and the calculated concentration of the samples. All the samples were
288	analyzed in duplicate.

289 2.5 Visualization of the microstructure of carrot serum pectin polysaccharides (AIRs) and
290 derived pectin subdomains

Cryo-scanning electron microscopy (cryo-SEM) was used to visualize the
microstructure of carrot serum pectic polysaccharides in solution. Approximately 1 mg of
lyophilized sample was dissolved overnight in 1 mL ultrapure water (organic free, 18 MΩ cm
resistance). A few drops of the sample was placed onto a slot on a stub with rivets, vitrified

and transferred into the cryo-stage at -140 °C in the cryo-preparation chamber (PP3010T
cryo-SEM preparation system, Quorom Technologies, UK). The sample was freeze-fractured,
sublimated at -90 °C for 25-30 min under controlled vacuum conditions and then sputter
coated with platinum using argon gas to prevent charging during electron beam targeting
(Kyomugasho et al., 2016). Finally, the sample was transferred onto the SEM stage and
examined using a JEOL JSM 7100F SEM (JEOL Ltd, Tokyo, Japan) for their microstructure
in solution.

302 2.6 Data analysis

To determine the statistical significance, duplicate means were analyzed by one-way analysis of variance using JMP statistical software (JMP Pro 13.0.0, SAS Institute Inc.). A student's t-test was used to compare significantly different (*p*-value <0.05) mean values of only two groups (i.e. between starting material and derived fraction), while a post-hoc test using Tukey's honest significant difference (HSD) was employed for mean values of more than two groups (i.e. among starting materials or derived fractions).

309 3. Results and discussion

3.1 Molecular properties of RG I-rich fractions and their respective carrot serum pectic
polysaccharides (AIRs)

312 *3.1.1 Monosaccharide composition*

First, it can be seen that the AIR of HT+HPH contained a higher amount of total 313 314 sugars and pectin-associated sugars compared to the AIRs of HPH+HT and HPH+LT+HT (Table 1). This suggests solubilization of distinct pectic polysaccharides into the serum phase 315 depending on the combinations of thermal (*i.e.* high and low) and mechanical treatments 316 applied in preparing the purées (Santiago et al., 2016). High concentrations of galacturonic 317 acid (GalA) as well as considerable amounts of rhamnose (Rha), arabinose (Ara), galactose 318 (Gal) and glucose (Glu) were detected in the alcohol insoluble residues (AIRs), while trace 319 amounts of fucose (Fuc), xylose (Xyl), mannose (Man) and glucuronic acid (GlucA) were 320

found. Among the monosaccharides, Glu and Man have been identified as non-pectin 321 322 associated sugars which possibly originate from residual starch, glucomannan and noncrystalline cellulose (Houben, Jolie, Fraeye, Van Loey, & Hendrickx, 2011; Massiot, Rouau, 323 & Thibault, 1988). The presence of these non-pectic sugars in carrot serum was also reported 324 by Kyomugasho, Willemsen, et al. (2015). 325 Looking at the RGI fractions, based on the weight of the recovered residues over the 326 amount of the starting materials, the percentage isolation yield of the RGI-rich fractions was 327 about 33.1 % for HT+HPH, 27.5 % for HPH+HT and 36.6 % for HPH+LT+HT. It was 328 observed that significantly (*p*-value <0.05) lower concentrations of Glu and Man were 329 exhibited by all the RGI-rich fractions compared to their corresponding initial materials 330 (AIRs), as presented in Table 1. This suggests the degradation and solubilization of Glu and 331 Man containing non-pectic polysaccharides. Besides the decrease in Glu content, remarkably 332 333 lower concentrations of GalA were observed in all the RGI-rich fractions. The concentration of GalA that remained in the RGI-rich fractions was approximately 16-18 % (w/w) of the 334 amount of GalA in the initial samples implying that about 82-84 % (w/w) of GalA present in 335 the initial material was degraded and solubilized. β-elimination reaction of methyl-esterified 336 GalA residues of the homogalacturonan subdomain has been reported as the main mechanism 337 of pectin degradation at alkaline pH and high temperature conditions (Sila et al., 2009). 338 Furthermore, the extent of GalA content reduction among the RGI-rich fractions 339

relative their respective AIRs was comparable suggesting a relatively similar susceptibility of their homogalacturonan to degradation under the given conditions. Given that the mole ratio of the remaining GalA to Rha was too high to represent only RGI, a repeating disaccharide of $[\rightarrow 4)-\alpha$ -D-GalA- $(1\rightarrow 2)-\alpha$ -L-Rha- $(1\rightarrow)$ (O'Neill, et al., 1996), the remaining GalA could probably also originated from RGII as its backbone is made up of GalA as well as residual HG. In the current work, 59-64 % (*w/w*) of the initial Rha concentration was retained while

63-69 % (*w/w*) Ara and 68-75 % (*w/w*) Gal were also found in the RGI-rich fractions.
Nevertheless, the considerable amounts of Rha, Ara and Gal compared to other sugars,

348 particularly GalA, indicates the presence of rhamnogalacturonan-rich serum pectin,

349 irrespective of the serum type obtained from different carrot purée preparations. Arabinose-

and galactose-containing polysaccharides covalently attached as side chains to rhamnose are

351 recognized structural components of RGI pectic subdomain (Voragen, Beldman & Schols,

352 2001; Endress, Mattes & Norz, 2006).

353 3.1.2. Sugar ratios: extent of branching of AIRs and their derived RGI fractions

To obtain information on a molecular level, the monosaccharide composition data was 354 used in calculating the molar ratios between pectin-related sugars (Houben et al., 2011). An 355 insight into the structural variations of the RGI-rich fractions compared to their AIRs is 356 presented in Table 2a. The molar ratio of GalA to the sum of Fuc, Rha, Ara, Gal and Xyl 357 358 describes the linearity of serum pectin, with a high value suggesting the presence of more linear pectin/HG-rich pectin; while the molar ratio of the sum of Ara and Gal to Rha provides 359 a global view of the extent of branching of RGI in which a high value suggests that the 360 branching point (Rha) largely contains sugar side chains rich in Ara and Gal. To determine 361 the contribution of RG to the pectin population, the molar ratio between Rha and GalA was 362 also determined (Houben et al., 2011; Kaya et al., 2014). In this context, it can be observed 363 that AIRs of HT+HPH and HPH+HT samples contained more linear and less branched pectin 364 compared to the AIR from the HPH+LT+HT sample which exhibited a higher degree of RGI 365 366 branching. This suggests structural variations among the initial serum pectic polysaccharides due to different processing combinations used in preparing the carrot purées which is 367 consistent with our previous work (Santiago, et al., 2016). Moreover, comparing the AIRs 368 369 with their derived RGI-rich fractions, a large difference in the linearity of pectin is noticeable due to the previously observed degradation of homogalacturonan chains and consequent loss 370 of GalA. By contrast, the extent of branching of RGI remained comparable in both RGI-rich 371

fractions and initial materials. This observation is supported by the higher proportion of Rha
to GalA in the RGI-rich fractions indicating a large contribution of RGI polysaccharides
(Table 2a). Regardless of the process combinations used to generate the starting serum pectic
polysaccharides, it can be deduced that the RGI-rich fractions isolated had very low amounts
of GalA-rich chains, while maintaining the branched RGI subdomain. This clearly suggests
that linkages between neutral sugars (both in the backbone and side chains) were less prone to
degradation under hot alkaline condition.

379 *3.1.3 Degree of methyl-esterification*

The degree of methyl-esterification (DM) of pectin, defined as the number of moles of 380 methyl esters per 100 moles of galacturonic acid residues, is an important characteristic 381 attributed to the linear homogalacturonan subdomain (Voragen et al., 2001). The serum pectic 382 polysaccharides from different processes (AIRs) contained high methyl-esterified GalA 383 residues ranging from 52 % to 65 %, as shown in Table 2a. The AIR of the HPH+LT+HT 384 sample had a significantly (p < 0.05) lower DM (~52 %) due to the action of endogenous 385 pectin methyl-esterase (Santiago et al., 2016), that was deliberately stimulated during the 386 carrot purée preparation (Figure 1). On the other hand, RGI-rich fractions derived from these 387 AIRs exhibited comparable very low DMs of 4.4 % to 5.6 %. These values are in close 388 agreement with the low DM (6%) of rhamnogalacturonan-rich pectin isolated from carrot 389 tissues through an enzymic liquefaction process (Schols & Voragen, 1994). In the current 390 work, the observed low DM (due to residual methyl esters) in the RGI-rich fractions is mainly 391 392 attributed to the loss of methyl-esterified GalA units owing to the base-catalyzed splitting of the homogalacturonan chains via a β -elimination reaction (Sila et al., 2009). The residual 393 methyl esters may be attributed to other substructures, since the GalA residues of the RGI 394 395 backbone are presumably not methyl-esterified (Kravtchenko, Arnould, Voragen, & Pilnik, 1992). Given that complete isolation of the HG was not achieved (as the rate of β -elimination 396 is retarded, as the reaction proceeds due to decreased DM), the RGI-rich fractions were 397

- probably associated with residual HG and/or RGII subdomains. The methyl groups can
 therefore be attributed to RGII which can be methyl-esterified at the C-6 position (Ishii et al.,
 1999) or presence of residual homogalacturonan chains with methyl-esterified GalA units.
- 401 *3.1.4 Molecular weight distribution and intrinsic viscosity*

The size exclusion elution profiles showing the RI, LS 92° and UV 280 nm signals of 402 the RGI-rich fractions and their corresponding initial material (AIRs) are presented in Figure 403 2. From the concentration profiles, it can be observed that all the pectic polymers of the initial 404 samples (AIRs) were eluted earlier, starting at ~38 min while polymers of RGI-rich fractions 405 eluted later, starting at ~44 min (Figure 2A). This elution time difference suggests a change in 406 the hydrodynamic volume of the polymers after subjecting the serum pectic polysaccharides 407 (AIRs) to hot alkaline conditions, due to degradation of the HG subdomains. The polymers of 408 RGI-rich fractions presented smaller hydrodynamic volumes than their respective AIRs. 409 410 Furthermore, the AIR of the HT+HPH sample showed a significantly (p < 0.05) higher M_w compared to AIRs of HPH+HT and HPH+LT+HT samples, indicating the solubilization of 411 412 different pectic populations into the serum phase during the purée preparation. By contrast, 413 the concentration peak profiles of the RGI-rich fractions appeared identical and their M_w were also relatively comparable (Table 2b). The M_w of RGI-rich fractions were significantly (p 414 <0.05) lower than their respective initial materials (AIRs) for both the high and low molecular 415 weight polymer populations. The polydispersity indices of the different AIRs ranged from 416 1.21 to 1.90, while slightly lower polydispersity indices (1.18-1.36) were observed in the RG-417 rich fractions. As previously discussed that carrot serum pectin had high degree of branching, 418 a comparison of its M_w to a commercial pectin reported to be highly branched (sugar beet 419 pectin) is made in this section. The average M_w (108 ± 11 kDa) of the RGI-rich fractions of 420 carrot serum pectic polysaccharides was lower compared to the M_w (188 kDa) of sugar beet 421 pectin RGI fraction (Morris et al., 2010). Besides the difference due to botanical source, 422

isolation procedures used in obtaining the RGI fractions can be attributed to these M_w 423 differences. In the aforementioned study, the homogalacturonan region of sugar beet pectin 424 was enzymatically hydrolyzed to obtain the RGI fraction in contrast to the use of chemical 425 means employed in the present study. 426 In terms of intrinsic viscosity, as AIR of HT+HPH sample had very high M_w polymers 427 (in high concentration), a high intrinsic viscosity was observed. However, the AIR of 428 HPH+HT sample that had significantly lower M_w polymers than AIR of the HT+HPH sample 429 showed comparable intrinsic viscosity that can be due to their relatively comparable 430 molecular structures (cf. Table 2a, 2b). The alcohol insoluble residue with the lowest M_w and 431 being rich in neutral sugar side chains (HPH+LT+HT) showed the lowest intrinsic viscosity 432 (Table 2). Neutral sugar-rich pectins was found to have lower intrinsic viscosity which can be 433 due to the presence of highly dense and spherical polymers that might either be branched 434 435 pectin molecules or relatively stable aggregates (Kravtchenko, Voragen & Pilnik, 1992). Moreover, from the UV 280 nm profile, a peak from ~58 min elution time is noticeable and 436 could be due to UV absorbing compounds such as proteins and polyphenols (Figure 2B). In 437 fact, according to Christiaens et al. (2015), the absorbance at 280 nm was attributed to 438 proteins, which were presumed to be attached to the high M_w polymers of water-soluble pectin 439 of carrot-derived waste streams. To confirm the presence (absence) of proteins, in the current 440 study, the protein content was determined based on the total nitrogen content of the samples 441 and ranged from 7.2 % to 13.4 % for the AIRs and, 1.2 % to 3.2 % for the RGI-rich fractions 442 (Table 2a). This possibly indicates that proteins could be associated with the neutral sugar 443 side chains of carrot pectin (Immerzeel et al., 2006), but not all detected proteins could be 444 attached to pectin especially in the AIRs. Proteins detected in commercial pectins such as 445 apple and lemon pectins were found to be mainly present in the neutral sugar side chains 446 (Kravtchenko, Penci, Voragen & Pilnik, 1993). 447

448 3.2 Molecular properties of HG-rich fractions and their respective de-esterified AIR

449 *3.2.1 Monosaccharide composition*

Based on the weight of the material recovered from the starting AIRs, the isolation 450 vields of the HG-rich fractions were about 43.4 %, 42 % and 44.9 % (w/w) for HT+HPH, 451 HPH+HT and HPH+LT+HT serum samples, respectively. The monosaccharide composition 452 of the HG-rich fractions differed from their respective de-esterified AIRs (Table 3). Low to 453 trace amounts of Rha, Ara and Gal were detected in the HG-rich fractions. In comparison to 454 the respective initial sugar content of the de-esterified AIR, the HG-rich fractions retained 455 Rha ranging from 26-30 % (w/w), Ara from 0-2 % (w/w) and Gal from 11-16 % (w/w). This 456 suggests the degradation of the linkages of neutral sugar side chains as well as most of the 457 rhamnogalacturonan I backbone of the carrot serum pectic polysaccharides. This also shows 458 that Ara-containing side chains of carrot serum pectic polysaccharides were the most labile to 459 460 acid followed by Gal, which is in agreement with the results on the mild acid hydrolysis of commercial pectins from apple, sugar beet, citrus (Kaya et al., 2014; Morris et al., 2010; 461 Thibault et al., 1993; De Vries et al., 1983) and pectin from the pericarp of unripe tomato 462 (Round et al., 2010). Moreover, an average of 87 % (w/w) of the initial GalA concentration 463 was recovered in the HG-rich fractions, which evidenced the isolation of the 464 homogalacturonan-rich carrot serum pectin. The HG-rich fractions obtained had about 89.4 to 465 92.0 mol % of GalA (Table 3), indicating a higher concentration of GalA to the total pectin-466 related sugars compared to the de-esterified AIR (57.8-67.1 mol % of GalA). This shows that 467 GalA linkages of carrot serum pectic polysaccharides were less susceptible to degradation 468 under mild acid condition compared to the RGI subdomain, which is in agreement with the 469 previous findings of several researchers (Kaya et al., 2014; Morris et al., 2010; Round et al., 470 2010; Thibault et al., 1993). 471

472 3.2.2. Degree of methyl-esterification and sugar ratios of de-esterified AIRs and HG-rich
473 fractions

The de-esterified AIRs and HG-rich fractions exhibited comparable low DM values ranging from 5.7 to 8.6 % (Table 4a), indicating that the mild acid conditions used to generate the HG-rich fractions did not induce further chemical de-methoxylation. These DM values are in close agreement with the DM of de-esterified and acid hydrolyzed sugar beet (5 %), apple (3 %) and citrus pectin (3 %) (Thibault et al.,1993).

To assess the linearity and the contribution of RG to pectin population as well as the 479 extent of RGI branching, the molar ratios of pectin-related sugars were evaluated as discussed 480 previously (cf. section 3.1.2). First, it must be noted that AIR and de-esterified AIR showed 481 similar sugar ratio trends for all the serum types, which suggests that subjecting the AIR to 482 alkaline conditions for 6 h at 4 °C did not alter other structural properties of serum pectic 483 484 polysaccharides, except the DM (cf. Table 2a and Table 4a). Hence, the de-esterified AIRs which served as starting material for isolation of HG-rich fractions were comparable to the 485 starting material (AIR) for isolation of RGI-rich fractions. 486

Comparing the de-esterified AIRs and HG-rich fractions, the former presented less 487 linear and more branched serum pectic polysaccharides (Table 4a). By contrast, HG-rich 488 fractions exhibited a lower concentration of the major pectin-related neutral sugars (Rha, Ara 489 and Gal) that consequently resulted in a reduction of their degree of RGI branching. 490 Moreover, the HG-rich fractions were highly linear as represented by high values of the sugar 491 ratios representing the linearity of pectin. This evidenced the degradation of neutral sugar side 492 chains as well as the RGI backbone of de-esterified AIRs. The results indicated the substantial 493 loss of the RGI subdomain of carrot serum pectic polysaccharides under the given acid 494 condition, however, a fragment of the RGI backbone was possibly still present in the GalA-495 rich fraction. Specifically, the HG-rich fraction of the HPH+LT+HT sample (initially having 496

the highest degree of RGI branching) showed a considerable molar ratio of Ara and Gal to
Rha compared to the other HG-rich fractions (Table 4a). This could possibly be due to the
observed higher initial degree of RGI branching of this specific sample (HPH+LT+HT) which
probably required longer hydrolysis time (Thibault et al., 1993).

501 *3.2.3 Molecular weight distribution and intrinsic viscosity*

Differences in the molecular weight and intrinsic viscosity of the HG-rich fractions 502 and their respective de-esterified AIRs were observed. Comparing all the different initial 503 materials (de-esterified AIRs), different peak profiles and molecular weights were detected 504 (Figure 3). These varying peak profiles can be related to the previously observed different 505 structural properties (e.g. linearity and degree of RGI branching) and possible differences on 506 the conformation of serum pectic polysaccharides in solution. Comparing the de-esterified 507 AIRs with their respective HG-rich fractions, differences in the peak maximum of their 508 509 concentration curves can be distinguished (Figure 3A). The peak maxima of the concentration curves of HG-rich fractions were situated at around ~55 min, while the peak maxima of de-510 esterified AIRs were observed between ~50 to 52 min. This indicates that a high population of 511 polymers in the HG-rich fractions had smaller hydrodynamic volume compared to the 512 polymers of de-esterified AIRs. Moreover, the HG-rich fractions were generally characterized 513 by lower M_w pectic polymers (compared to AIRs) with two polymer populations, one eluting 514 at approximately 42-48 min (population I) and the other at 48-62 min (population II). For 515 population I, the very high M_w values obtained can be possibly due to aggregated polymers. 516 These high apparent M_w polymers comprised less than 10 % of the injected total mass fraction 517 which can be due to the aggregation of some pectic polymers, given the high linearity of 518 pectin (Table 4b), very low DM, hence high sensitivity to divalent ions (Shpigelman et al., 519 2014). For polymer population II, low M_w pectic polymers comprising more than 50 % of the 520 total mass fraction (Table 4b). Given its high concentration, this population of low M_w pectic 521

polymers can be considered as the representative M_w values of the HG-rich fractions, which is 522 523 in close agreement with the values of 17 kDa- 37 kDa for the homogalacturonan fraction of citrus pectins (Kaya et al., 2014; Yapo et al., 2007) and 17 kDa- 22 kDa for sugar beet pectins 524 (Morris et al., 2010; Thibault et al., 1993) obtained by mild acid hydrolysis. In the present 525 study, the polydispersity indices of the isolated HG-rich fractions, being 1.03 to 1.14 for 526 polymer population II (the most abundant population), suggested a relatively more 527 homogenous M_w distribution of polymers compared to the de-esterified AIR with 528 polydispersity indices of 1.28-1.98. 529 Furthermore, although the de-esterified AIRs exhibited higher M_w , they presented 530 relatively lower intrinsic viscosities particularly for the high M_w polymer population 531

532 compared to the HG-rich fractions. Specifically, the intrinsic viscosities of HG-rich fractions

of HT+HPH and HPH+HT samples eluted at 44 to 48 min were 1.39 dL/g and 1.79 dL/g,

respectively, which are higher than the intrinsic viscosities of their corresponding de-

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esterified AIRs, being 0.84 dL/g and 0.94 dL/g. This can probably be explained by their
different hydrodynamic volumes and structural properties, in which the de-esterified AIRs
were characterized by more branched pectic polymers and probably suggesting higher
flexibility in solution leading to low intrinsic viscosity. Polymer population II of the HG-rich

539 fractions, predominantly containing high concentration of low M_w polymers, had an intrinsic

viscosity of 0.34-0.46 dL/g, which are relatively lower than the intrinsic viscosity of 0.70-1.16

541 dL/g for acid hydrolyzed citrus and sugar beet pectins (Kaya et al., 2014; Morris et al., 2010;

542 Yapo, Lerouge, Thibault, & Ralet, 2007; Thibault et al., 1993). The lower intrinsic viscosity

of the homogalacturonan-rich fractions observed in the current work can probably be

attributed to the presence of RGI fragments. Polymer flexibility, which is reported to

545 influence the intrinsic viscosity (Kravtchenko et al., 1992), increases with HG-rich fractions

546 containing fragments of RGI than HG fractions that are almost entirely free from branched

rhamnogalacturonan I. Aggregation and high polydispersity of pectic polymers can also influence the M_w and intrinsic viscosity values (Endress, Matess & Norz 2006). Moreover, the UV 280 nm profiles generally showed a peak from ~58 min which can be due to proteins and other UV-absorbing molecules such as polyphenols (Figure 3B). In fact, the de-esterified AIRs had a protein content ranging from 4.9-8.2 %, while the HG-rich fractions had 1.5-3.3 % protein (Table 4a). The protein detected in HG-rich fractions was probably associated with the residual RGI subdomain after acid hydrolysis and dialysis.

554 3.3 Microstructure of carrot serum AIRs and the partitioned subdomains

AIRs of the lyophilized serum obtained from differently-prepared carrot purées, and 555 their partitioned subdomains were viewed under a cryo-SEM to further investigate their 556 microstructure in solution. Figure 4 shows representative cryo-SEM micrographs of 0.1 % 557 solutions of alcohol insoluble residues (AIRs), RGI- and HG-rich fractions. In the 558 559 micrographs, white branch-like and/or strand-like structures can be observed and are mostly composed of pectic polymers together with a little vitreous ice. Specifically, the different 560 AIRs are composed of branch-like structures. Although the AIRs of the HT+HPH and 561 HPH+LT+HT samples were previously shown to compose of serum pectic polysaccharides 562 with different M_w , degree of RGI branching and DM, there was no clear/distinguishable 563 differences in their microstructures under cryo-SEM. The RGI-rich fraction of HT+HPH 564 sample exhibited a dense population of short branch-like structures compared to its AIR as 565 well as to the RGI-rich fraction of HPH+LT+HT (Figure 4). In terms of the HG-rich fractions, 566 the HT+HPH sample showed more visible long strand-like features with limited branch-like 567 structures (if any). Furthermore, comparing the different AIRs with their respective RGI and 568 HG-rich fractions, more distinguishable structural features can be identified. The 569 microstructures of the AIRs are comparable to their respective RGI-rich fractions owing to the 570 branch-like features. By contrast, the HG-rich fractions exhibit strand-like structures with no 571

branch-like features. These cryo-SEM micrographs show that AIRs and RGI-rich fractions of
can exhibit comparable microstructures in solutions, which are different from microstructures
exhibited by HG-rich fractions, an observation that can be greatly attributed to their overall
structure differences. Carrot serum pectic polysaccharides is composed of highly branched
pectic populations; which is in agreement with our physico-chemical characterization results
as well as the findings of Houben et al. (2011).

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579 Conclusion

Different processing combinations, applied in preparing the carrot purées, can be used 580 to create distinct molecular structures of serum (solubilized) pectic polysaccharides. High 581 temperature treatment of carrot pieces followed by blending and high pressure 582 homogenization resulted in high concentration of serum pectin with high M_w , DM and 583 584 linearity. Furthermore, the structure of serum pectic polysaccharides, isolated from the serum phase of differently-processed carrot purées, was successfully partitioned into its constituting 585 rhamnogalacturonan I and homogalacturonan subdomains through selective chemical 586 hydrolysis. On the one hand, a rhamnogalacturonan I-rich fraction, mainly characterized by a 587 high concentration of rhamnose, arabinose and galactose, was obtained by subjecting the 588 serum pectic polysaccharides to hot alkaline conditions to selectively degrade the 589 homogalacturonan subdomain. On the other hand, a homogalacturonan-rich fraction of, highly 590 linear galacturonic acid-rich polysaccharides, was isolated through mild acid hydrolysis to 591 degrade the linkages between neutral sugar side chains. Physico-chemical characterization 592 was complemented by cryo-SEM micrographs indicating that RGI subdomains (which are 593 highly branched) exhibited branch-like features, while the HG subdomains showed strand-like 594 595 structures. This study provides an insight into the structure of carrot serum pectin and revealing the structural characteristics of serum pectin is an important step towards directing 596

597	structural changes to attain specific functionalities such as gelling, thickening or emulsifying.
598	For more information, it might be interesting to explore the specific sugar linkages present.

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Figure 1. Carrot purée preparation indicating the sequence of mechanical and thermal
treatments with the respective sample codes (in italics). HT: high temperature treatment;
HPH: high pressure homogenization; LT: low temperature treatment.

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Figure 2. Size exclusion elution profiles of the RGI-rich fractions and alcohol insoluble
 residues (AIR) of the lyophilized serum obtained from differently-prepared carrot purées (A)

residues (AIX) of the typpinized serum obtained from differently-prepared carrot purees (A)
 log molar mass distribution (thick lines) superimposed on concentration chromatogram (thin

curves) (B) Light scattering signal at 92° angle (thick curves) superimposed on UV

absorbance chromatogram at 280 nm (thin curves). Solid lines for HT+HPH, long dash for

780 HPH+HT and square dot lines for HPH+LT+HT sample. All grey colored lines represent de-

- resterified AIRs, while black colors for HG-rich fractions.
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Figure 3. Size exclusion elution profiles of the HG-rich fractions and the respective de- de-

resterified AIR of the (lyophilized) serum obtained from differently-prepared carrot purées (A)

786 log molar mass distribution (thick lines) superimposed on concentration chromatogram (thin

787 curves) (B) Light scattering signal at 92° angle (thick curves) superimposed on UV

absorbance chromatogram at 280 nm (thin curves). Solid lines for HT+HPH, long dash for

789 HPH+HT and square dot lines for HPH+LT+HT sample. All grey colored lines represent de-

resterified AIRs, while black colors for HG-rich fractions.

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Figure 4. Representative cryo-SEM images of carrot serum pectic polysaccharides with the respective RGI-rich and HG-rich fractions. Scale bars $=1\mu m$ 795

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Tables:

Table 1. Average monosaccharide composition (\pm SD) of the alcohol insoluble residues (AIRs) of lyophilized serum from differently-prepared carrot purées and isolated rhamnogalacturonan I-rich fractions (RGI). Different capital letters (in superscript) indicate statistical differences (*p*-value <0.05) among AIRs or RGI-rich fractions, while small letters indicate statistical difference between AIR and RGI-rich fraction. The values in parentheses indicate the mol % of each sugar. (HT: high temperature treatment at 95 °C for 30 min; HPH: high pressure homogenization at 100 MPa; LT: low temperature treatment at 60 °C for 40 min).

						Mon	osaccharides (mg/	g AIR)				
Sample codes		Fuc	Rha	Ara	Gal	Gluc	Xyl	Man	GalA	GlucA	Total sugars	Total amount of pectin- associated sugars
	AIR	1.8 ± 0.2^{-Aa}	$39.3 \pm 0.9^{\text{Aa}}$	$60.4 \pm 3.7^{\text{Aa}}$	$77.5 \pm 10.2^{\text{Aa}}$	$30.5 \pm 6.7 ^{\rm Aa}$	2.2 ± 0.8 Aa	8.5 ± 1.7 ^{Aa}	455.2 ± 71.9 ^{Aa}	$4.43 \pm 3.63^{\text{Aa}}$	679.05 ± 100.57	636.5 ± 87.8
HT+HPH	RGI	(0.3) 1.2 ± 0.1 Aa (0.6)	$(6.0) \\ 26.6 \pm 1.6^{\text{Aa}} \\ (12.3)$	(10.5) 43.4 ± 1.4 ^{Ab} (22.0)	(11.6) $62.4 \pm 1.4^{\text{Ab}}$ (26.3)	$(5.1) \\ 4.4 \pm 0.3 \\ (1.9) $	$(0.5) \\ 1.4 \pm 0.2^{ABa} \\ (0.7)$	(1.3)	$(63.9) \\ 97.4 \pm 10.6 \\ (38.1)$	$\begin{array}{c} (0.8) \\ 3.28 \pm 0.69 \\ (1.3) \end{array}^{\rm Aa}$	242.69 ± 16.59	232.6 ± 15.4
	AIR	1.5 ± 0.1^{Aa}	$26.3 \pm 0.7^{\text{Ba}}$	$51.4 \pm 0.4^{\text{Aa}}$	$75.2 \pm 2.8^{\text{Aa}}$	$69.4 \pm 0.0^{\text{Ba}}$	$1.9 \pm 0.2^{\text{Aa}}$	9.9 ± 0.6^{Aa}	$416.4 \pm 12.5^{\text{Aa}}$	$8.36 \pm 0.82^{\text{Aa}}$	660.25 ± 18.17	572.6 ± 16.8
HPH+HT	RGI	(0.3) 0.9 ± 0.1 Aa (0.5)	(4.5) 16.7 ± 1.3 ^{Bb} (9.2)	(9.6) 35.7 ± 2.3 ^{Bb} (21.5)	(11.7) 56.1 ± 3.5 ^{Ab} (28.2)	(10.8) 5.7 ± 1.4^{Ab} (2.9)	(0.4) 1.1 ± 0.2 Aa (0.7)	(1.5) 2.4 ± 0.0 ^{Ab} (1.2)	$(60.1) \\ 75.4 \pm 4.1 \\ (35.2)$	(1.2) 1.47 ± 1.76^{Ab} (0.7)	195.53 ± 14.66	185.9 ± 11.5
	AIR	$1.7\pm0.1 \ ^{Aa}$	$19.6\pm1.1^{\ Ca}$	$66.2\pm5.6 \ ^{Aa}$	$117.6\pm10.2^{\ Ba}$	$66.9\pm8.6^{\ Ba}$	4.0 ± 1.6 ^{Aa}	$16.8\pm1.8 ^{Ba}$	$358.4 \pm 38.5 \ ^{Aa}$	$7.88\pm0.35\ ^{Aa}$	659.08 ± 67.74	567.6 ± 57.1
HPH+LT+HT	RGI	(0.3) 1.0 ± 0.0 ^{Aa} (0.5)	(3.1) 12.0 ± 1.2 ^{Bb} (5.8)	(11.5) 43.9 ± 0.6 Ab (23.3)	(16.9) 84.3 ± 0.0 ^{Bb} (37.3)	(10.3) 7.1 \pm 2.0 ^{Ab} (3.1)	(0.7) 1.84 ± 0.04 ^{Ba} (1.0)	(2.4) 4.9 ± 0.6 ^{Bb} (2.2)	(53.6) 65.3 ± 0.3 ^{Bb} (26.8)	$(1.2) \\ 0.25 \pm 0.23 \\ (0.1) \\ (1.2) \\ Ab \\ (0.1) \\ Ab \\ Ab \\ (0.1) \\ Ab \\ Ab \\ (0.1) \\ Ab \\ (0.1) \\ $	220.64 ± 5.00	208.3 ± 2.2

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Table 2a. Average pectin-related sugar ratios, DM and protein content (± SD) of the rhamnogalacturonan I-rich fractions (RGI) and their respective alcohol insoluble residues (AIR) obtained from the lyophilized serum of differently-prepared carrot purées. Different capital letters (in superscript) indicate statistical differences (*p*-value <0.05) among AIRs or RGI-rich fractions, while small letters indicate statistical difference between AIR and RGI-rich fraction. (HT: high temperature treatment at 95 °C for 30 min; HPH: high pressure homogenization at 100 MPa; LT: low temperature treatment at 60 °C for 40 min).

		RGI (Rha/GalA) Branching	of RGI (Ara+Gal/Rha)	Protein (% d.b.)	DM (%)
R 2.2 ±	$\pm 0.1^{ABa}$ 0.09 \pm	0.01 ^{Aa}	3.7 ± 0.1^{-Aa}	$9.6 \pm 0.04^{\text{Aa}}$	$65.3 \pm 0.5^{\text{Aa}}$
		0.02 ^{Ab}	$3.9\pm0.1~^{\text{Aa}}$	$1.8\pm0.08^{\;Ab}$	$4.4\pm0.7^{\;Ab}$
					$65.1\pm1.1^{\ Aa}$
GI 0.6 -	$\pm 0.1^{Ab}$ 0.26 \pm	0.01 ^{ABb}	$5.4\pm0.4^{\rm \ Aa}$	$1.2\pm0.03^{\;Ab}$	$4.6 \pm 1.1^{\text{Ab}}$
			$9.1\pm0.3^{\ Ca}$		$52.1\pm1.7^{\text{ Ba}}$
GI 0.4 :	$\pm 0.1^{Bb}$ 0.22 \pm	0.02 ^{Bb}	$10.5 \pm 1.1^{\text{Ba}}$	3.2 ± 0.09^{Bb}	$5.6\pm0.3^{\;Ab}$
	(GalA/Fuc+Rh IR 2.2 = GI 0.6 = IR 2.3 = GI 0.6 = IR 1.7	(GalA/Fuc+Rha+Ara+Gal+Xyl) IR 2.2 ± 0.1^{ABa} $0.09 \pm$ GI 0.6 ± 0.1^{Ab} $0.32 \pm$ IR 2.3 ± 0.1^{Aa} $0.07 \pm$ GI 0.6 ± 0.1^{Ab} $0.26 \pm$ IR 1.7 ± 0.2^{Ba} $0.06 \pm$	(GalA/Fuc+Rha+Ara+Gal+Xyl) Controlution of ROT (Rha/GalA) Drateling (R 2.2 ± 0.1^{ABa} 0.09 ± 0.01^{Aa} GI 0.6 ± 0.1^{Ab} 0.32 ± 0.02^{Ab} (R 2.3 ± 0.1^{Aa} 0.07 ± 0.01^{Ba} GI 0.6 ± 0.1^{Ab} 0.26 ± 0.01^{Abb} (R 1.7 ± 0.2^{Ba} 0.06 ± 0.01^{Ca}	(GalA/Fuc+Rha+Ara+Gal+Xyl) Controlation of ROT (Rna GalA) Diancing of ROT (Ana GalA) IR 2.2 ± 0.1^{ABa} 0.09 ± 0.01^{Aa} 3.7 ± 0.1^{Aa} GI 0.6 ± 0.1^{Ab} 0.32 ± 0.02^{Ab} 3.9 ± 0.1^{Aa} IR 2.3 ± 0.1^{Aa} 0.07 ± 0.01^{Ba} 4.8 ± 0.2^{Ba} GI 0.6 ± 0.1^{Ab} 0.26 ± 0.01^{ABb} 5.4 ± 0.4^{Aa} IR 1.7 ± 0.2^{Ba} 0.06 ± 0.01^{Ca} 9.1 ± 0.3^{Ca}	(GalA/Fuc+Rha+Ara+Gal+Xyl) Controlation of ROT (Rna GalA) Branching of ROT (Ana GalA) Interning of ROT (Ana GalA)

Table 2b. Average molecular weight, intrinsic viscosity and mass fraction (± SD) for each corresponding polymer population of alcohol insoluble residue (AIR) and its rhamnogalacturonan I-rich fraction. Different capital letters (in superscript) indicate statistical differences (*p*-value <0.05) among AIRs or RGI-rich fractions, while small letters indicate statistical difference between AIR and RGI-rich fraction. (HT: high temperature treatment at 95 °C for 30 min; HPH: high pressure homogenization at 100 MPa; LT: low temperature treatment at 60 °C for 40 min).

Sample	codes	M,	v(kDa)	Intrinsic vis	scosity (dl/g)	Fraction of mass found		
		Population I	Population II	Population I	Population II	Population I	Population II	
HT+HPH	AIR	$1370.0 \pm 42.4^{\text{Aa}}$	$130.0 \pm 8.5^{\text{Aa}}$	2.30 ± 0.03 ^{Aa}	$1.14 \pm 0.03^{\text{Aa}}$	$0.19 \pm 0.02^{\text{Aa}}$	$0.41 \pm 0.02^{\text{Aa}}$	
ΠΙ+ΠΡΠ	RGI	$98.5\pm00.3^{\rm Ab}$	$10.4\pm0.4~^{\rm Ab}$	$0.37\pm0.01~^{Ab}$	$0.08\pm0.01~^{\rm Ab}$	$0.29\pm0.01~^{Ab}$	$0.34\pm0.02^{\;\mathrm{Aa}}$	
HPH+HT	AIR RGI	$577.5 \pm 33.2^{\ Ba} \\ 119.5 \pm 00.7^{\ Bb}$	57.1 ± 1.8^{Ba} 14.8 ± 0.1^{Bb}	$\begin{array}{c} 2.22 \pm 0.05 \\ 0.37 \pm 0.01 \\ ^{Bb} \end{array}$	$\begin{array}{c} 0.94 \pm 0.02 ^{\ Ba} \\ 0.09 \pm 0.01 ^{\ Ab} \end{array}$	$\begin{array}{l} 0.09 \pm 0.01^{\ Ba} \\ 0.24 \pm 0.01^{\ Bb} \end{array}$	$\begin{array}{c} 0.53 \pm 0.01 \; ^{Ba} \\ 0.41 \pm 0.02 \; ^{Bb} \end{array}$	
HPH+LT+HT	AIR RGI	$\begin{array}{c} 132.5 \pm 06.4 \\ ^{Ca} \\ 119.0 \pm 01.4 \\ ^{Bb} \end{array}$	$\begin{array}{c} 15.8 \pm 2.6 \\ 15.4 \pm 0.9 \\ ^{\text{Ba}} \end{array}$	$\begin{array}{c} 1.42 \pm 0.03 ^{\ Ba} \\ 0.25 \pm 0.01 ^{\ Cb} \end{array}$	$\begin{array}{c} 0.09 \pm 0.01 {}^{\rm Ca} \\ 0.08 \pm 0.01 {}^{\rm Aa} \end{array}$	$\begin{array}{c} 0.46 \pm 0.01 ^{Ca} \\ 0.19 \pm 0.01 ^{Cb} \end{array}$	$\begin{array}{c} 0.13 \pm 0.01 ^{\ Ca} \\ 0.45 \pm 0.01 ^{\ Bb} \end{array}$	

Table 3. Average monosaccharide composition (\pm SD) of the homogalacturonan-rich fractions (HG) and de-esterified alcohol insoluble residues (de-AIR) of lyophilized serum obtained from differently-prepared carrot purées. Different capital letters (in superscript) indicate statistical differences (*p*-value <0.05) among de-esterified AIRs or HG-rich fractions, while small letters indicate statistical difference between de-esterified AIR and HG-rich fraction. The values in parentheses indicate the mol % of each sugar. (HT: high temperature treatment at 95 °C for 30 min; HPH: high pressure homogenization at 100 MPa; LT: low temperature treatment at 60 °C for 40 min).

						Monosacch	arides (mg/g de-	-esterified AIR)				
Sample codes		Fuc	Rha	Ara	Gal	Gluc	Xyl	Man	GalA	GlucA	Total sugars	Total amount of pectin- associated sugars
	De-esterified AIR	$1.9 \pm 0.1^{\text{Aa}}$ (0.3)	$36.8 \pm 0.6^{\text{Aa}}$ (6.5)	$57.1 \pm 2.1 \stackrel{ABa}{(11.1)}$	$75.3 \pm 2.0^{\text{Aa}}$ (12.2)	$11.7 \pm 2.9^{\text{Aa}}$ (1.9)	3.7 ± 0.0 (0.7) Aa	$6.0 \pm 1.8 \stackrel{\text{Aa}}{(1.0)}$	$436.9 \pm 34.9^{\text{Aa}}$ (65.6)	$4.7 \pm 0.9^{\text{Aa}}$ (0.7)	634.2 ± 45.33	611.7 ± 39.6
HT+HPH	HG	(0.3) 0.1 ± 0.0 Ab (0.0)	(0.5) 11.2 \pm 0.3 ^{Ab} (3.3)	(11.1) 1.3 ± 1.2 Ab (0.4)	(12.2) 8.0 ± 0.3 ^{Ab} (2.2)	(1.9) 6.4 ± 2.3 Aa (1.7)	(0.7) 3.5 ± 2.8^{Aa} (1.1)	(1.0) 4.0 ± 2.0 ^{Aa} (1.1)	$\begin{array}{c} (65.6)\\ 354.1 \pm 29.5 \\ (89.7)\end{array}$	(0.7) 1.44 ± 0.3 ^{Ab} (0.4)	389.9 ± 38.6	378.1 ± 34.1
HPH+HT	De-esterified AIR	$1.5 \pm 0.3 \stackrel{\text{Aa}}{(0.3)}$	$27.6 \pm 1.8 Ba$ (5.1)	$51.8 \pm 4.7 \stackrel{\text{Ba}}{=} (10.6)$	$73.2 \pm 10.0^{\text{Aa}}$ (12.4)	$13.4 \pm 5.7^{\text{Aa}}$ (2.3)	(0.4)	$7.9 \pm 2.5 \stackrel{\text{Aa}}{(1.3)}$	425.4 ± 42.8 ^{Aa} (67.1)	$2.9 \pm 1.0^{\text{Aa}}$ (0.4)	605.9 ± 71.1	581.7 ± 60.0
nrn+n1	HG	$\begin{array}{c} 0.0 \pm 0.0 \\ (0.0) \end{array}^{\rm Ab}$	$7.5 \pm 0.6 \stackrel{Bb}{(2.3)}$	$\begin{array}{c} 0.7 \pm 0.5 \\ (0.2) \end{array}^{\rm Ab}$	$10.0 \pm 0.6 \stackrel{Ab}{(2.8)}$	$4.8 \pm 0.2^{\text{Aa}}$ (1.4)	$ \begin{array}{c} 1.1 \pm 0.0 \\ (0.4) \end{array} $	$2.2 \pm 1.4 \\ (0.6) \\ $	352.6 ± 13.0 ^{Aa} (92.0)	$1.4 \pm 0.2 \stackrel{Aa}{=} (0.4)$	380.3 ± 16.4	371.9 ± 14.7
	De-esterified AIR	$1.6 \pm 0.0^{\text{Aa}}$ (0.3)	20.8 ± 0.2 ^{Ca} (3.8)	$67.3 \pm 0.2^{\text{Aa}}$ (13.3)	$121.6 \pm 3.6^{\text{Ba}}$ (20.0)	10.7 ± 0.3 ^{Aa} (1.8)	$2.3 \pm 0.5 \stackrel{\text{Aa}}{=} (0.4)$	$9.4 \pm 1.7 \stackrel{Aa}{=} (1.5)$	377.8 ± 31.2 ^{Aa} (57.8)	$6.7 \pm 1.4^{\text{Aa}}$ (1.0)	618.0 ± 39.2	591.3 ± 35.7
HPH+LT+HT	HG	$0.0 \pm 0.0^{\text{Ab}}$ (0.0)	5.3 ± 0.4 ^{Cb} (1.7)	$0.2 \pm 0.0^{\text{Ab}}$ (0.1)	$19.6 \pm 3.4^{\text{Bb}}$ (5.6)	$5.0 \pm 0.3^{\text{Ab}}$ (1.4)	$1.8 \pm 0.2^{\text{Aa}}$ (0.6)		335.7 ± 39.6 ^{Aa} (89.4)	1.8 ± 0.6^{Ab} (0.5)	371.8 ± 44.5	362.8 ± 43.6

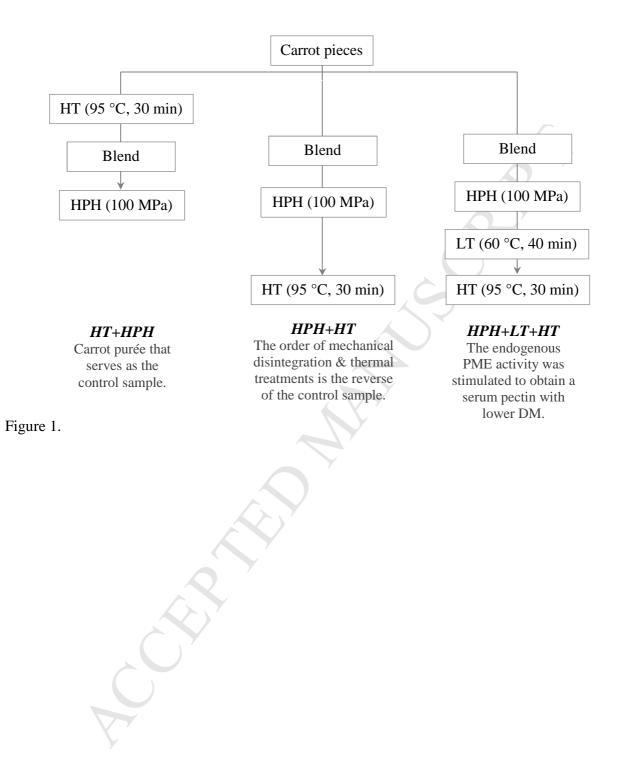
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Table 4a. Average pectin-related sugar ratios, DM and protein content (\pm SD) of homogalacturonan-rich fractions compared to the de-esterified alcohol insoluble residues (de-esterified AIR) obtained from the lyophilized serum of differently-prepared carrot purées. Different capital letters (in superscript) indicate statistical differences (*p*-value <0.05) among de-esterified AIRs or HG-rich fractions, while small letters indicate statistical difference between de-esterified AIR and HG-rich fraction. (HT: high temperature treatment at 95 °C for 30 min; HPH: high pressure homogenization at 100 MPa; LT: low temperature treatment at 60 °C for 40 min).

Sar	nple codes	Linearity of pectin (GalA/Fuc+Rha+Ara+Gal+Xyl)	Branching of RGI (Ara+Gal/Rha)	Contribution of RGI (Rha/GalA)	Protein (% d.b.)	DM (%)
HT+HPH	De-esterified AIR	2.1 ± 0.1 ^{Aa}	3.6 ± 0.1 Aa	0.10 ± 0.01 ^{Aa}	$7.4\pm0.1^{-\mathrm{Aa}}$	$8.6 \pm 1.0^{\operatorname{Aa}}$
ΠΙ+ΠΡΠ	HG	12.6 ± 0.1 ^{Ab}	0.8 ± 0.1 ^{Ab}	$0.04 \pm 0.01^{\text{Ab}}$	$3.1\pm0.1~^{\rm Ab}$	$8.0\pm0.3~^{\rm Aa}$
HPH+HT	De-esterified AIR HG	$\begin{array}{c} 2.3 \pm 0.1 ~^{\rm Aa} \\ 16.1 \pm 0.1 ~^{\rm Bb} \end{array}$	$\begin{array}{c} 4.5 \pm 0.2^{Ba} \\ 1.3 \pm 0.2^{Ab} \end{array}$	$\begin{array}{c} 0.08 \pm 0.01 \\ 0.03 \pm 0.01 \\ ^{ABb} \end{array}$	$\begin{array}{c} 4.9 \pm 0.2^{\; Ba} \\ 1.5 \pm 0.1^{\; Bb} \end{array}$	$\begin{array}{c} 7.0 \pm 1.0 \\ 6.0 \pm 0.3 \\ \end{array}^{Ba}$
HPH+LT+HT	De-esterified AIR HG	$\begin{array}{c} 1.5 \pm 0.1^{\; \mathrm{Ba}} \\ 11.1 \pm 1.2^{\; \mathrm{Cb}} \end{array}$	$\begin{array}{c} 8.9 \pm 0.2^{\ Ca} \\ 3.4 \pm 0.8^{\ Bb} \end{array}$	$\begin{array}{c} 0.07 \pm 0.01 \\ ^{Ba} \\ 0.02 \pm 0.01 \\ ^{Bb} \end{array}$	$\begin{array}{c} 8.2 \pm 0.1 \\ 3.3 \pm 0.1 \\ \end{array}^{Aa}$	$\begin{array}{c} 5.9 \pm 0.7^{\; Ba} \\ 5.7 \pm 0.7^{\; Ba} \end{array}$

Table 4b. Average molecular weight, intrinsic viscosity and mass fraction (\pm SD) for each corresponding polymer population of de-esterified alcohol insoluble residue (de-esterified AIR) and derived homogalacturonan-rich fractions (HG). Different capital letters (in superscript) indicate statistical differences (*p*-value <0.05) among de-esterified AIRs or HG-rich fractions, while small letters indicate statistical difference between de-esterified AIR and HG-rich fraction. (HT: high temperature treatment at 95 °C for 30 min; HPH: high pressure homogenization at 100 MPa; LT: low temperature treatment at 60 °C for 40 min).

		$M_w(kDa)$		Intrinsic viscosity (dl/g)		Fraction of mass found	
Sample codes		Population I	Population II	Population I	Population II	Population I	Population II
HT+HPH	De-esterified AIR	$534.0 \pm 125^{\text{Aa}}$	$61.8 \pm 12.5^{\text{Aa}}$	$0.84 \pm 0.10^{\text{Aa}}$	0.57 ± 0.11^{-Aa}	$0.09 \pm 0.02^{\text{Aa}}$	$0.51 \pm 0.10^{\mathrm{Aa}}$
	HG	222.0 ± 9.2^{Ab}	$17.2 \pm 0.1^{\text{Ab}}$	$1.39\pm0.03^{\rm \ Ab}$	$0.44\pm0.01~^{Ab}$	$0.05\pm0.01~^{\rm Aa}$	$0.55\pm0.01^{~Aa}$
HPH+HT	De-esterified AIR HG		$63.4 \pm 1.7 {}^{\text{Aa}} \\ 23.4 \pm 8.1 {}^{\text{Ab}}$	$\begin{array}{c} 0.94 \pm 0.07 \\ 1.79 \pm 0.01 \\ \end{array}^{\rm Ab}$	$\begin{array}{c} 0.61 \pm 0.02 \; ^{\rm Aa} \\ 0.46 \pm 0.02 \; ^{\rm Ab} \end{array}$	$\begin{array}{c} 0.05 \pm 0.01 \; ^{Aa} \\ 0.03 \pm 0.01 \; ^{Aa} \end{array}$	$\begin{array}{c} 0.60 \pm 0.04 ^{Aa} \\ 0.57 \pm 0.04 ^{Aa} \end{array}$
HPH+LT+HT	De-esterified AIR HG		$\begin{array}{l} 32.1 \pm 2.9^{\ Ba} \\ 13.2 \pm 1.2^{\ Bb} \end{array}$	$\begin{array}{c} 0.67 \pm 0.10^{\;Ba} \\ 0.78 \pm 0.03^{\;Ba} \end{array}$	$\begin{array}{c} 0.46 \pm 0.01 \\ 0.34 \pm 0.01 \\ ^{Bb} \end{array}$	$\begin{array}{c} 0.24 \pm 0.02^{\;Ba} \\ 0.25 \pm 0.01^{\;Ba} \end{array}$	$\begin{array}{c} 0.46 \pm 0.11 \\ 0.31 \pm 0.06 \\ ^{Ba} \end{array}$



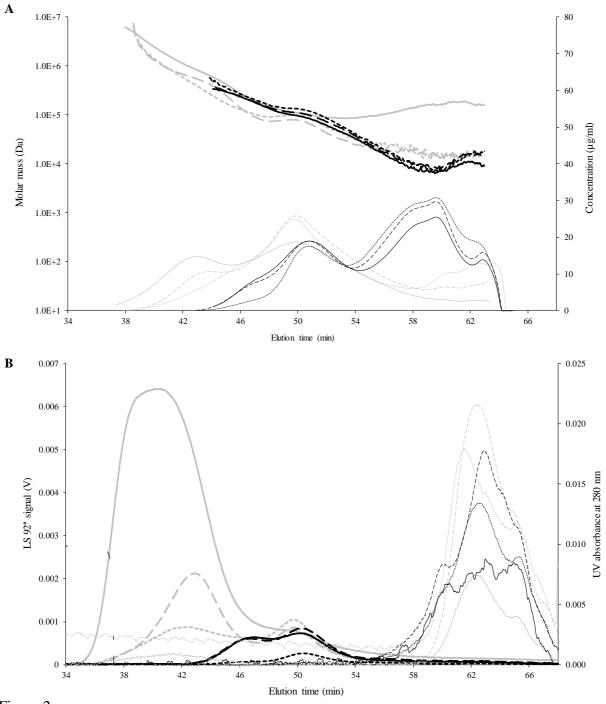


Figure 2.

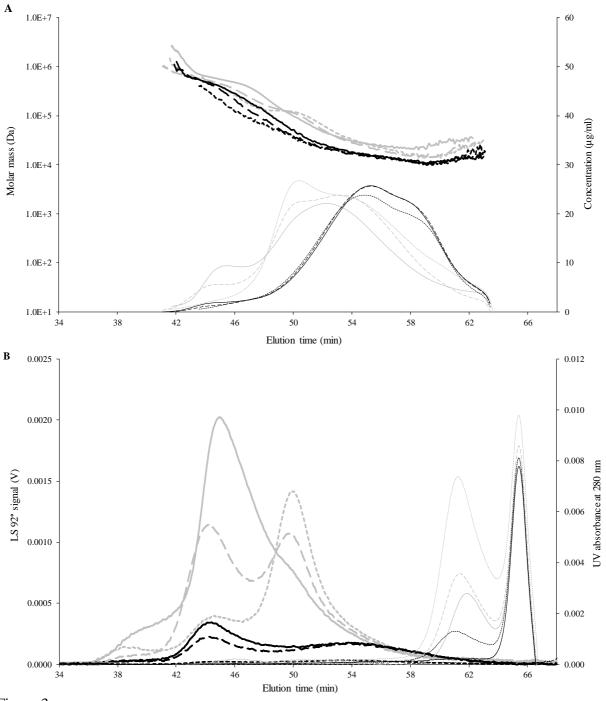
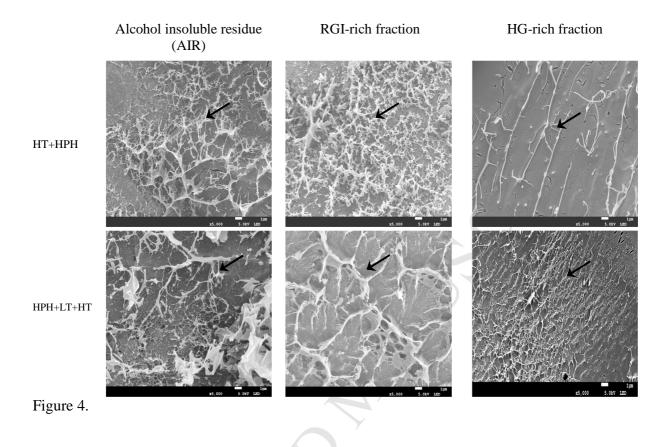


Figure 3.



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